

Univerza v Ljubljani, Fakulteta za farmacijo, 2009

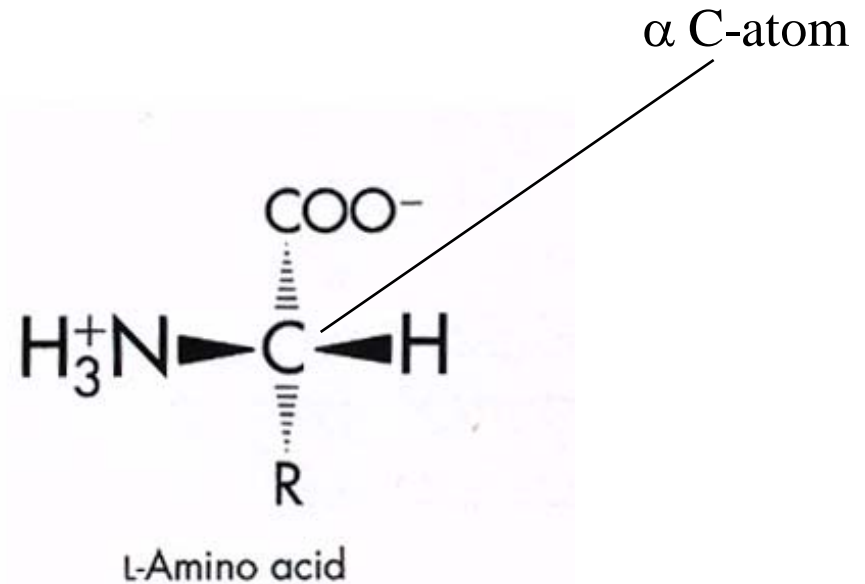
MOLEKULARNA ENCIMOLOGIJA

Matjaž Zorko
Medicinska fakulteta

GLEJ: http://ibk.mf.uni-lj.si/teaching/lab_medicina/default.html

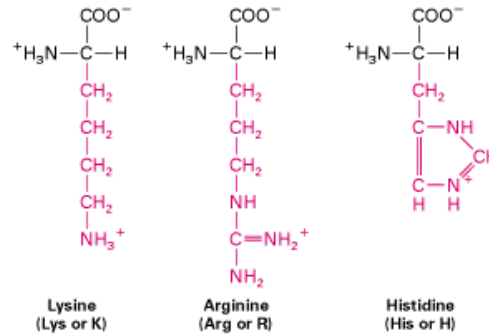
I. Glavne lastnosti (20 standardnih) aminokislin

- klasifikacija AK
- velikost
- naboj
- polarnost
- hidrofobnost
- aromatske AK
- optična aktivnost

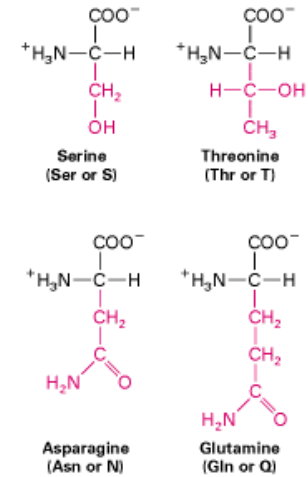


PAZI: več 100 nestandardnih AK (post-transl. modif. & D-AK)!

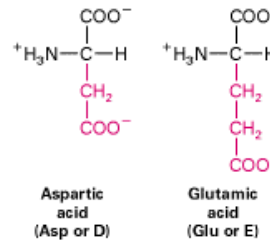
Basic amino acids



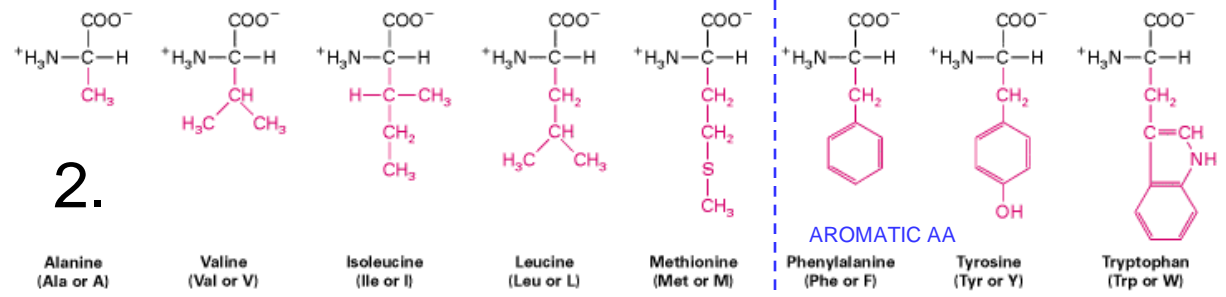
Polar amino acids with uncharged R groups



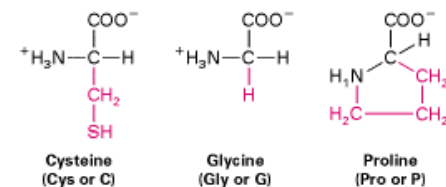
Acidic amino acids



HYDROPHOBIC AMINO ACIDS



SPECIAL AMINO ACIDS



Klasifikacija (R!):

1. hidrofилne (polarne):

- bazične
- kisle
- nenabit R

1.

2. hidrofobne (nepolarne)

3. 'posebne'

2.

3.

Velikost AK-ostanka:

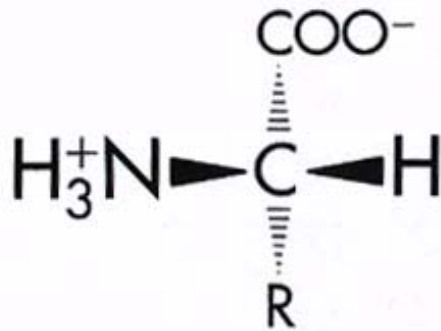
100-130 Da ($\overline{110}$ Da),
a nekaj izjem!

AK ostanek (residue)

AK radikal

dodaj 18 (H₂O) za maso AK!

odštej 56 za maso stranske verige!

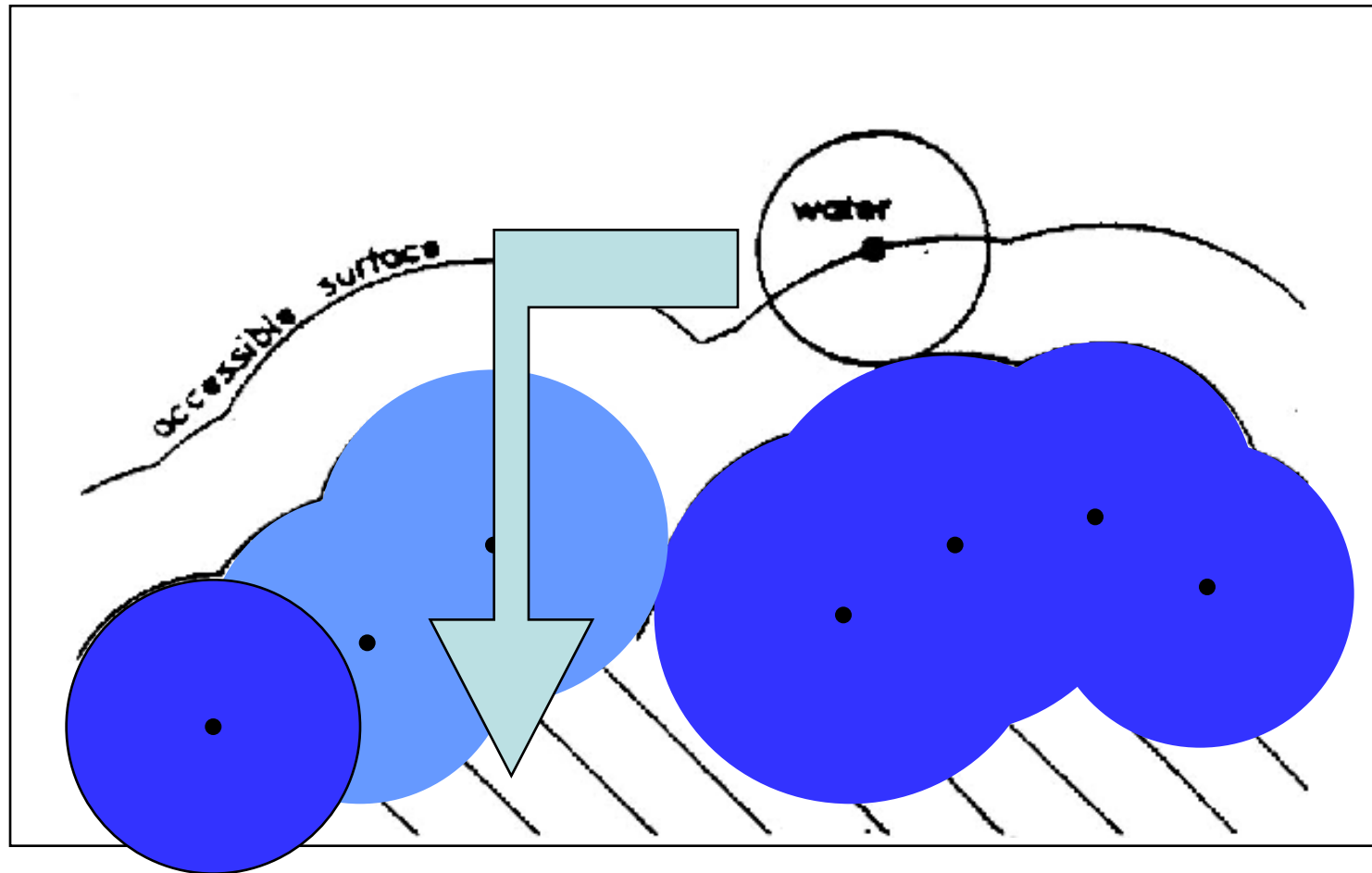


L-Amino acid

Table 1.1 Properties of Individual Amino Acid Residues

Residue	Residue mass ^a (daltons)	Van der Waals volume ^b (Å ³)	Frequency in proteins ^c (%)
Ala (A)	71.09	67	8.3
Arg (R)	156.19	148	5.7
Asn (N)	114.11	96	4.4
Asp (D)	115.09	91	5.3
Cys (C)	103.15	86	1.7
Gln (Q)	128.14	114	4.0
Glu (E)	129.12	109	6.2
Gly (G)	57.05	48	7.2
His (H)	137.14	118	2.2
Ile (I)	113.16	124	5.2
Leu (L)	113.16	124	9.0
Lys (K)	128.17	135	5.7
Met (M)	131.19	124	2.4
Phe (F)	147.18	135	3.9
Pro (P)	97.12	90	5.1
Ser (S)	87.08	73	6.9
Thr (T)	101.11	93	5.8
Trp (W)	186.21	163	1.3
Tyr (Y)	163.18	141	3.2
Val (V)	99.14	105	6.6

Velikost je pomembna pri zamenjavah aminokislin!

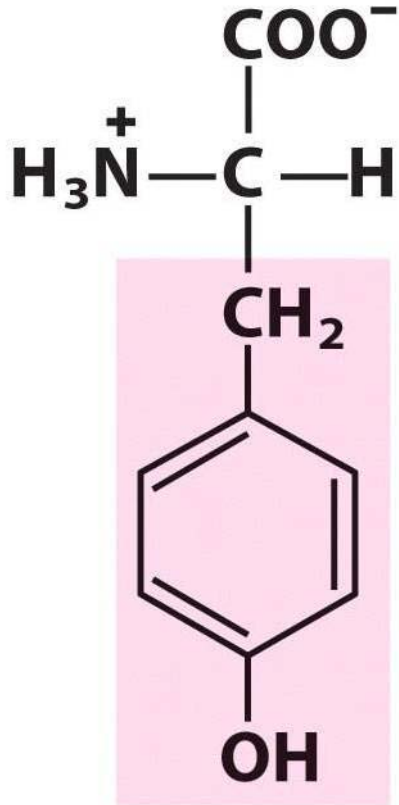


Samo R, N- and C- konci so so pomembni pri proteinih

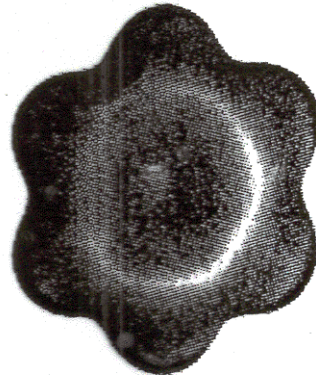
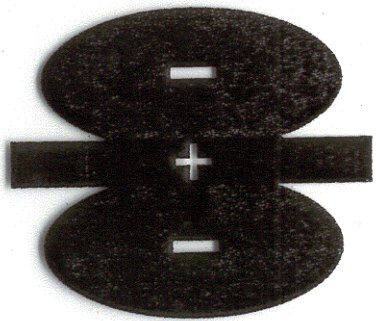
NABOJ:
Odvisen od pK_a vrednosti (25°C)

Amino acid	α -COOH group	α -NH ₃ ⁺ group	Side chain (R)
Alanine	2.3	9.9	
Glycine	2.4	9.8	
Phenylalanine	1.8	9.1	
Serine	2.1	9.2	
Valine	2.3	9.6	
Aspartic acid	2.0	10.0	3.9
Glutamic acid	2.2	9.7	4.3
Histidine	1.8	9.2	6.0
Cysteine	1.8	10.8	8.3
Tyrosine	2.2	9.1	10.9 !
Lysine	2.2	9.2	10.8
Arginine	1.8	9.0	12.5

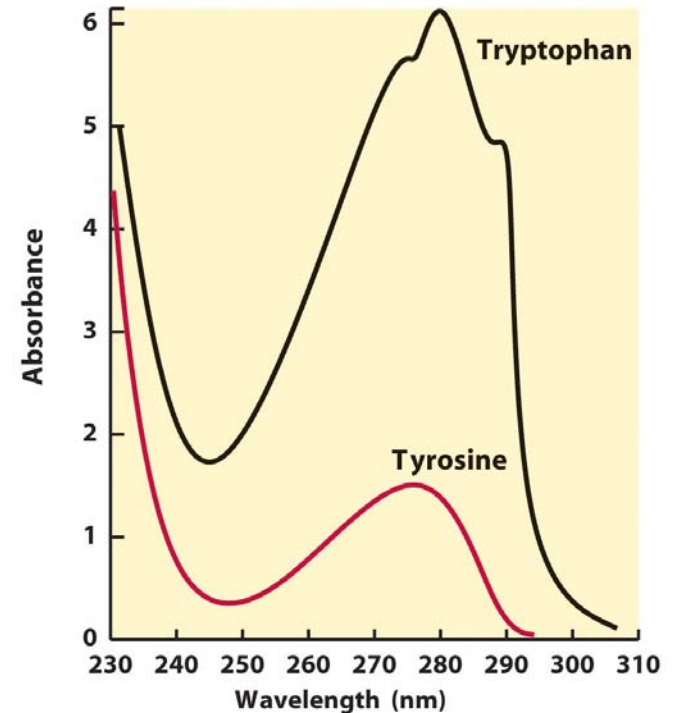
Aromatske AK (Phe, Tyr, Trp)

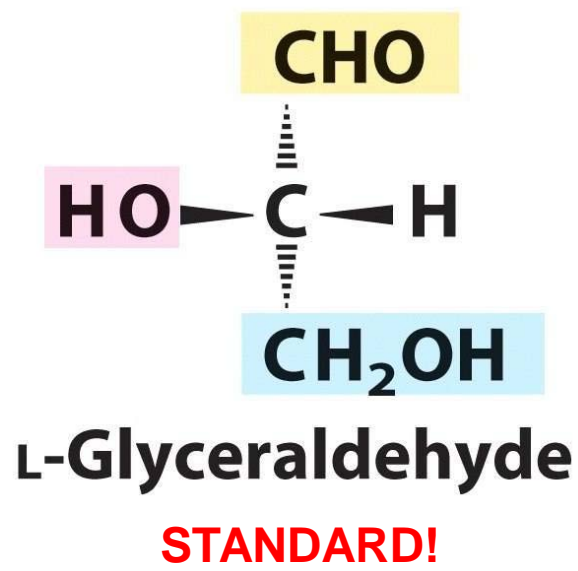
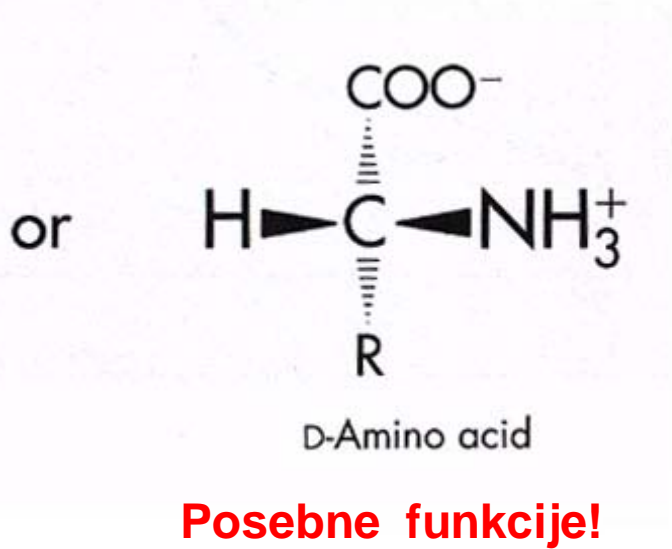
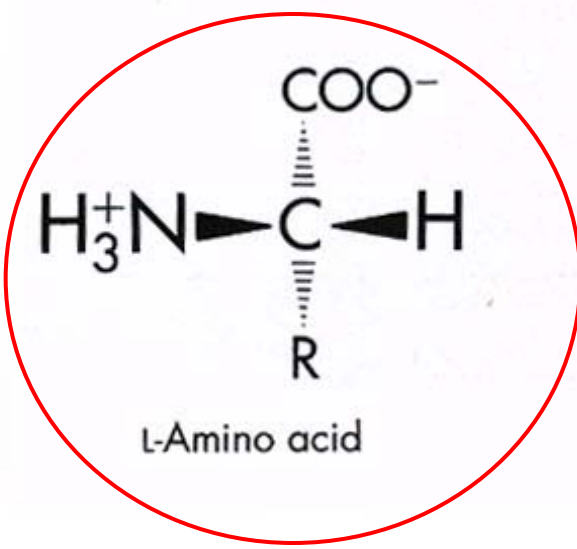
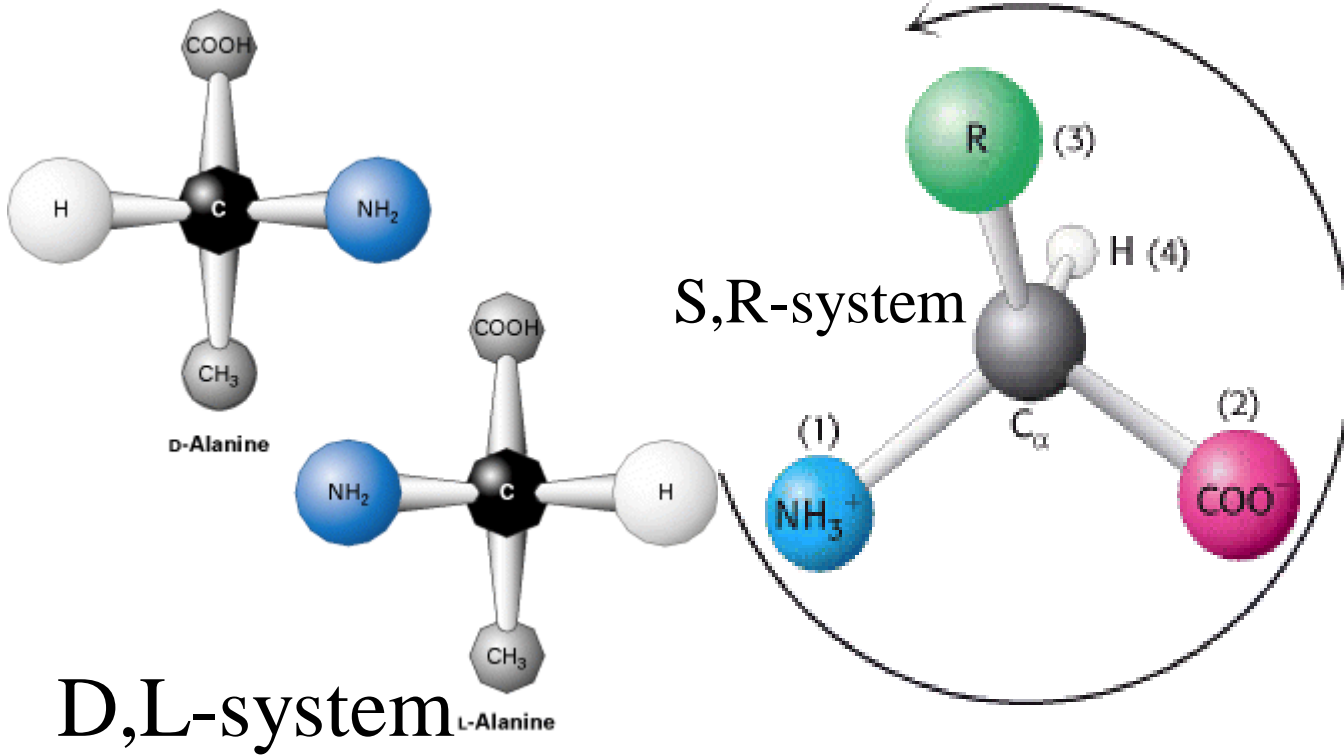


- velike
- resonanca: -OH v Tyr bolj 'kisla'
- polarnost: posebne interakcije
- absorbirajo UV svetlobo



B





Ne-esencialne Esencialne

Alanine

Histidine

Arginine

Isoleucine

Asparagine

Leucine

Aspartate

Lysine

Cysteine

Methionine

Glutamate

Phenylalanine

Glutamine

Threonine

Glycine

Tryptophan

Proline

Valine

Serine

Tyrosine

Bliskovit pregled proteinske sinteze

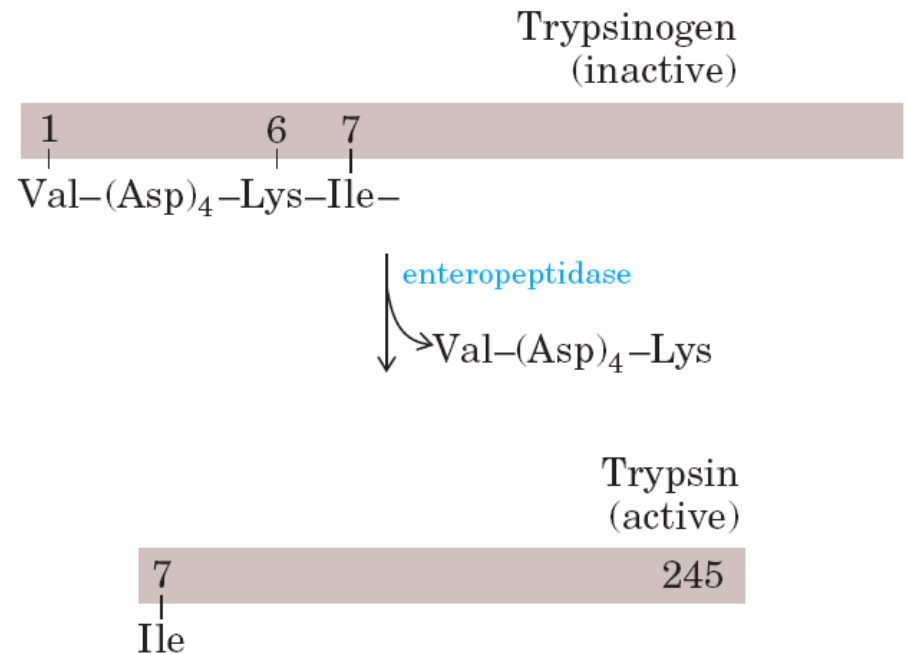
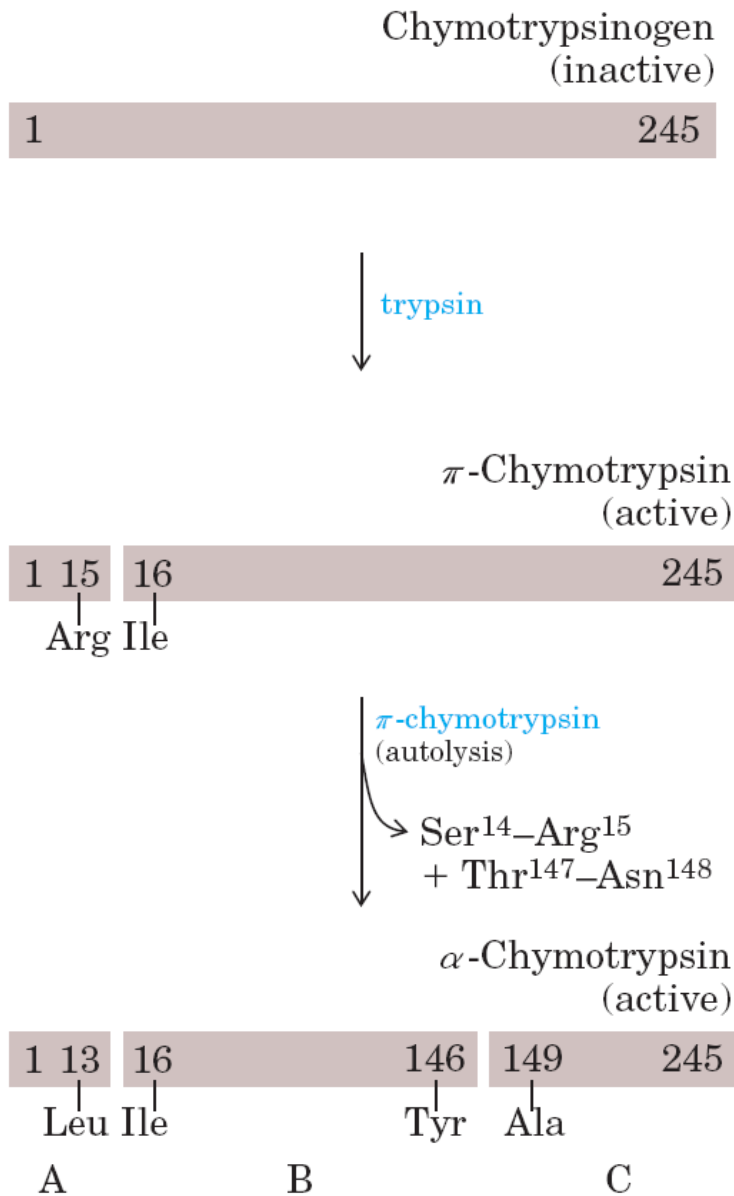
- DNA → primarni transkript (pre-mRNA)
- pre-mRNA se procesira
- mRNA je zapis za I. str. proteinov (I. struktura)
- nekatere AK se modificirajo
- 1D → 3D strukturo
- kontrola kvalitete delovanja

Post-translacijske modifikacije vpeljejo nove funkcionalne skupine in imajo tudi regulatorno vlogo. Znanih je več kot 80.

Pomembne modifikacije:

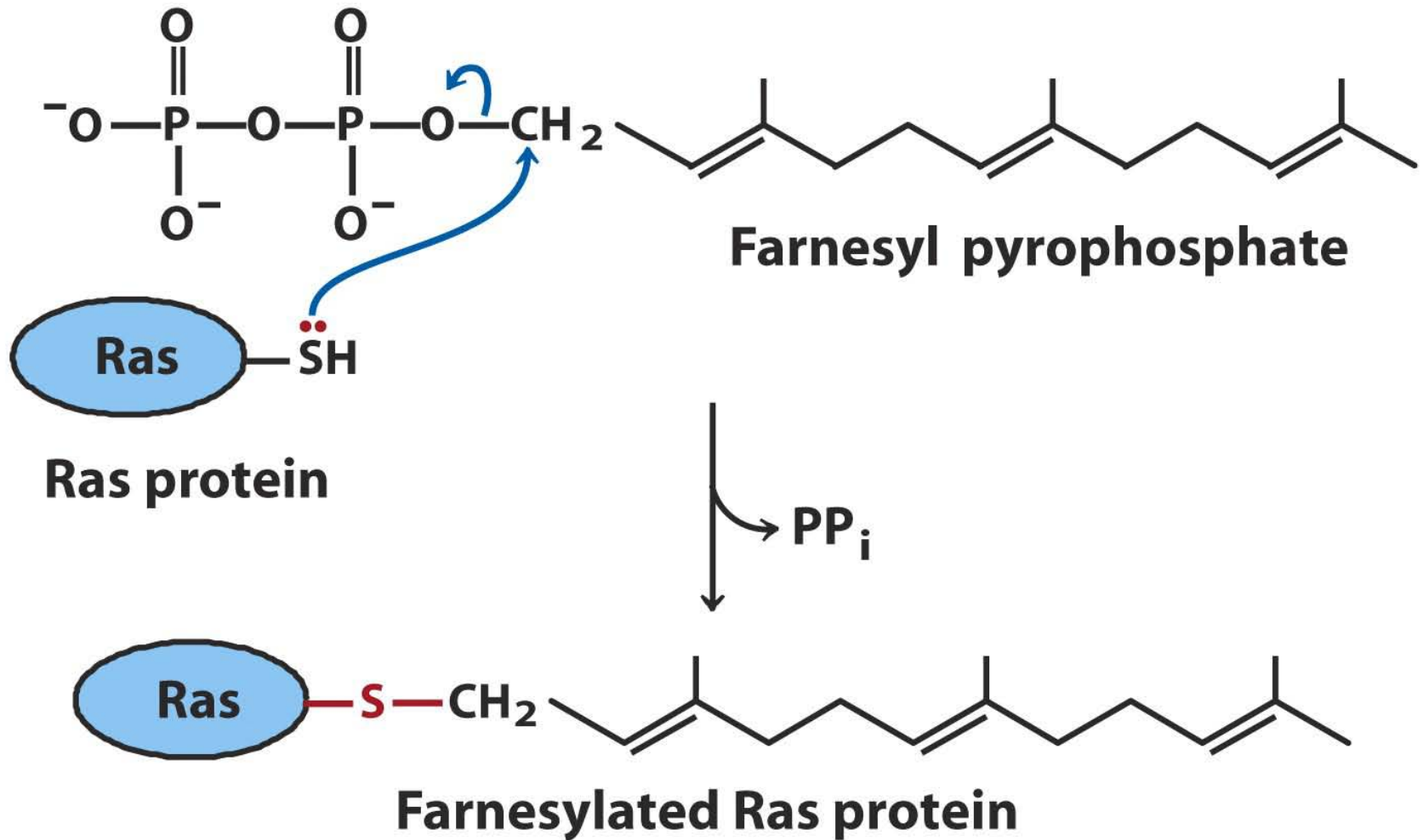
- proteolitično procesiranje
- spremembe N- in C-koncev
- glikozilacija
- pripenjanje lipidov
- sulfatiranje
- γ -karboksi-glutaminska kislina
- hidroksilacija
- fosforilacija
- ADP-ribozilacija
- disulfidni mostički

Proteolitično procesiranje: 'aktivacija' encimov



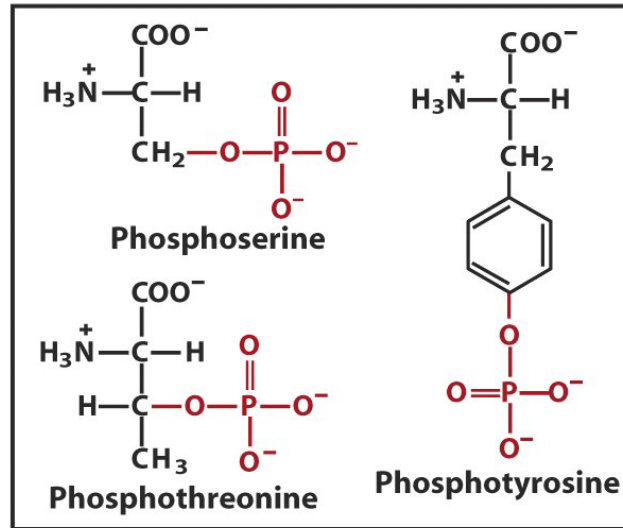
Namen: encim postane aktiven tam kjer deluje in ne na mestu sinteze!

Pripenjanje lipidov: interakcija z membrano, 3D struktura

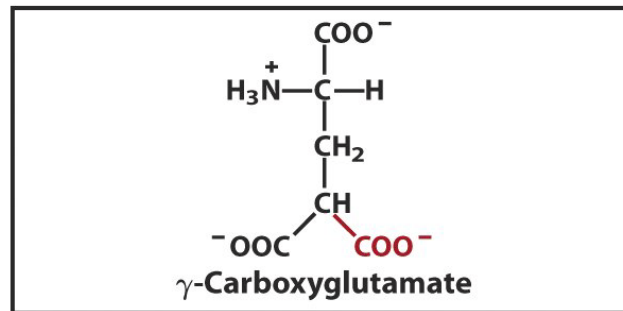


Pripenjanje malih funkcionalnih skupin

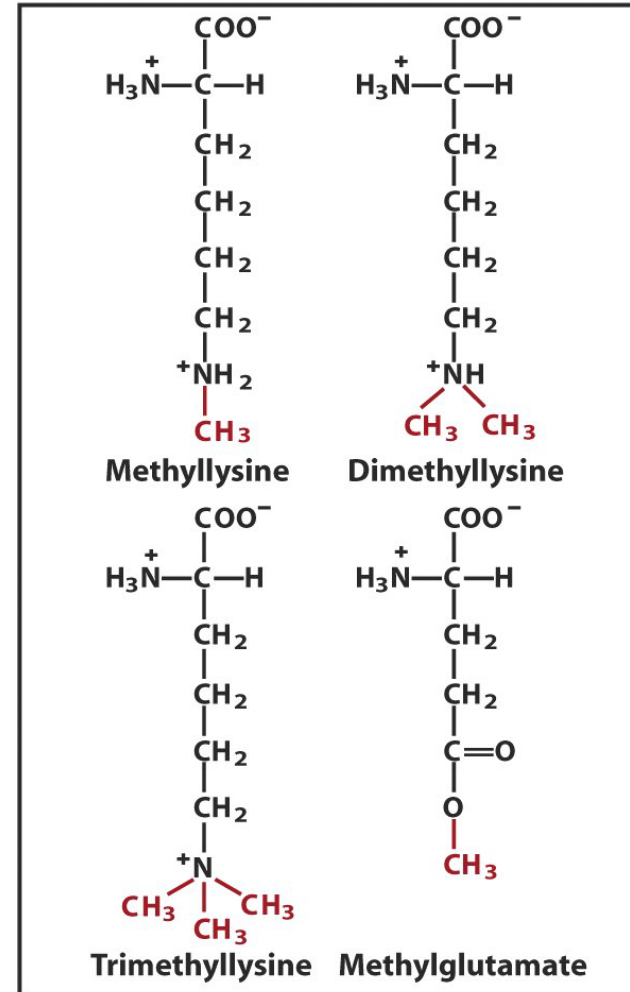
- fosforilacija
- γ -karboksi-glutamat
- metilacija
- sulfatiranje
- hidroksilacija



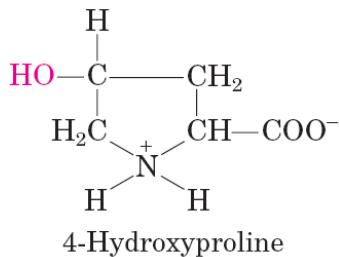
(a)



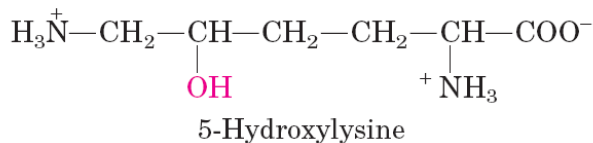
(b)



(c)

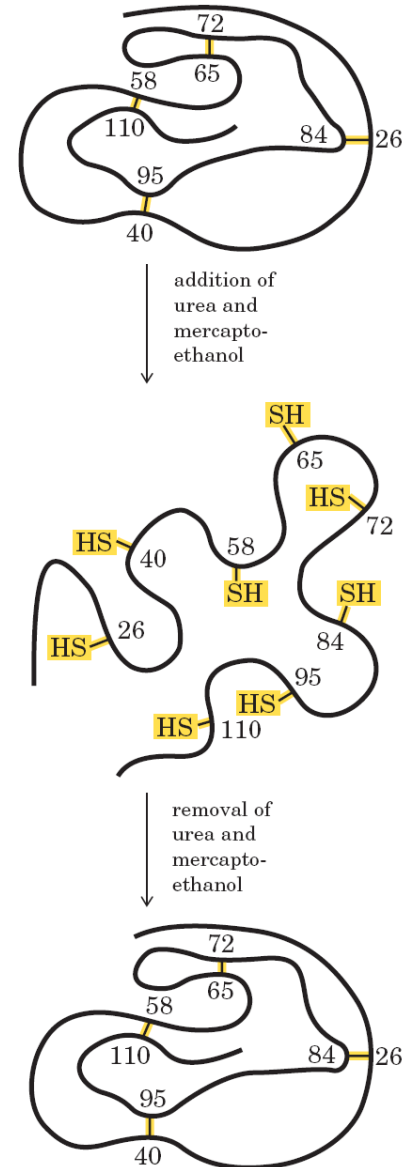
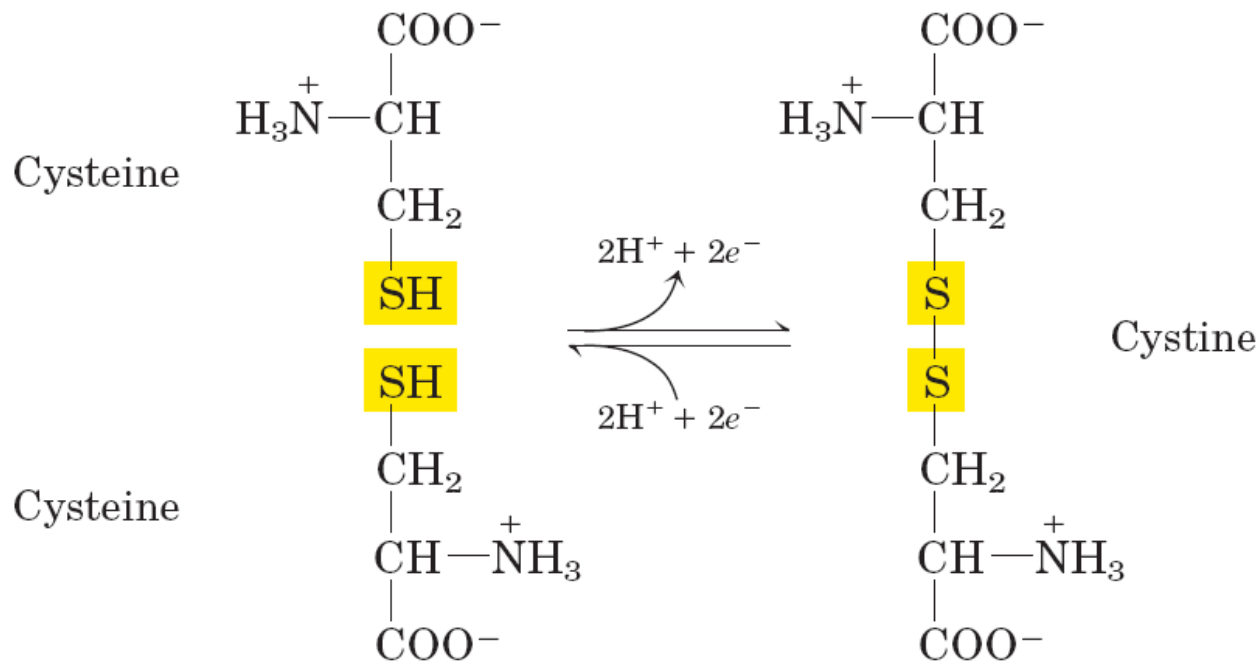


4-Hydroxyproline



5-Hydroxylysine

Nastanek disulfidne vezi: stabilizacija 3D-strukture



Samosestavljanje polipeptidne verige: od primarne do terciarne (kvartarne) strukture

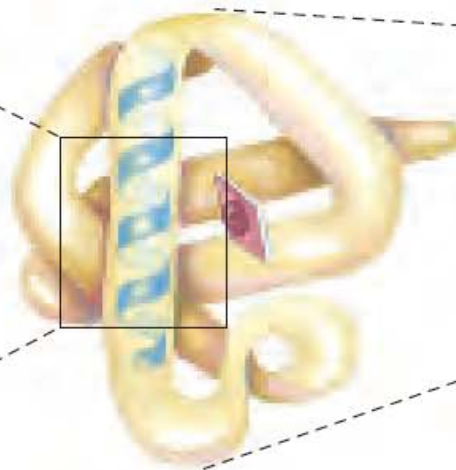
Primary structure



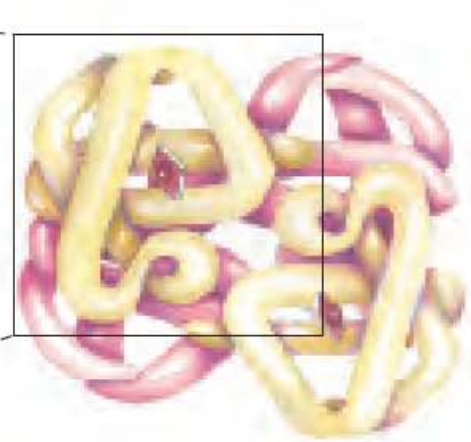
Secondary structure



Tertiary structure



Quaternary structure



Amino acid residues

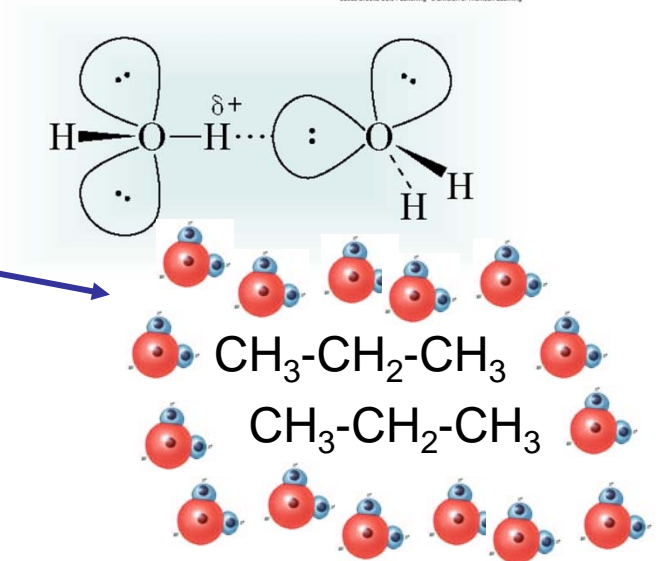
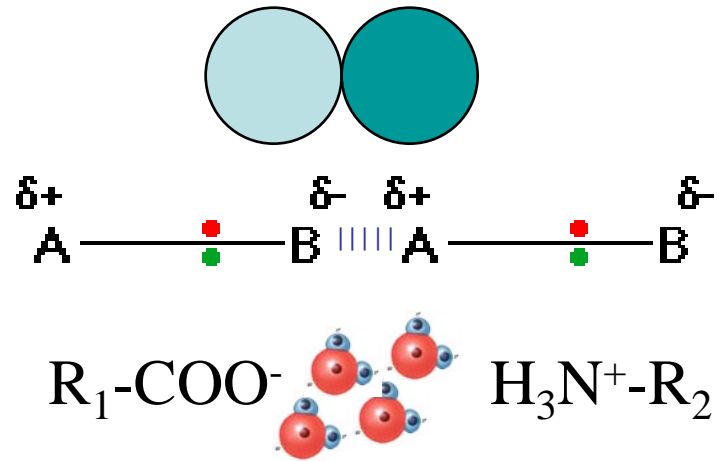
α Helix

Polypeptide chain

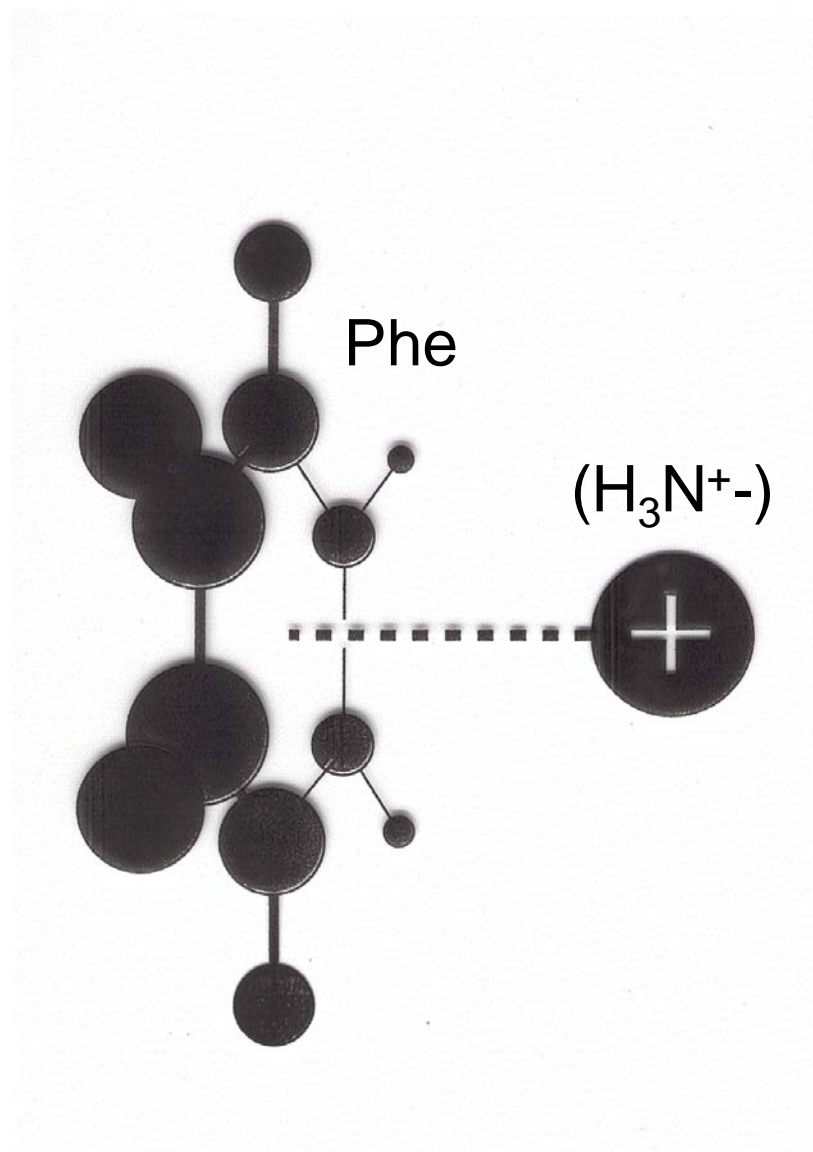
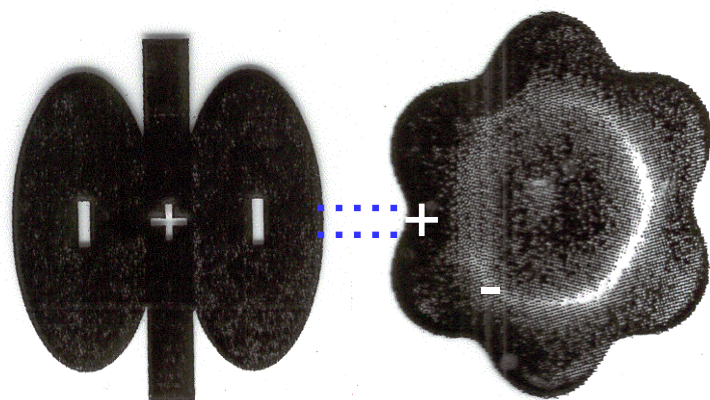
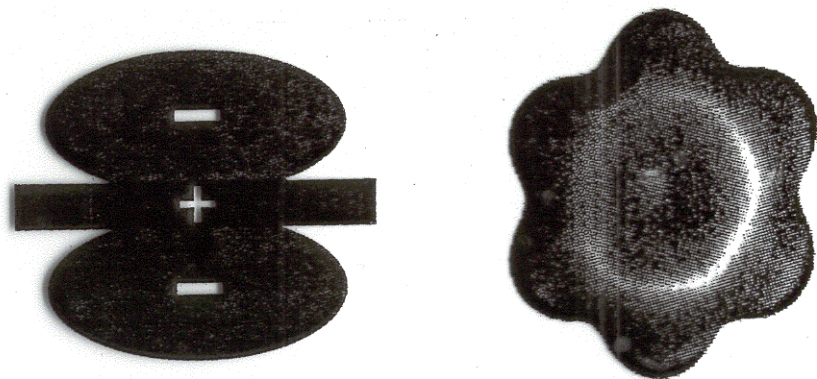
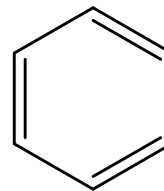
Assembled subunits

ŠIBKE INTERAKCIJE

- odboj pri dotiku
- van der Waalsove interakcije
- elektrostatske interakcije
- vodikova vez
- hidrofobne interakcije
- ion - π interakcije (aromatske AK)



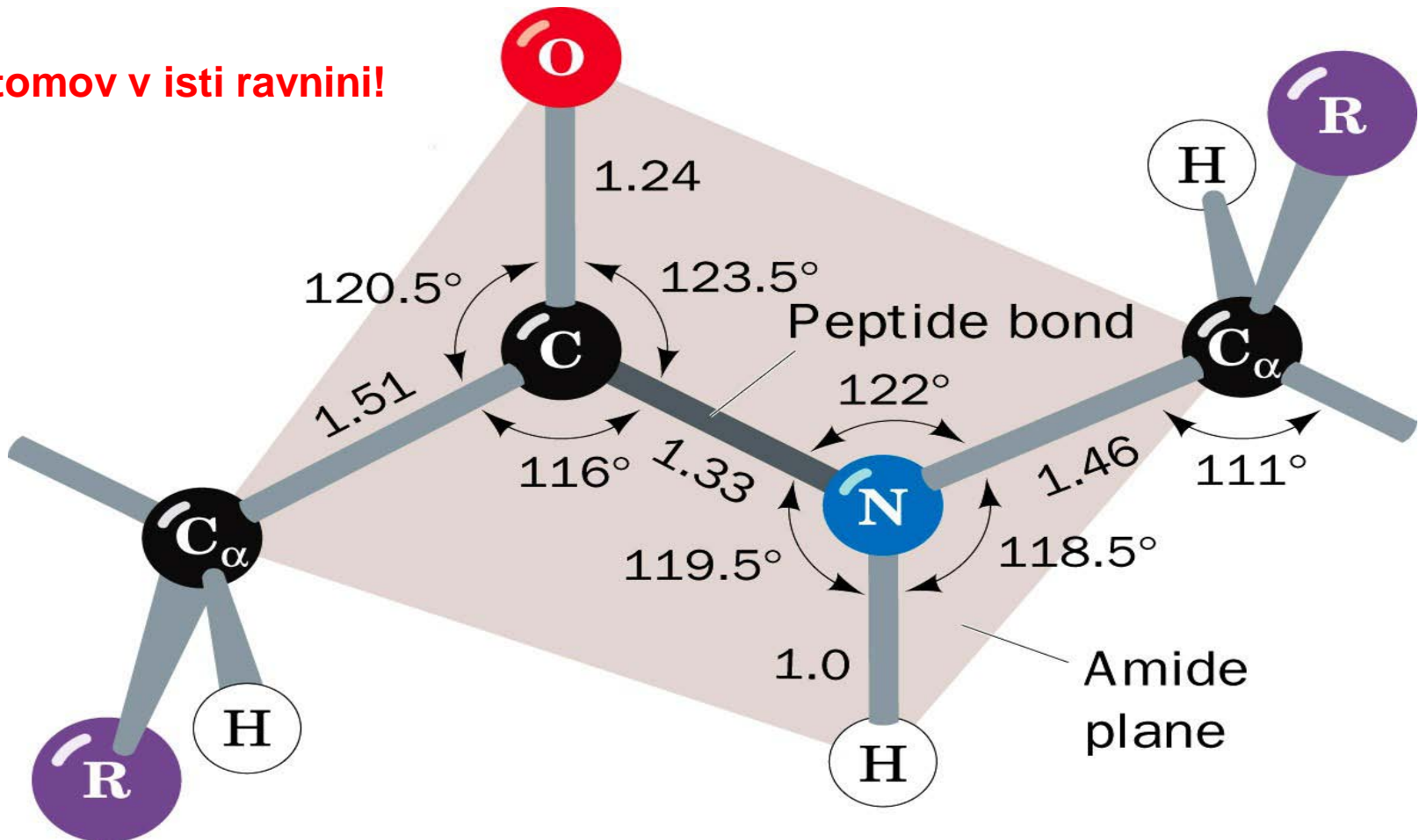
Ion - π interakcije:



Peptidna vez

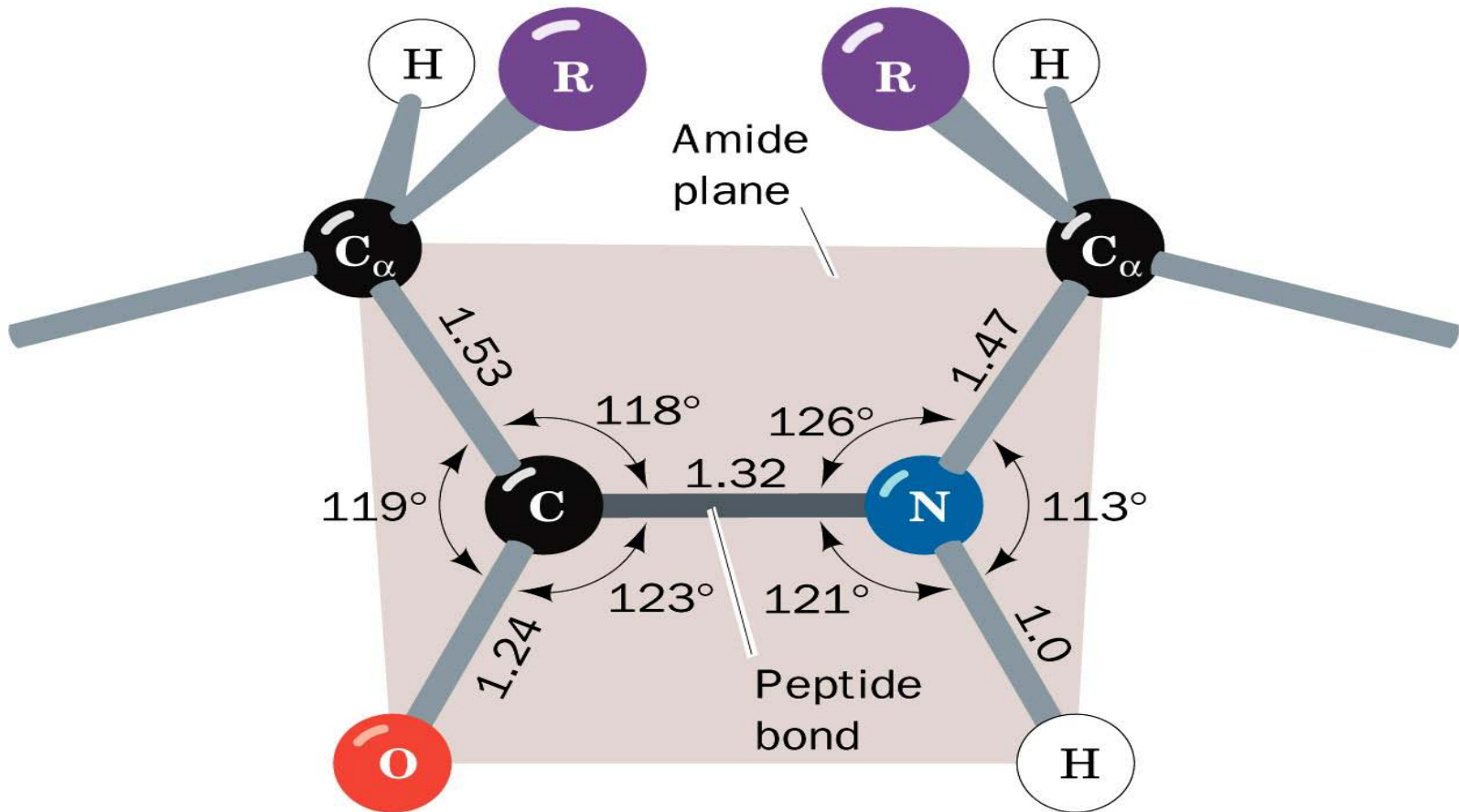


6 atomov v isti ravnini!



Velika večina peptidnih vezi je v trans konfiguraciji.

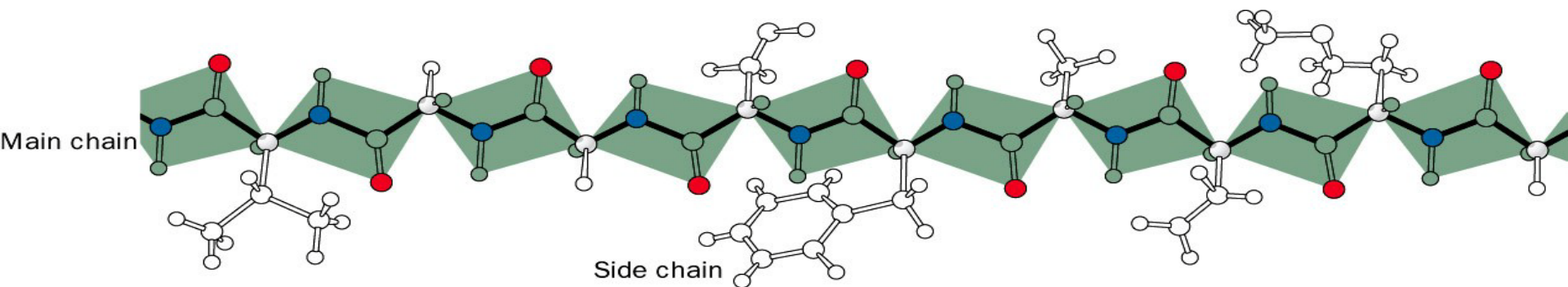
Cis-peptidna vez



V zavojih ob Pro je približno 6% peptidnih vezi cis

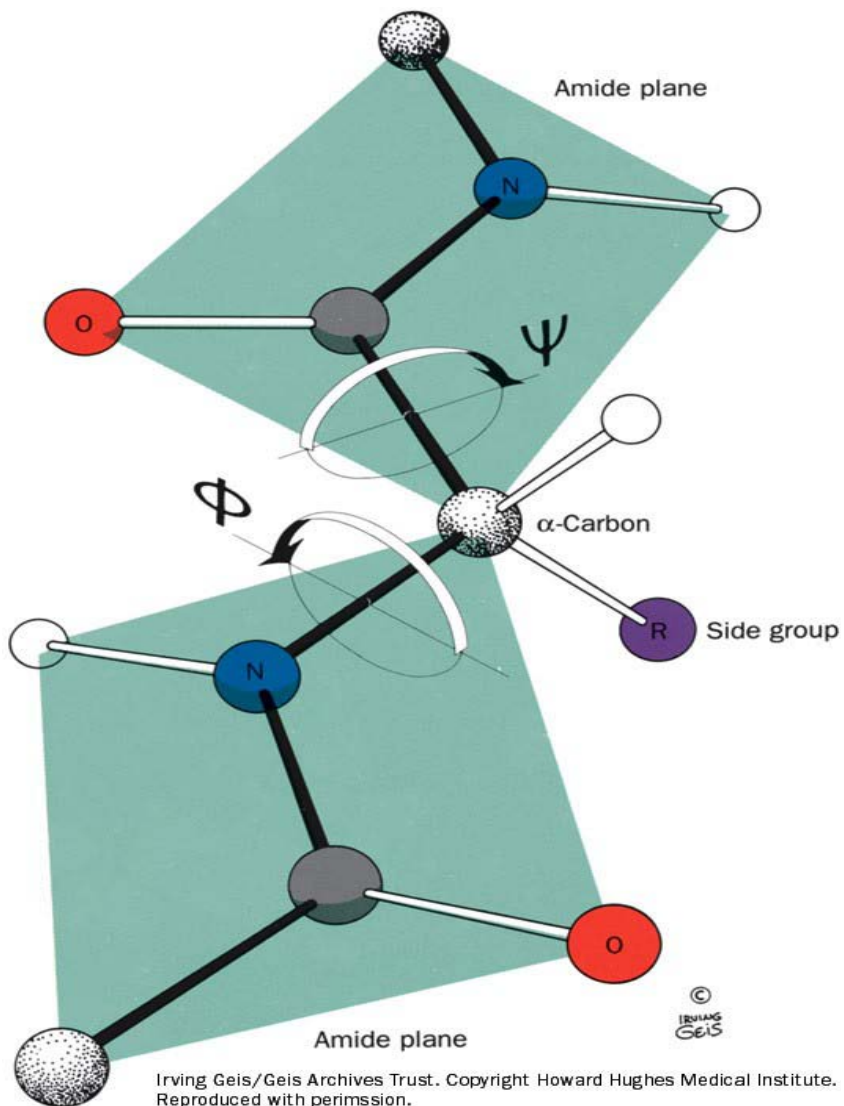
Polipeptidna veriga v iztegnjeni konformaciji

**POZOR: OMEJENA GIBLJIVOST ZARADI OMEJITEV
V VRTLJIVOSTI OKROG PEPTIDNE VEZI.**

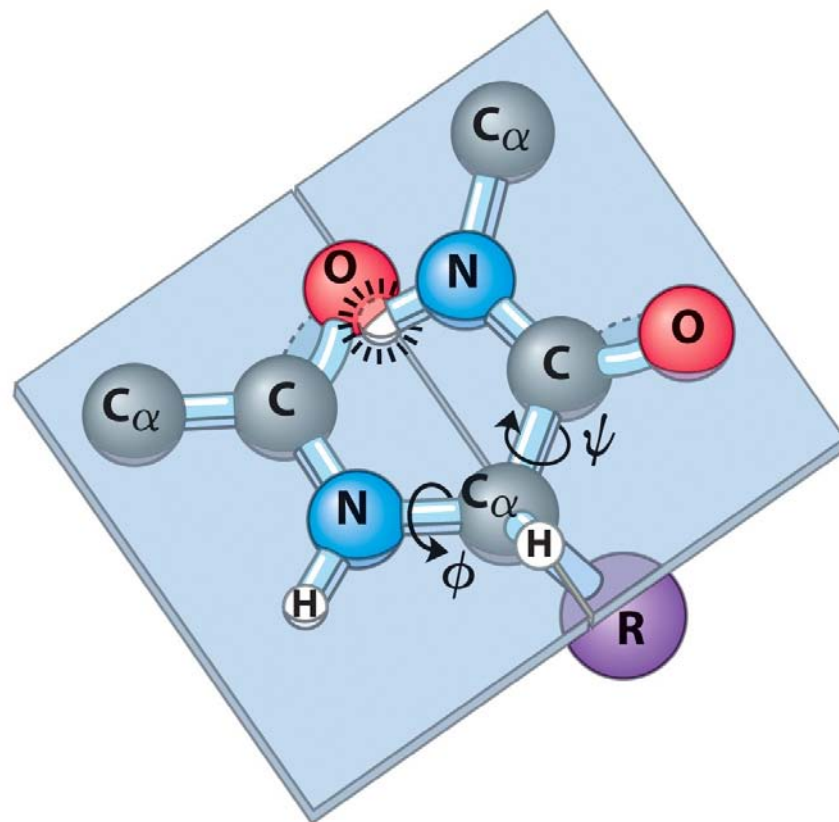


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Dihedralna kota Φ (fi) in Ψ (psi)



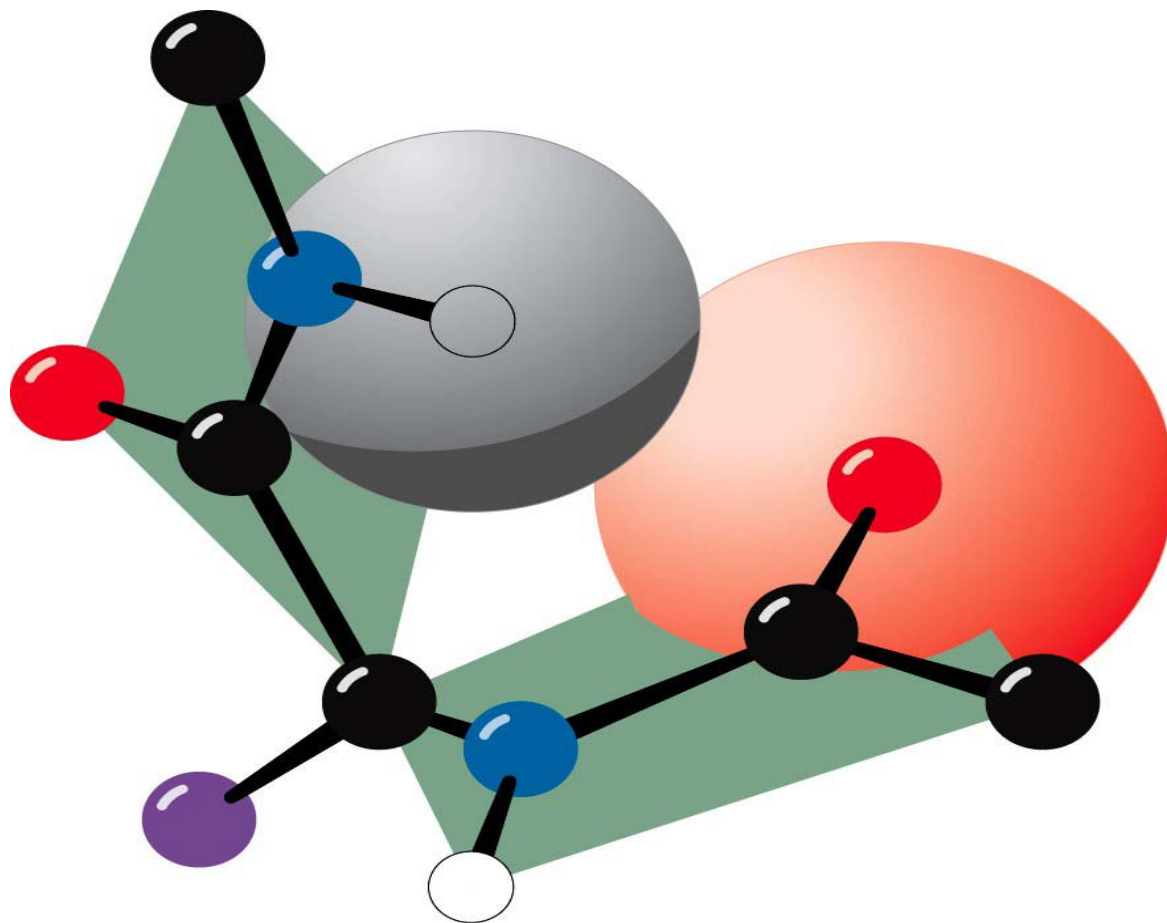
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Začetni položaj:

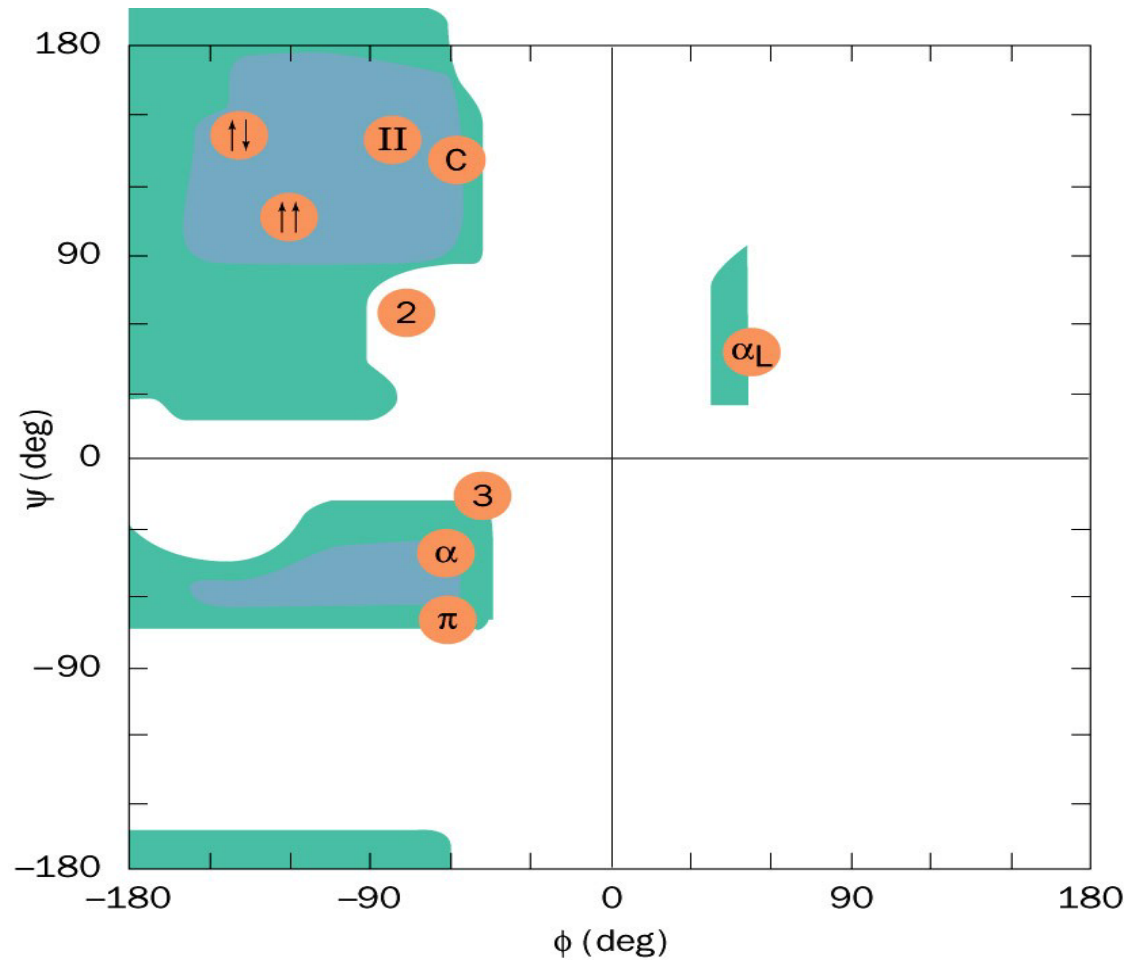
$$\Phi = \Psi = 0 \text{ deg}$$

Sterično oviranje pri vrtenju sosednjih AK ostankov

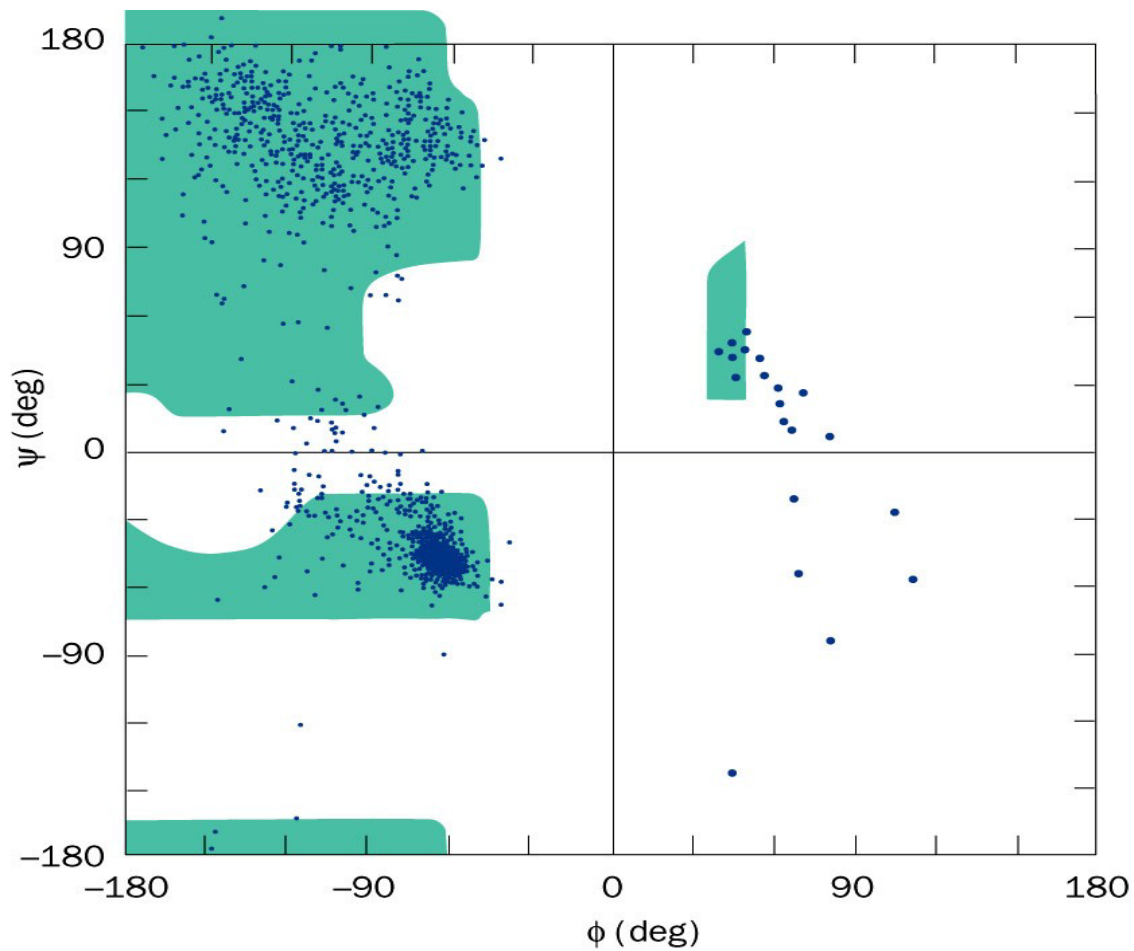


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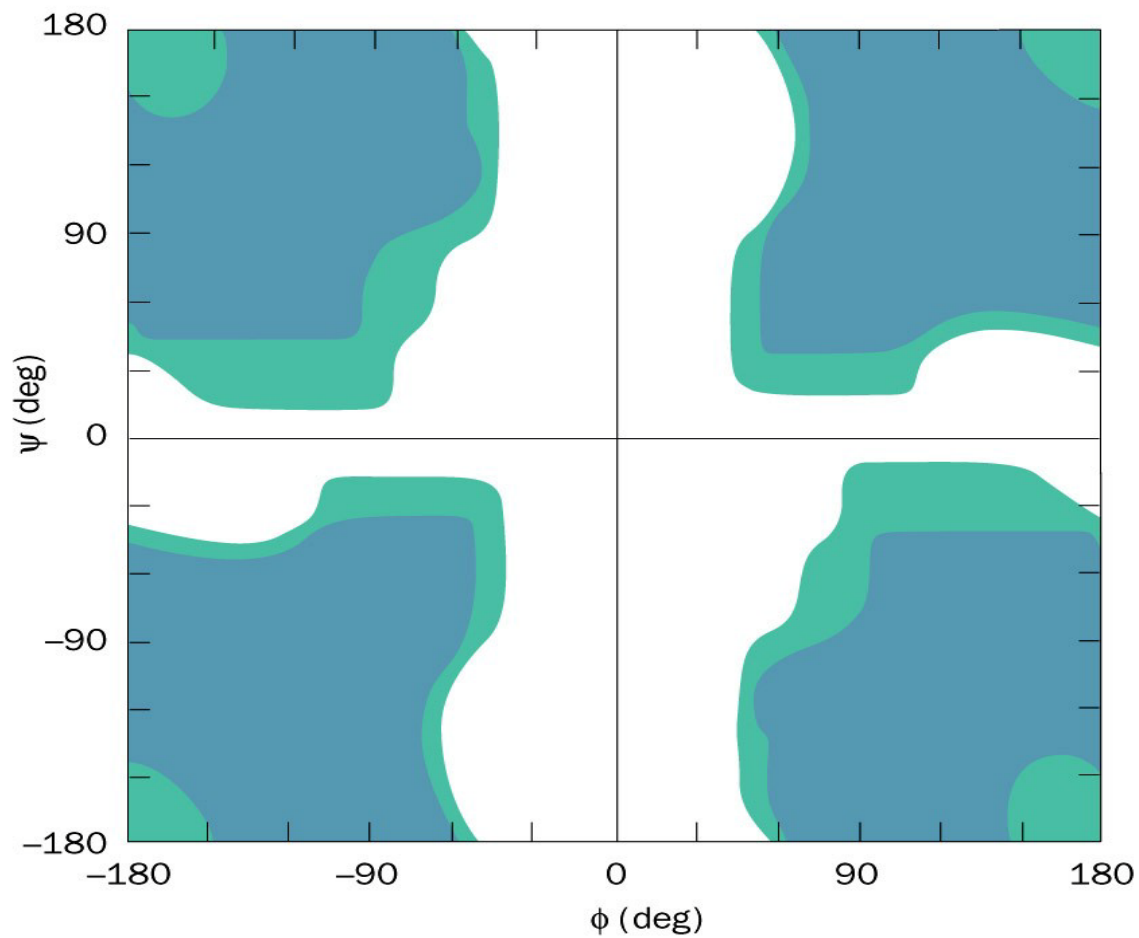
Ramachandranov diagram kaže 'dovoljena' področja v 2D prostoru $\Phi - \Psi$



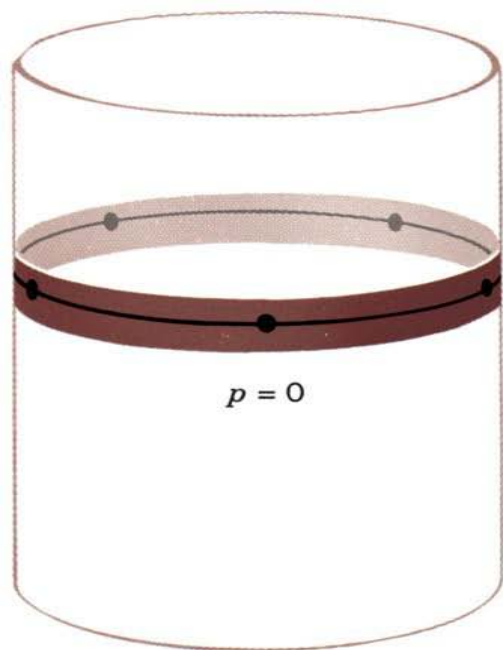
Izmerjene vrednosti Φ in Ψ pri proteinih



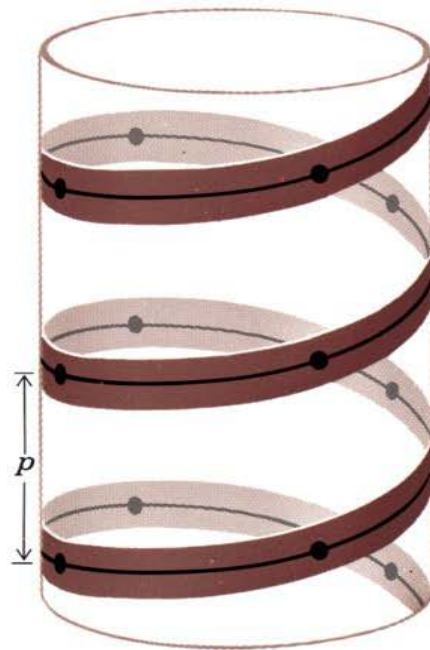
Ramachandranov diagram za poli-Gly



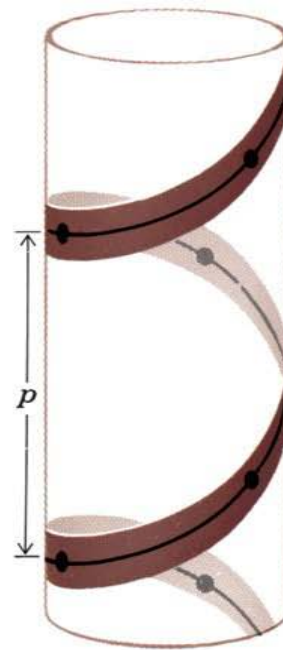
Različni heliksi (vijačnice)



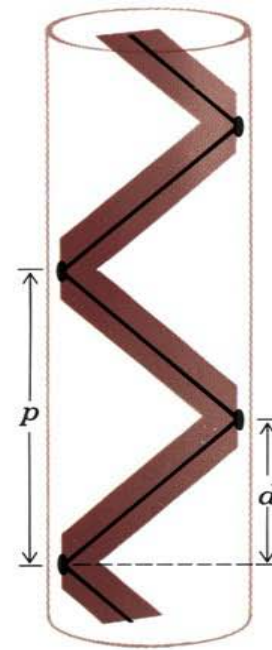
$n = 5$



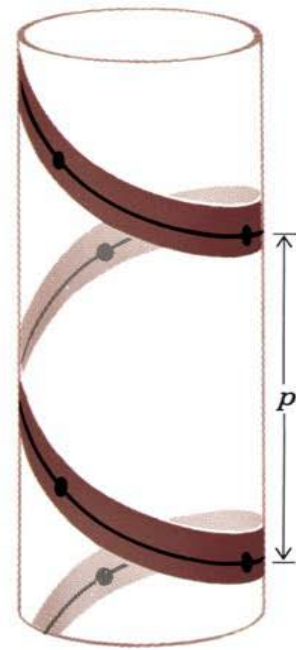
$n = 4$



$n = 3$



$n = 2$

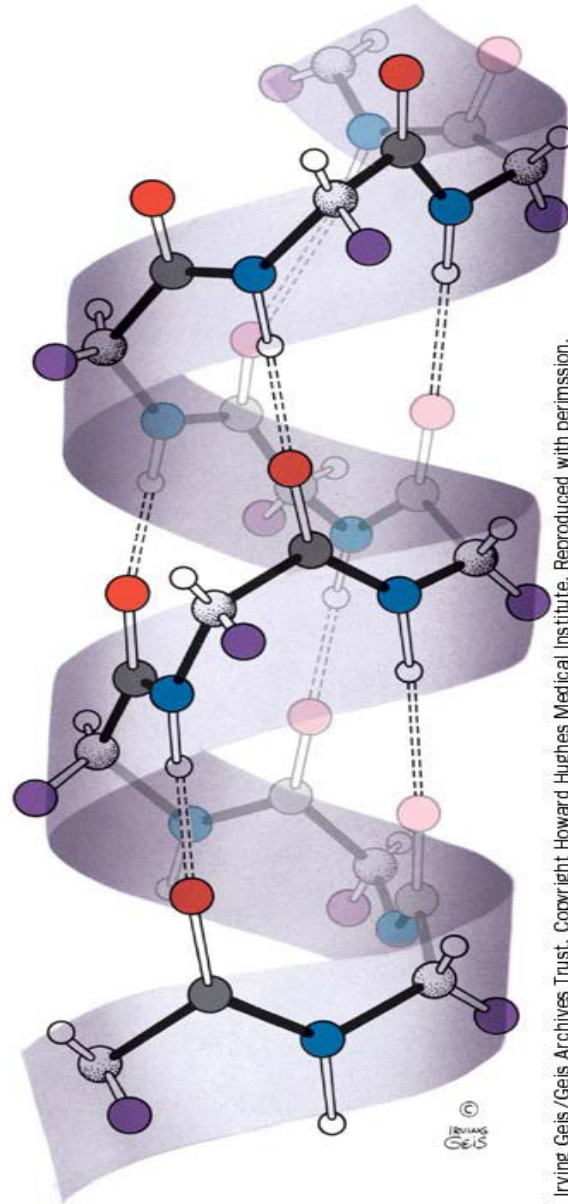


$n = -3$

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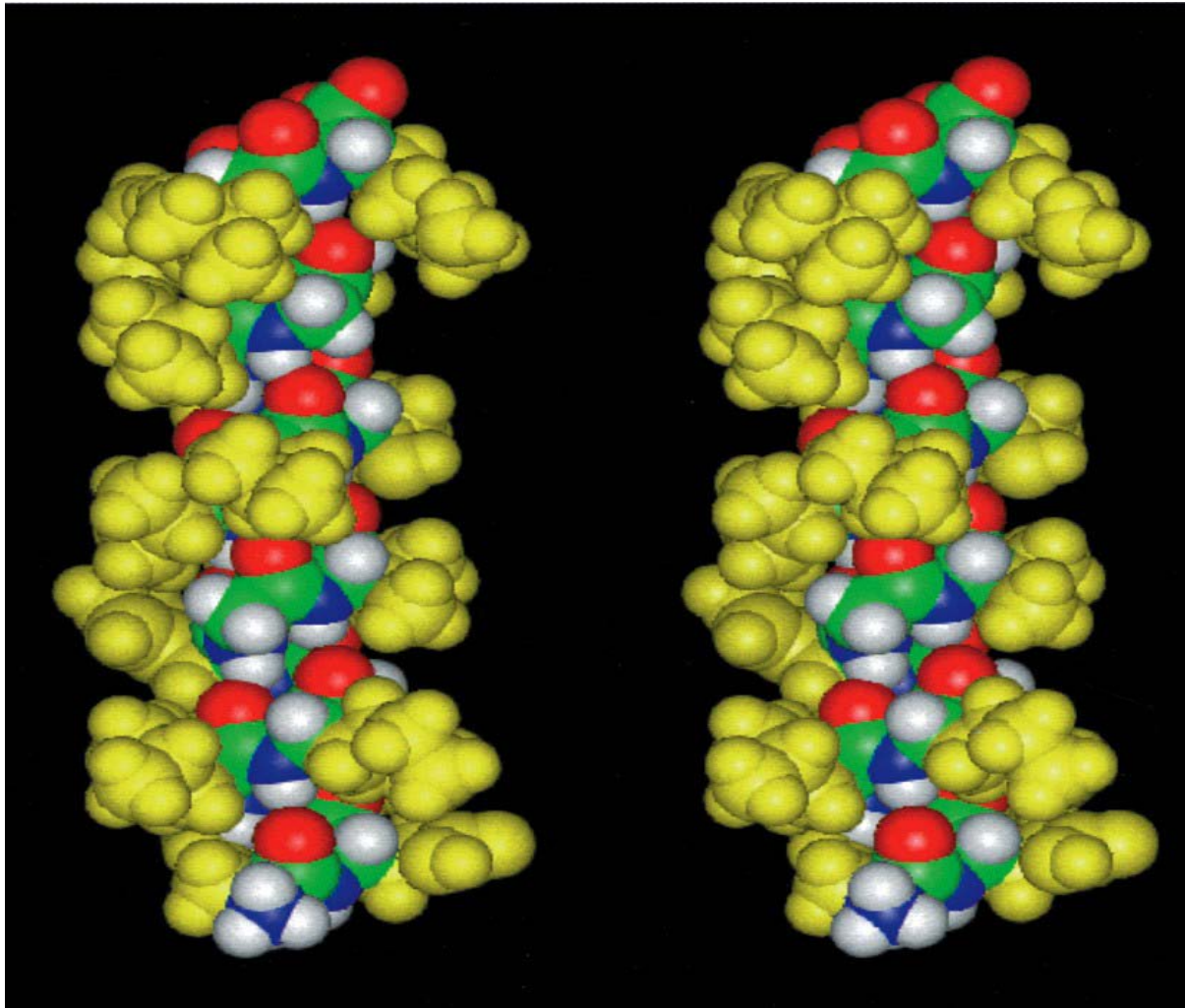
desnosučna α -vijačnica



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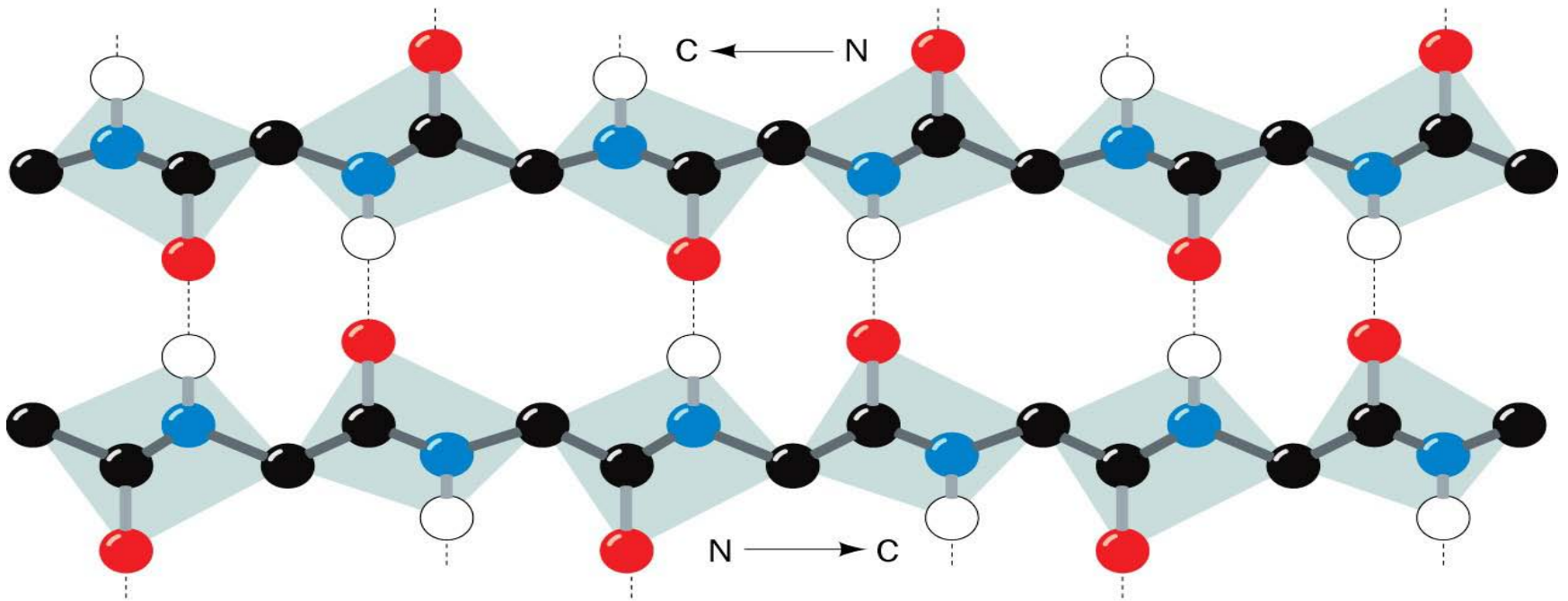
Stereoskopski prikaz α -vijačnice (H_E Mb)

POZOR: AK RADIKALI MOLIJU NAVZVEN; ČE SO VELIKI SKUPAJ, SE OVIRAJO; V NOTRANJOSTI NI VRZELI (NITI ZA MOLEKULE VODE).



β struktura (antiparalelna)

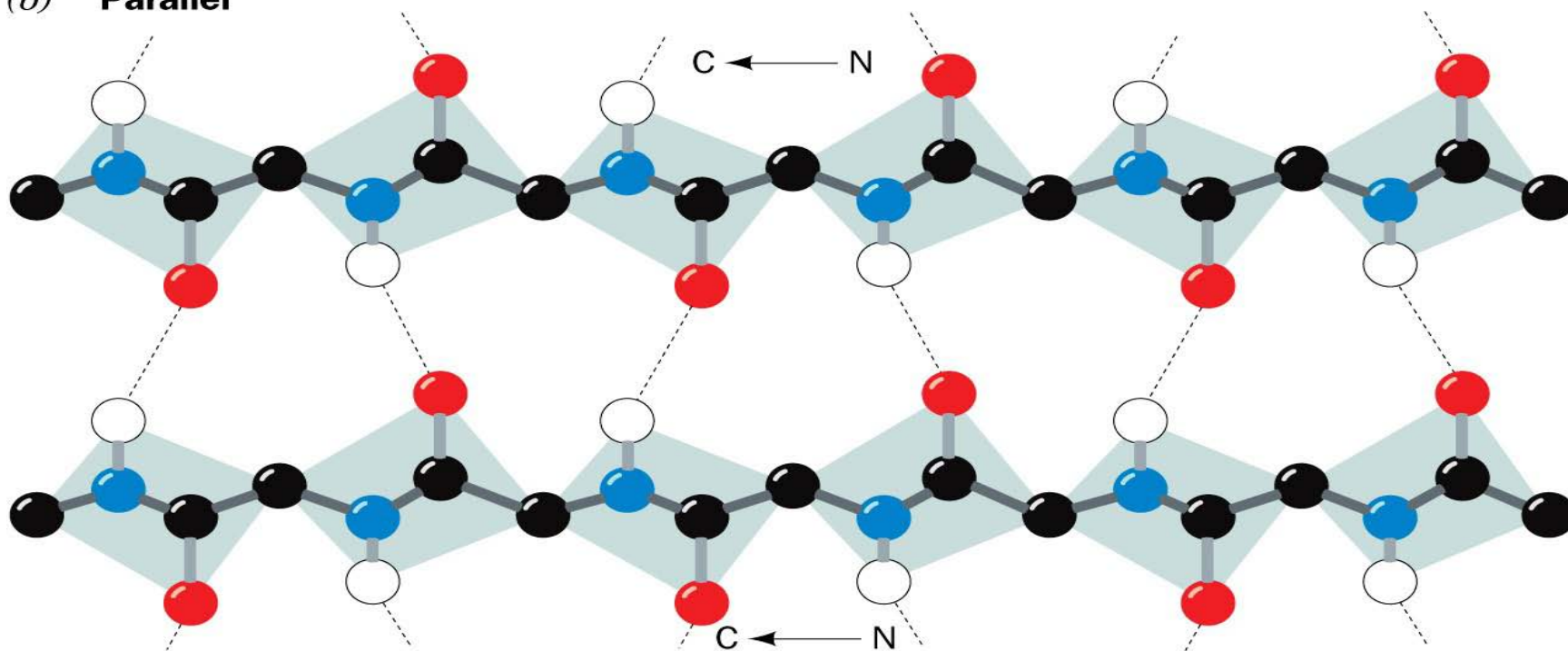
(a) **Antiparalelna**



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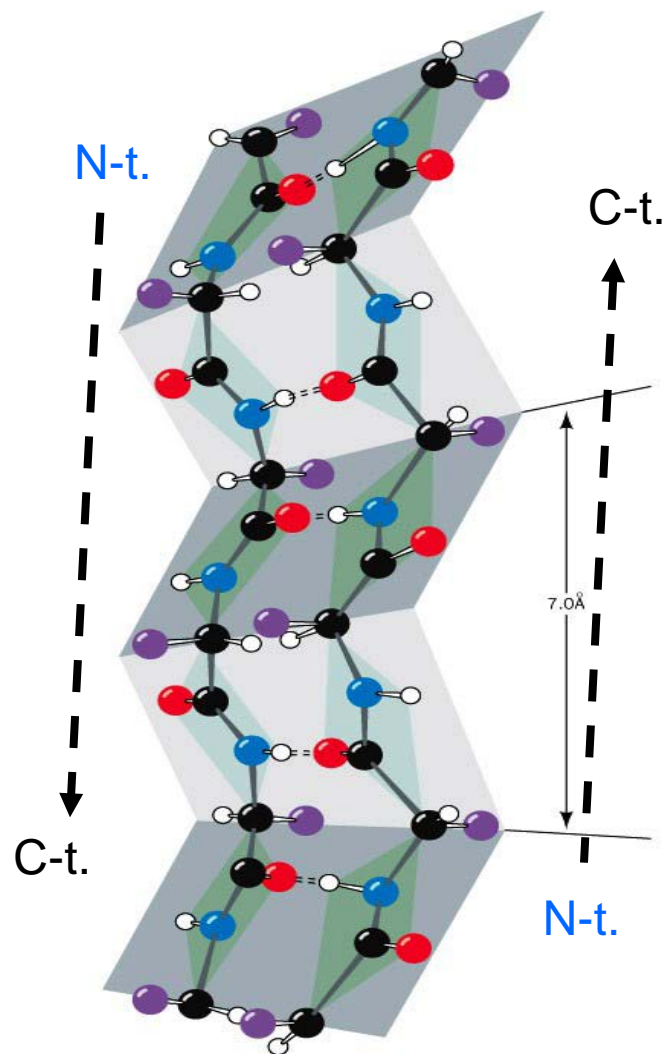
β struktura (paralelna)

(b) **Parallel**

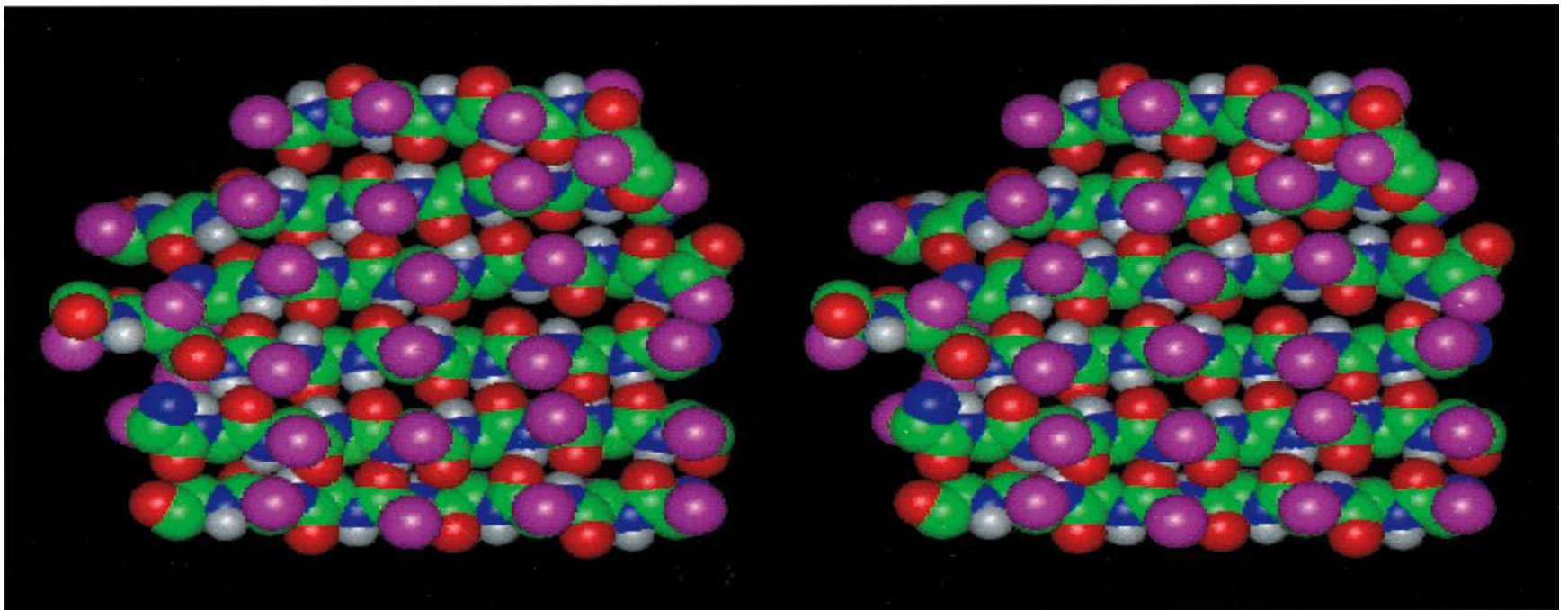


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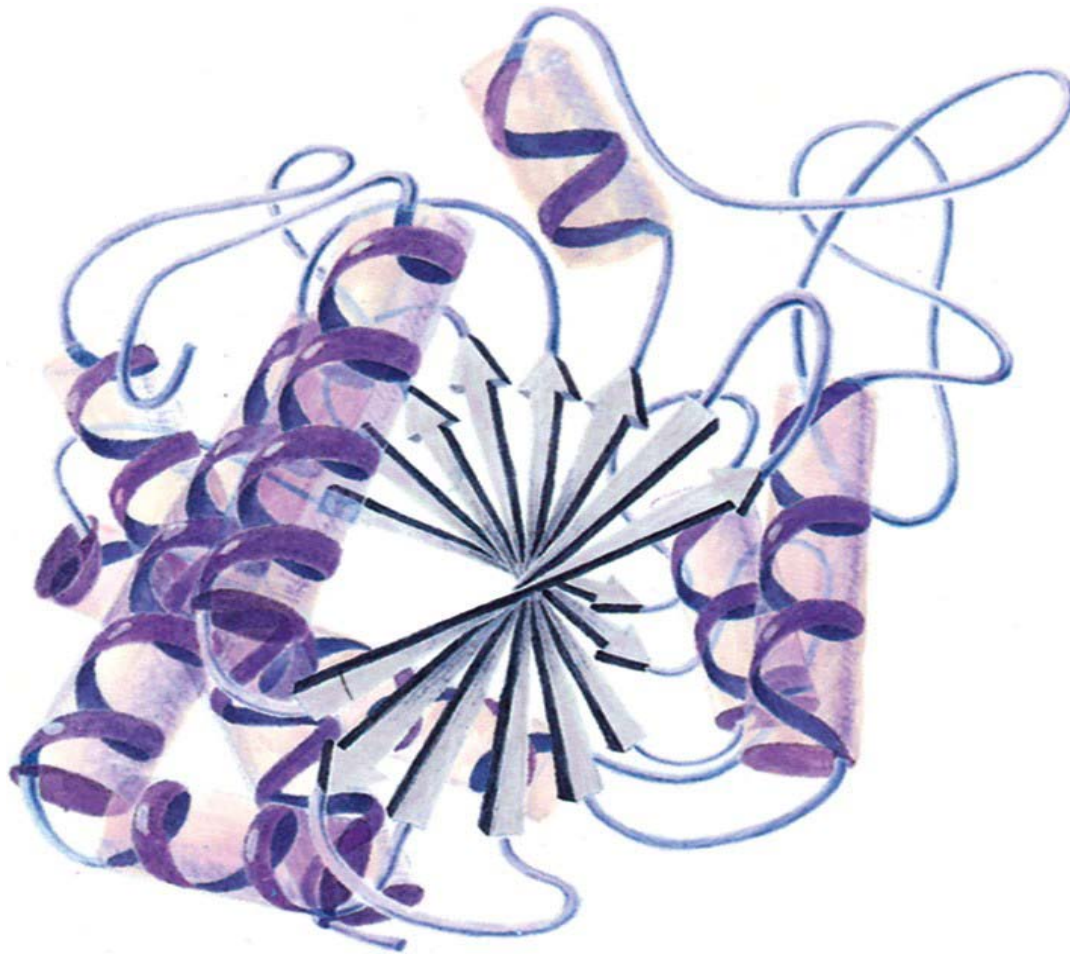
Nagubani list



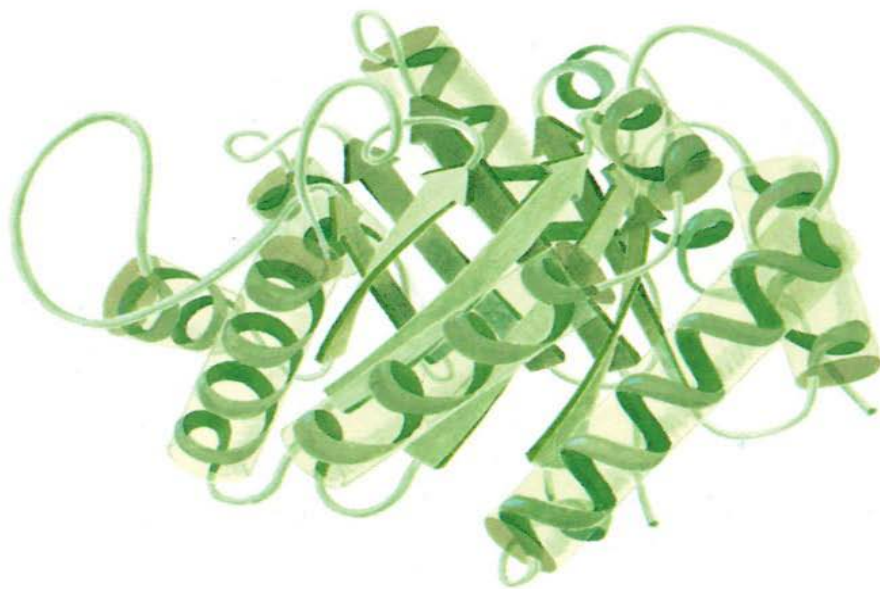
Stereoskopski prikaz nagubane lista iz 6 verig



Nagubani list je pogosto zvit in podprt z vijačnicami.



Nagubani list je pogosto zvit in podprt z vijačnicami
(trioza-fosfat izomeraza)



Povezave verig v nagubanem listu

(a)



(b)



ANTIPARALELNA

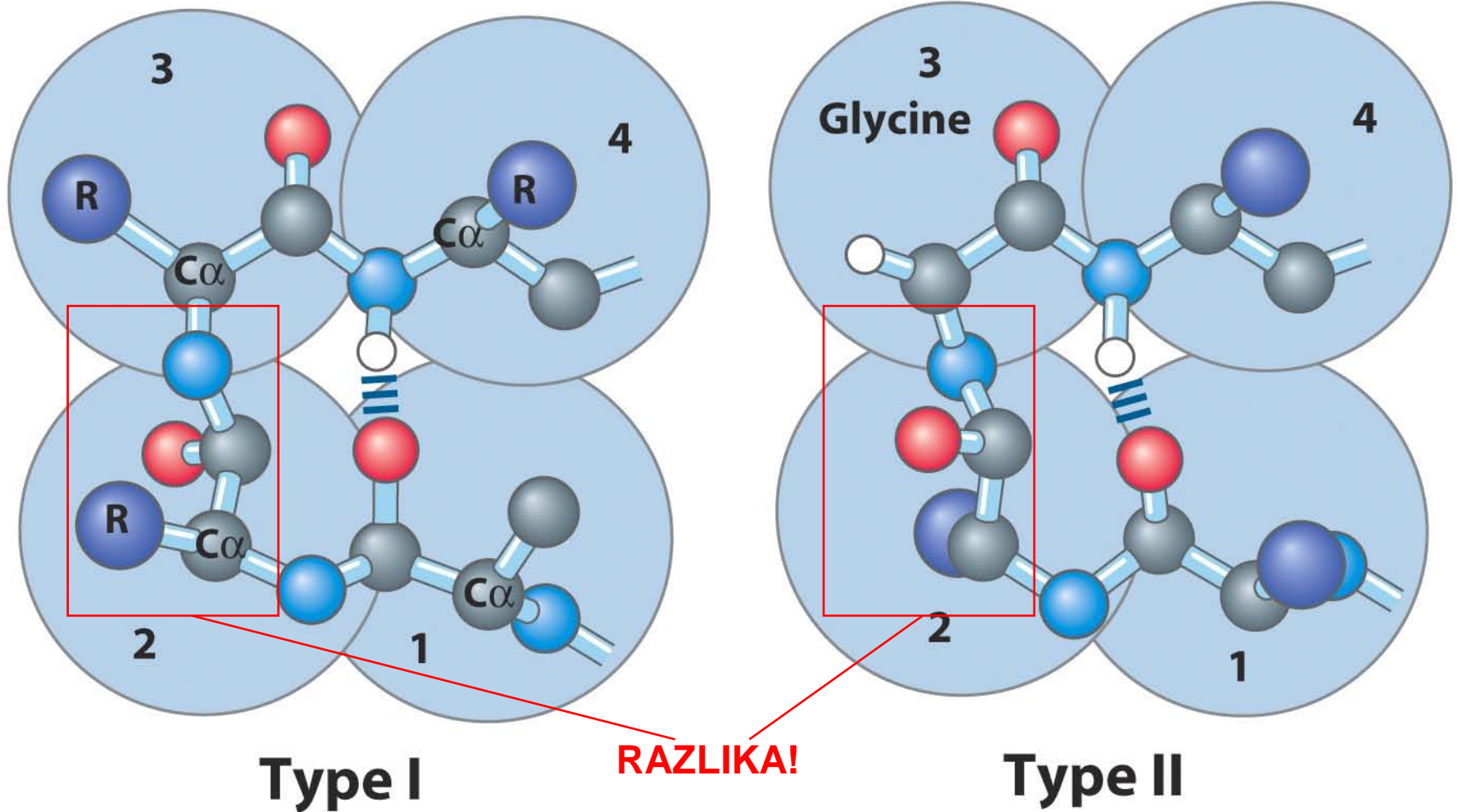
(c)



PARALELNA

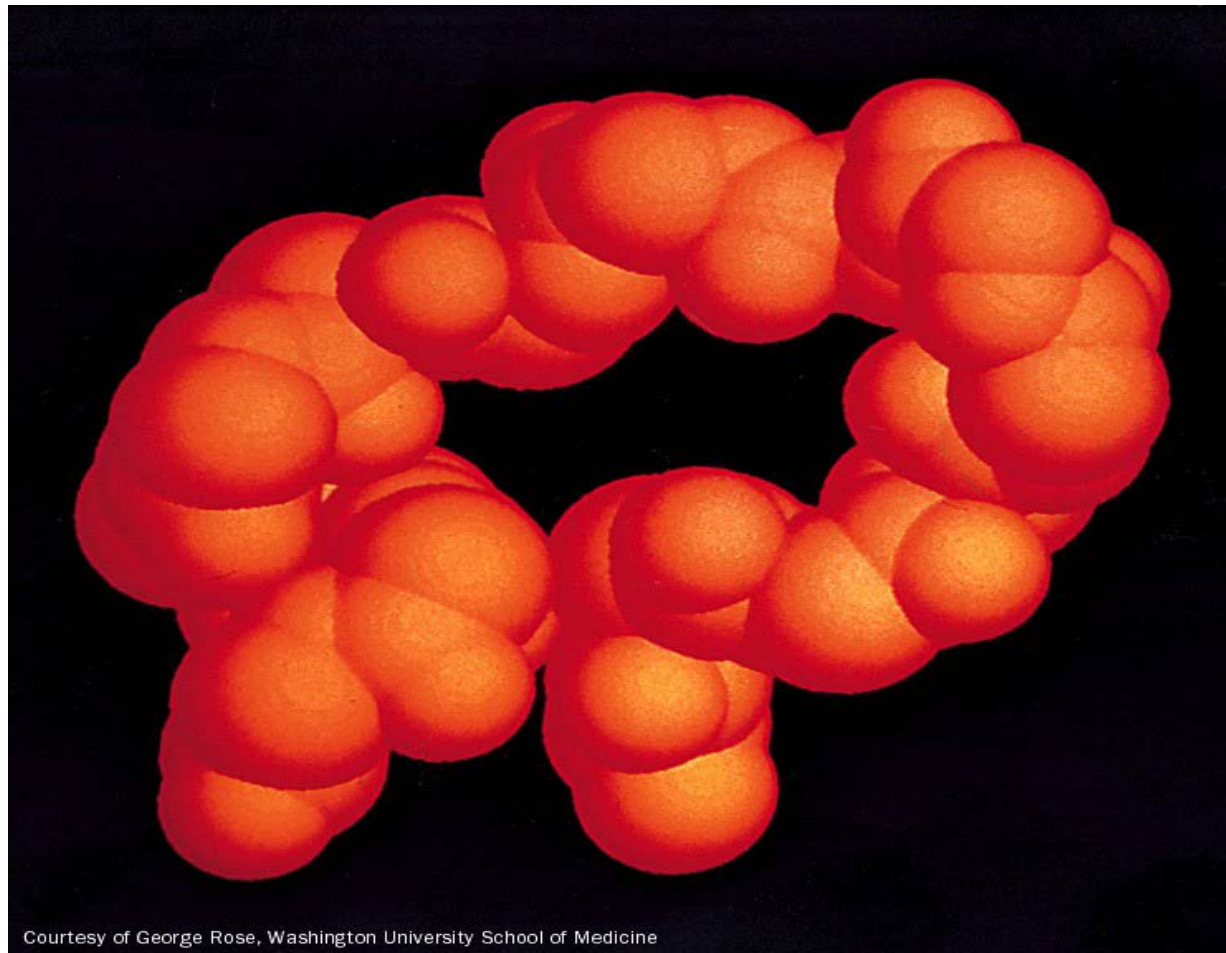
β -zavoji

(a) β Turns

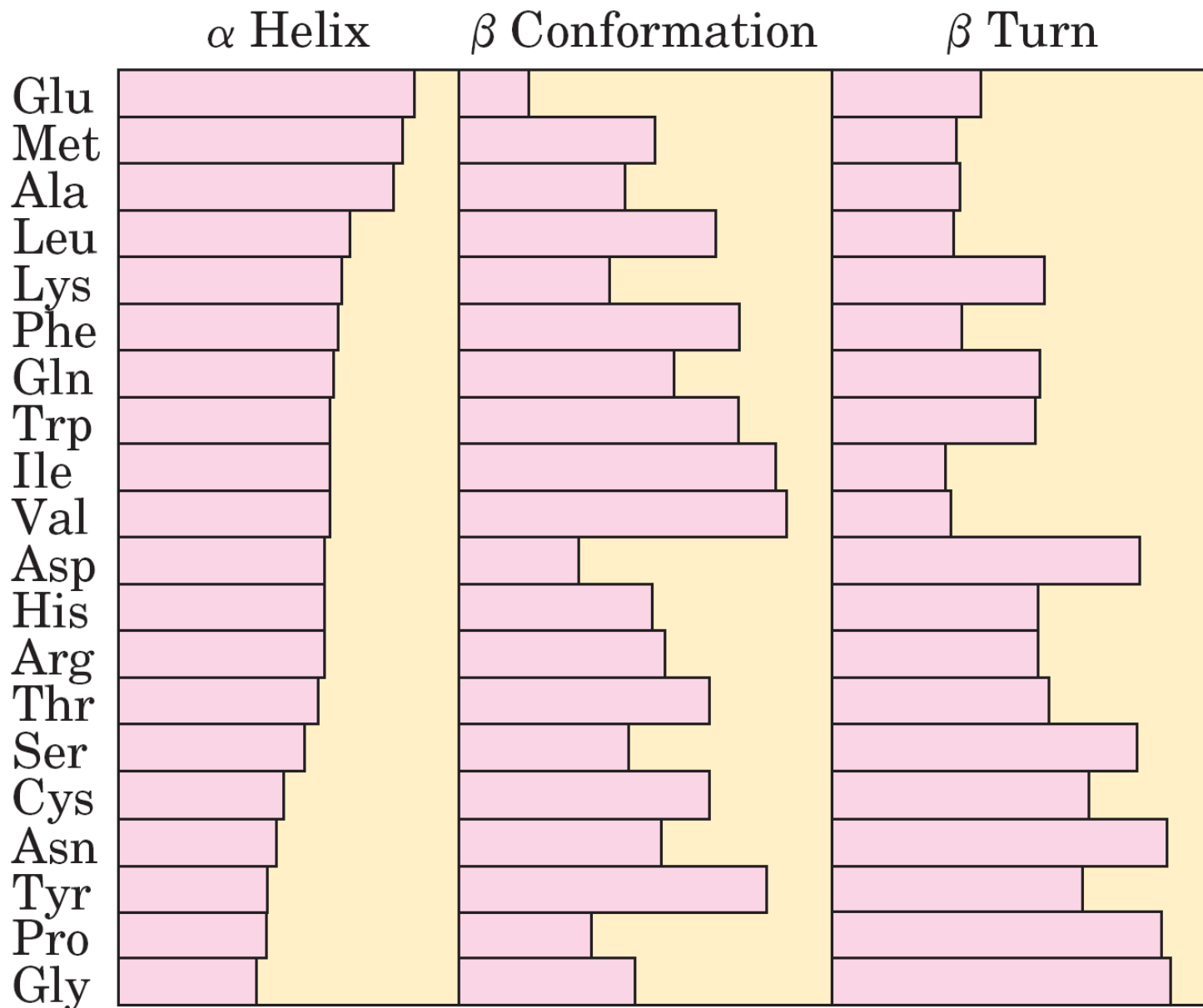


Zanke: Ω zanka je navadno iz 40 do 54 AK
(citokrom c)

Ω

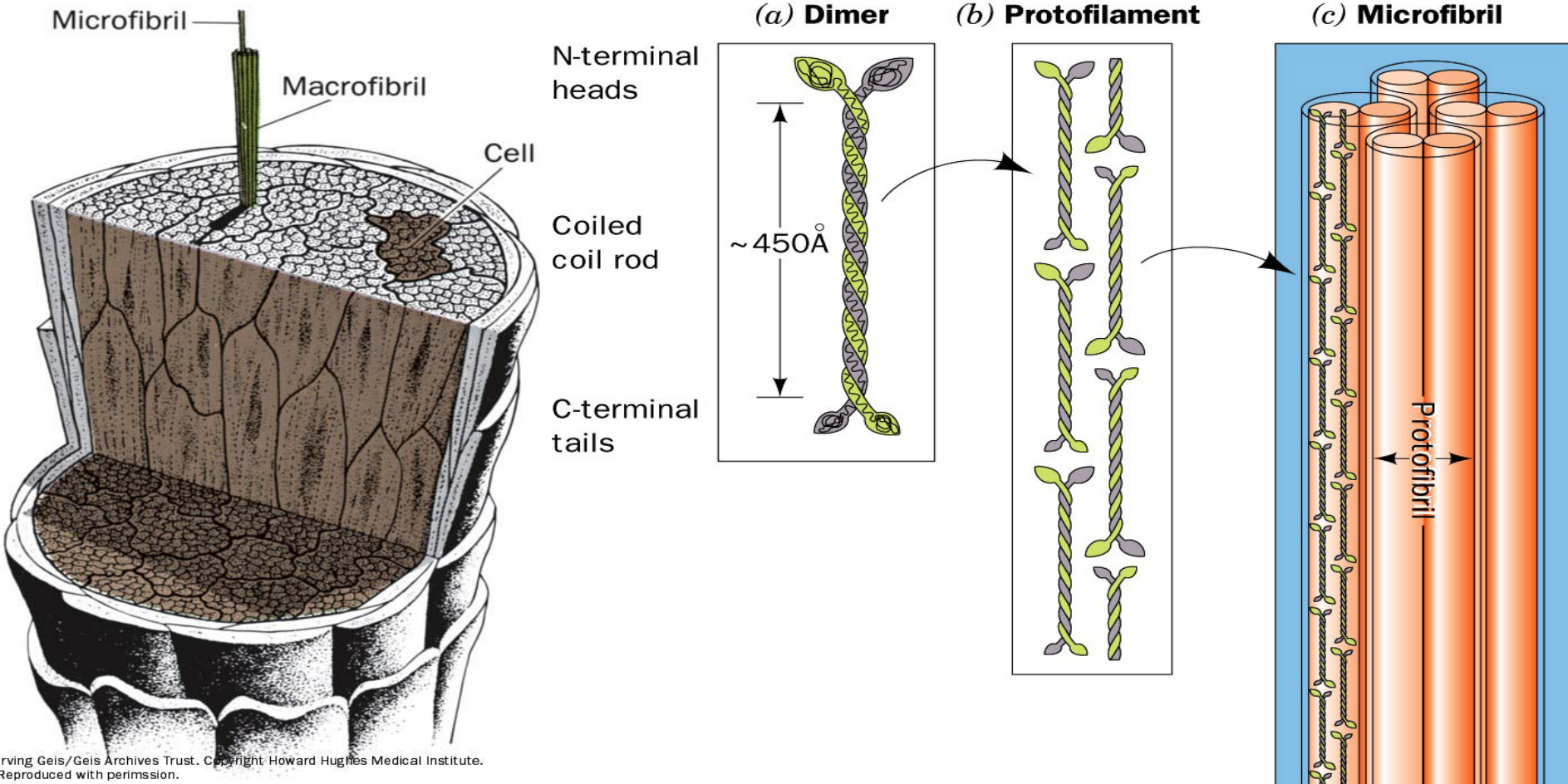


Verjetnost za nahajanje AK v določeni sekundarni strukturi

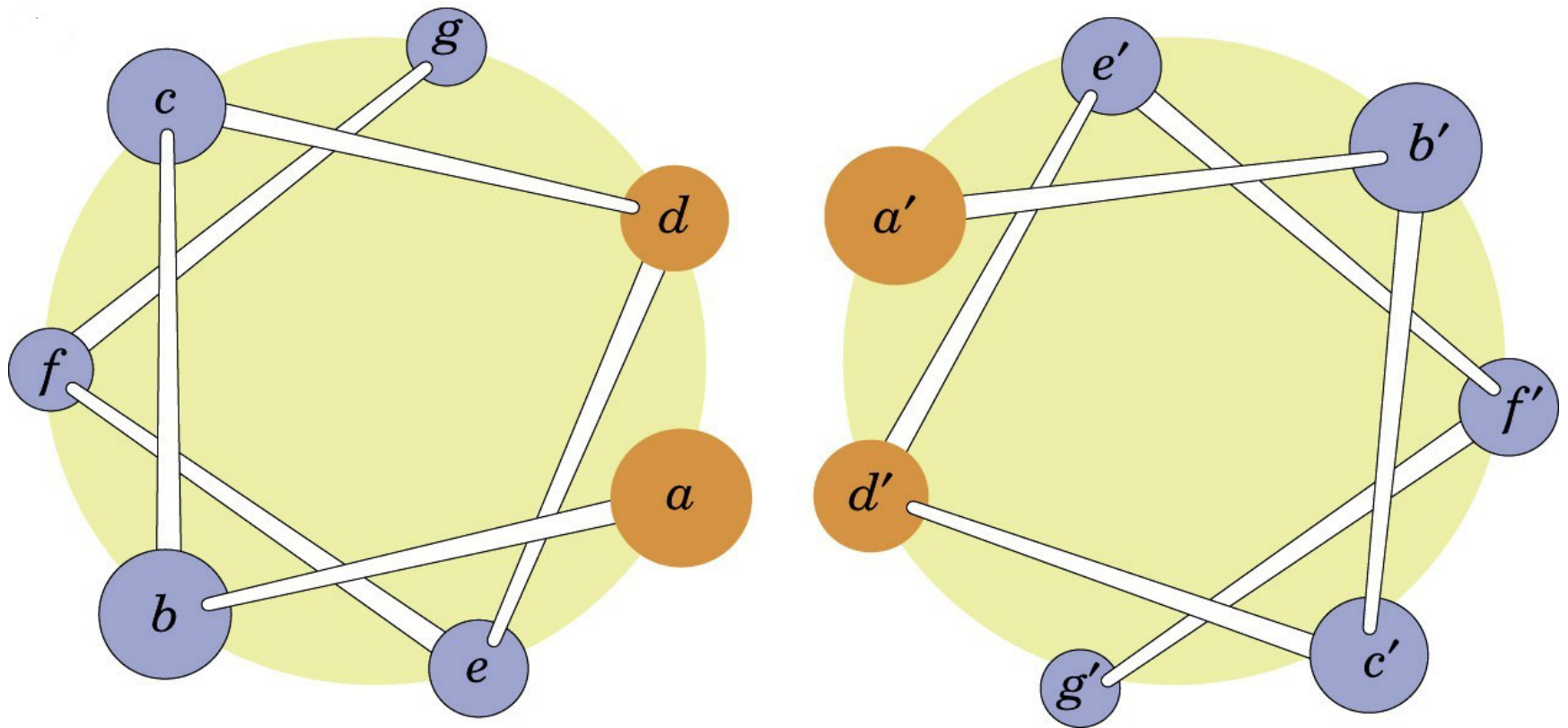


FIBRILARNI PROTEINI

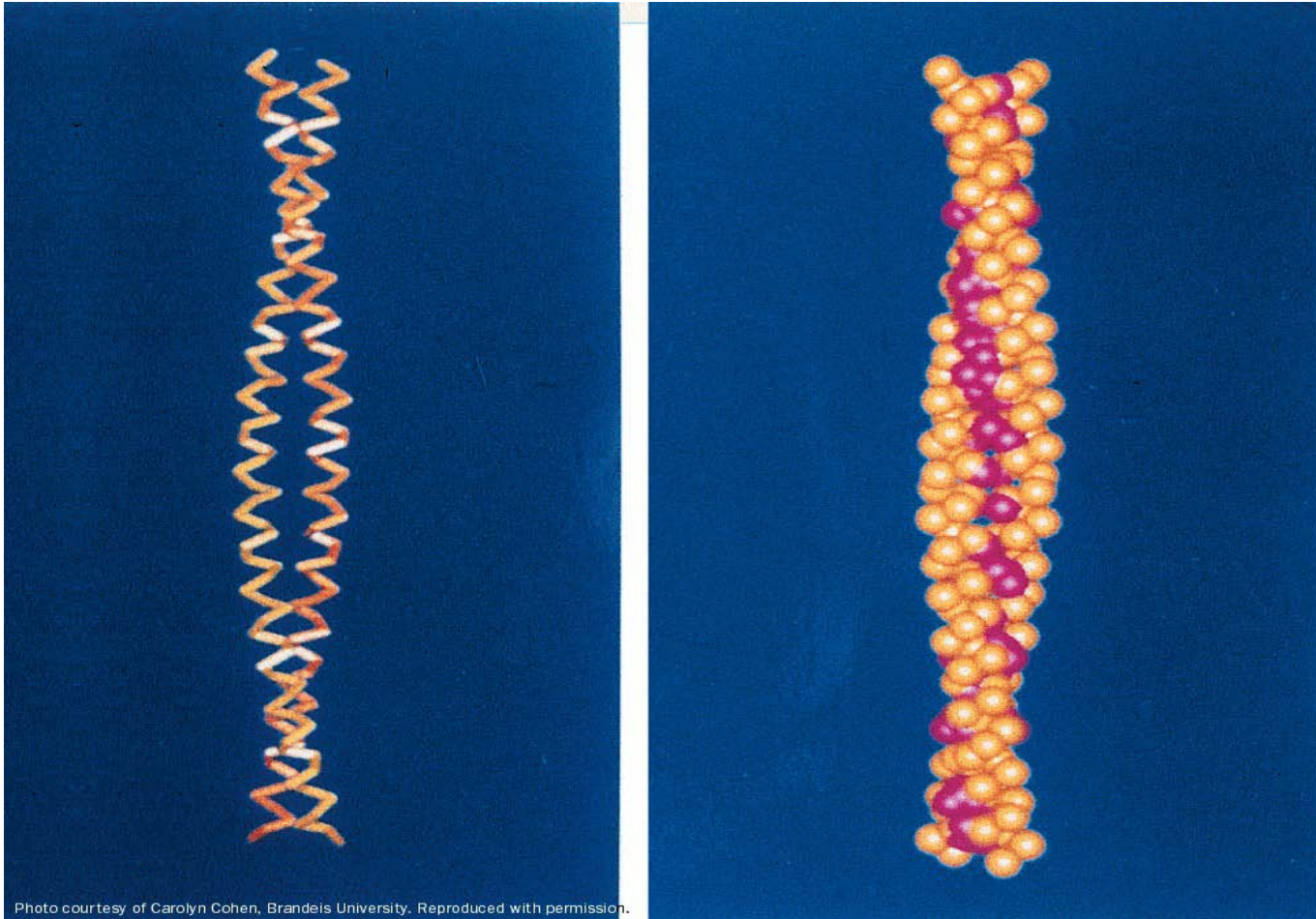
Lasje so iz α -keratina



Nepolarna področja α heliksov držijo skupaj hidrofobne interakcije



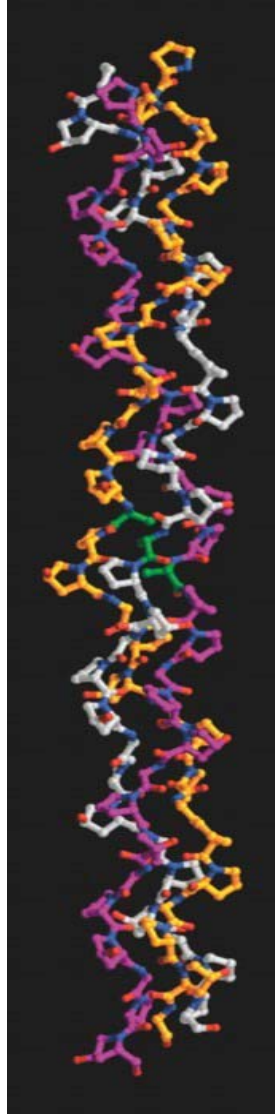
Za optimalno vzpostavitev hidrofobnih vezi se morata vijačnici oviti ena okrog druge.



Trojni heliks v kolagenu (niso α -heliksi!)

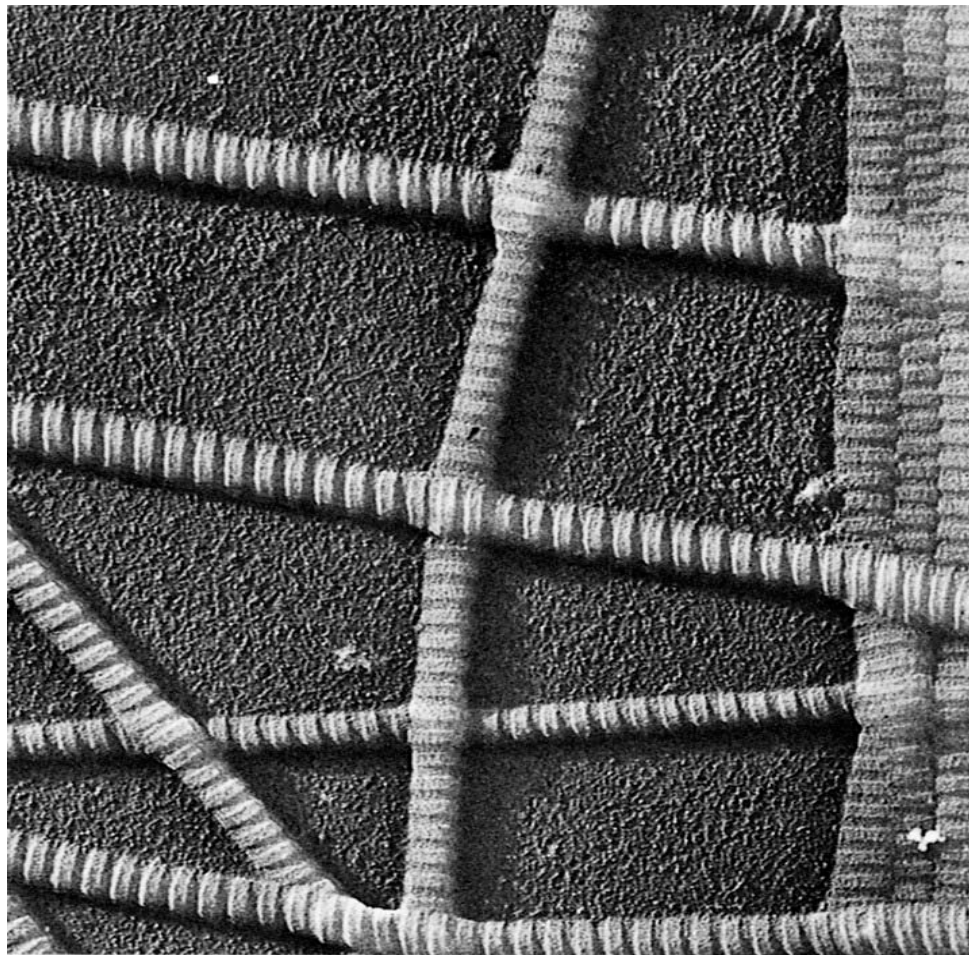



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GEIS
Irving Geis/Geis Archives Trust. Copyright Howard Hughes Medical Institute. Reproduced with permission.



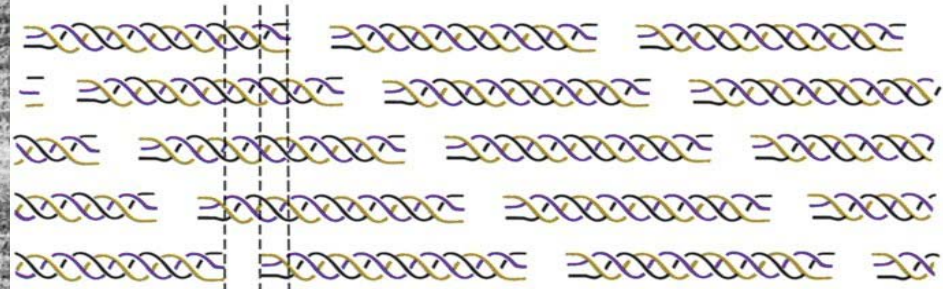
**POZOR: ČE VERIGE LOČIMO,
VIJAČNICE RAZPADEJO**

Struktura kolagenskih vlaken



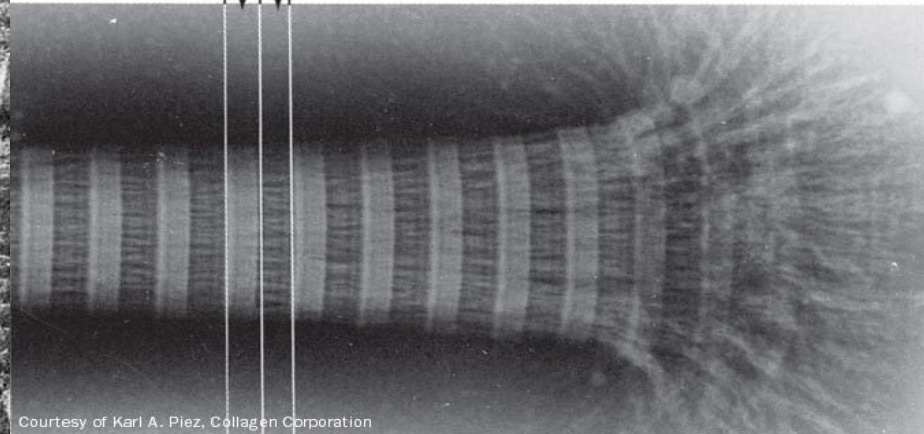
Collagen molecule 

Packing of molecules



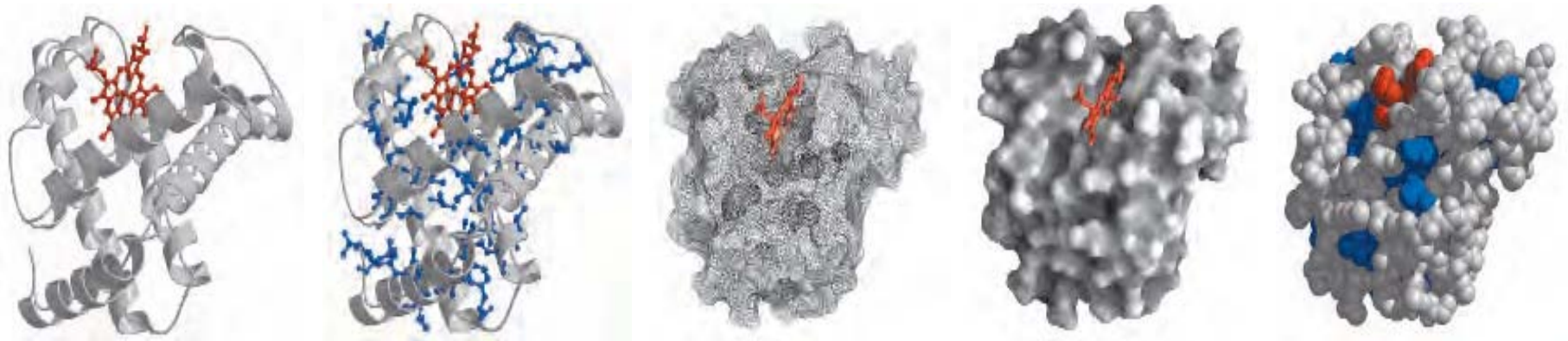
Hole zone
0.6D

Overlap zone
0.4D



Courtesy of Karl A. Piez, Collagen Corporation

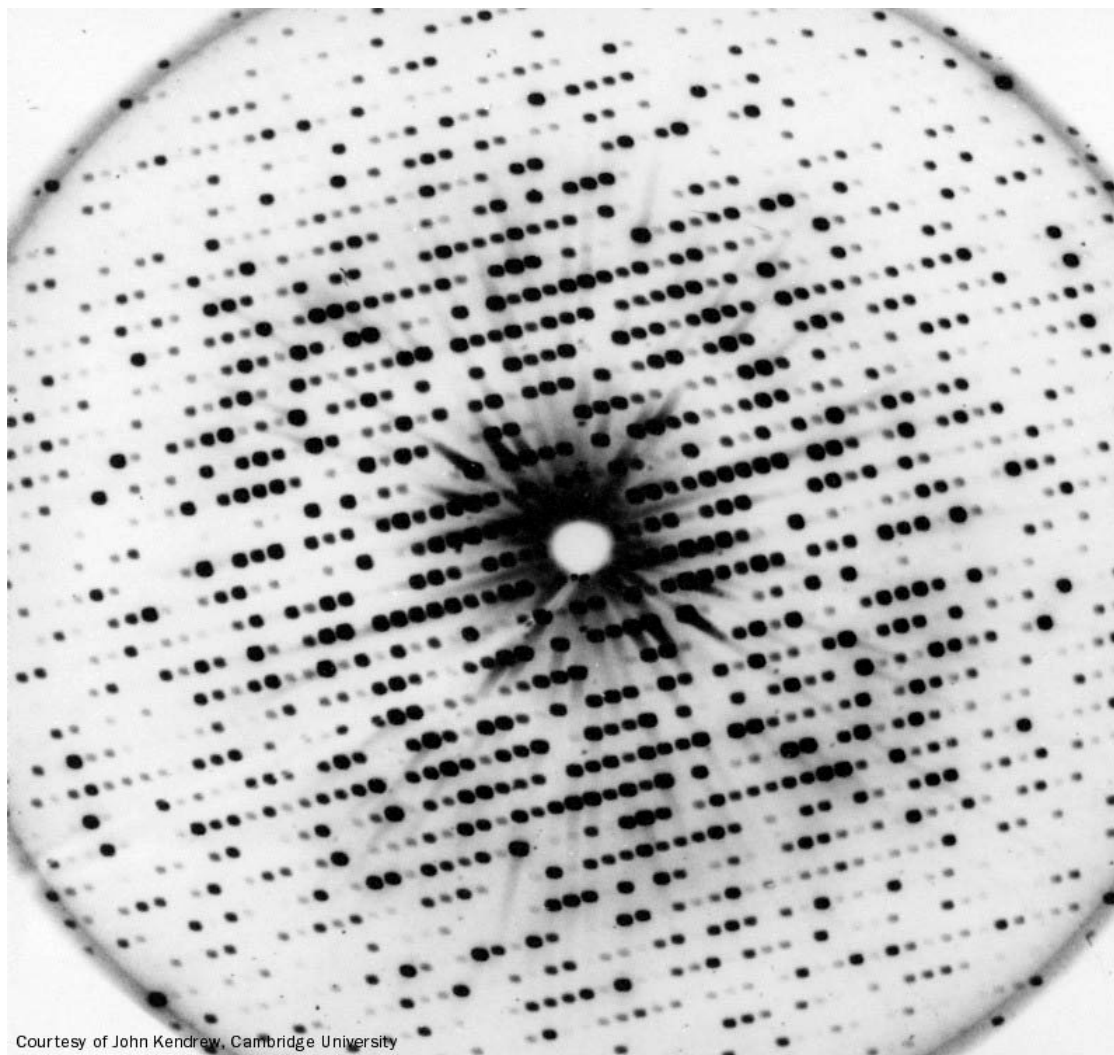
TERCIARNA STRUKTURA GLOBULARNIH PROTEINOV



Pet različnih prikazov terciarne strukture mioglobina.

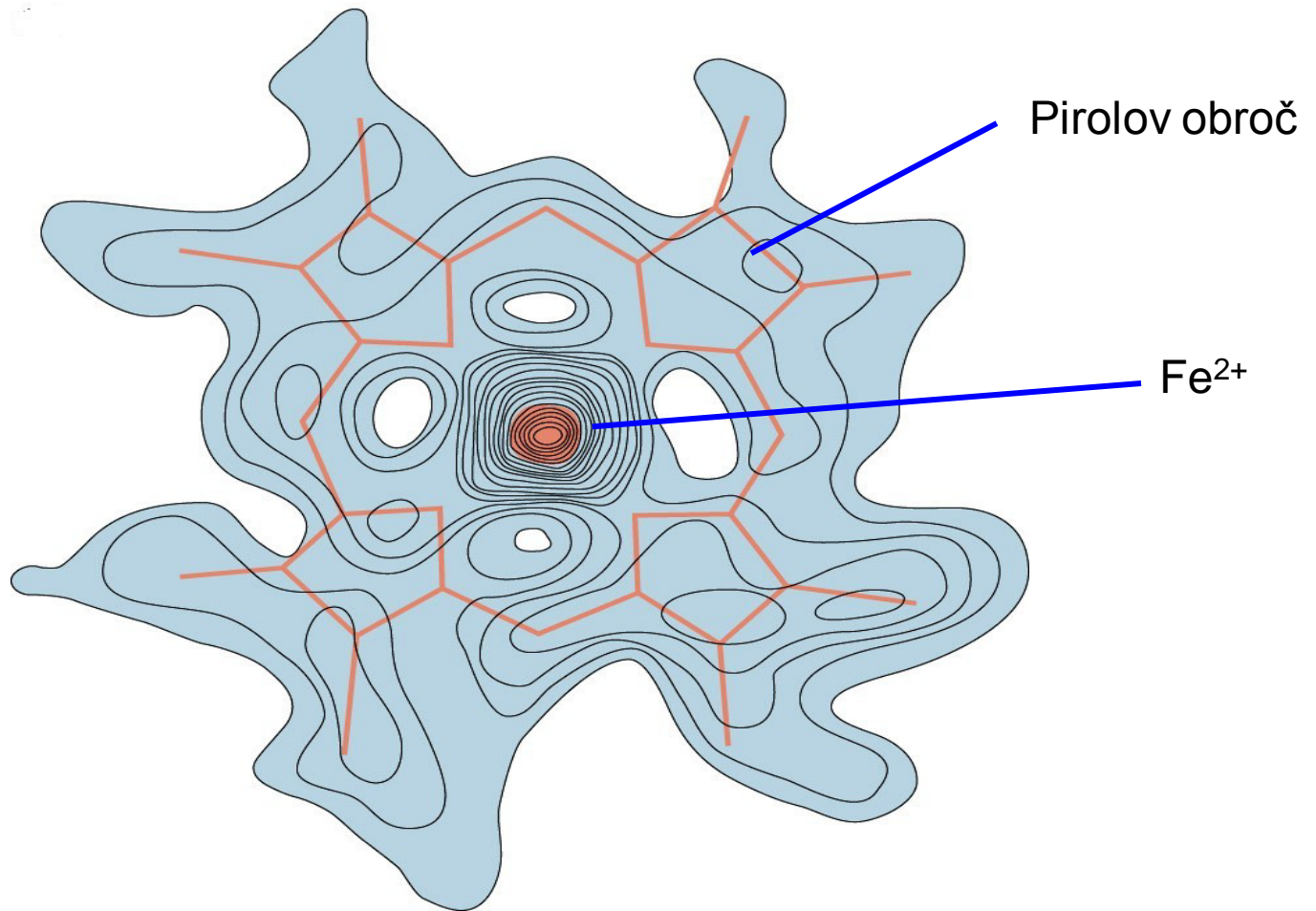
- III. struktura: zvitje v nativno konformacijo
- vse vrste II. strukture so običajno vključene (tudi neurejena struktura)
- III. strukturo vzdržujejo številne šibke interakcije med AK ostanki
- III. struktura je stabilna a fleksibilna
- poznati III. strukturo pomeni poznati položaj vseh atomov v prostoru (3D)

Fotografija lis na filmu po uklonu rentgenskih žarkov skozi kristal Mb.

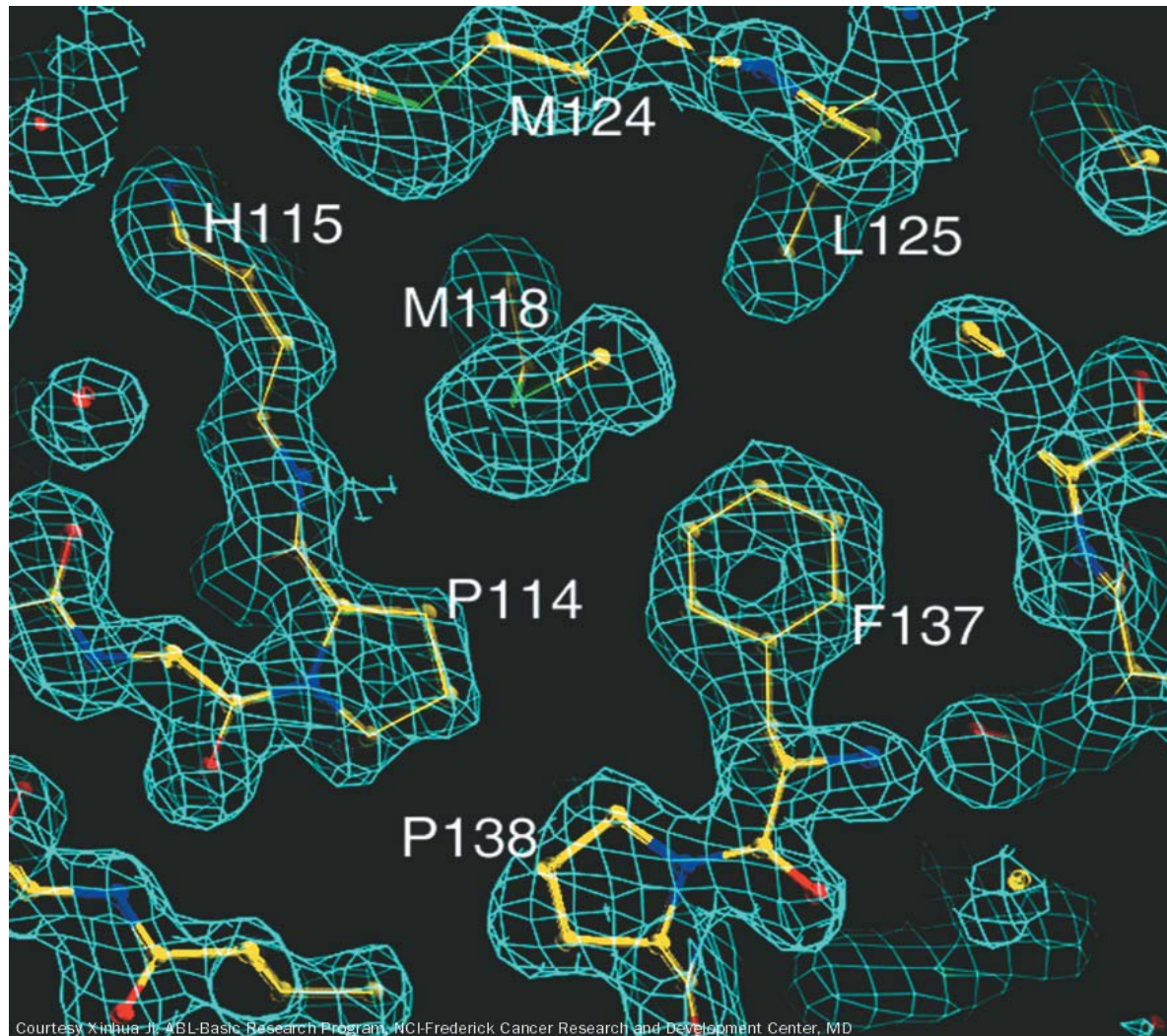


Courtesy of John Kendrew, Cambridge University

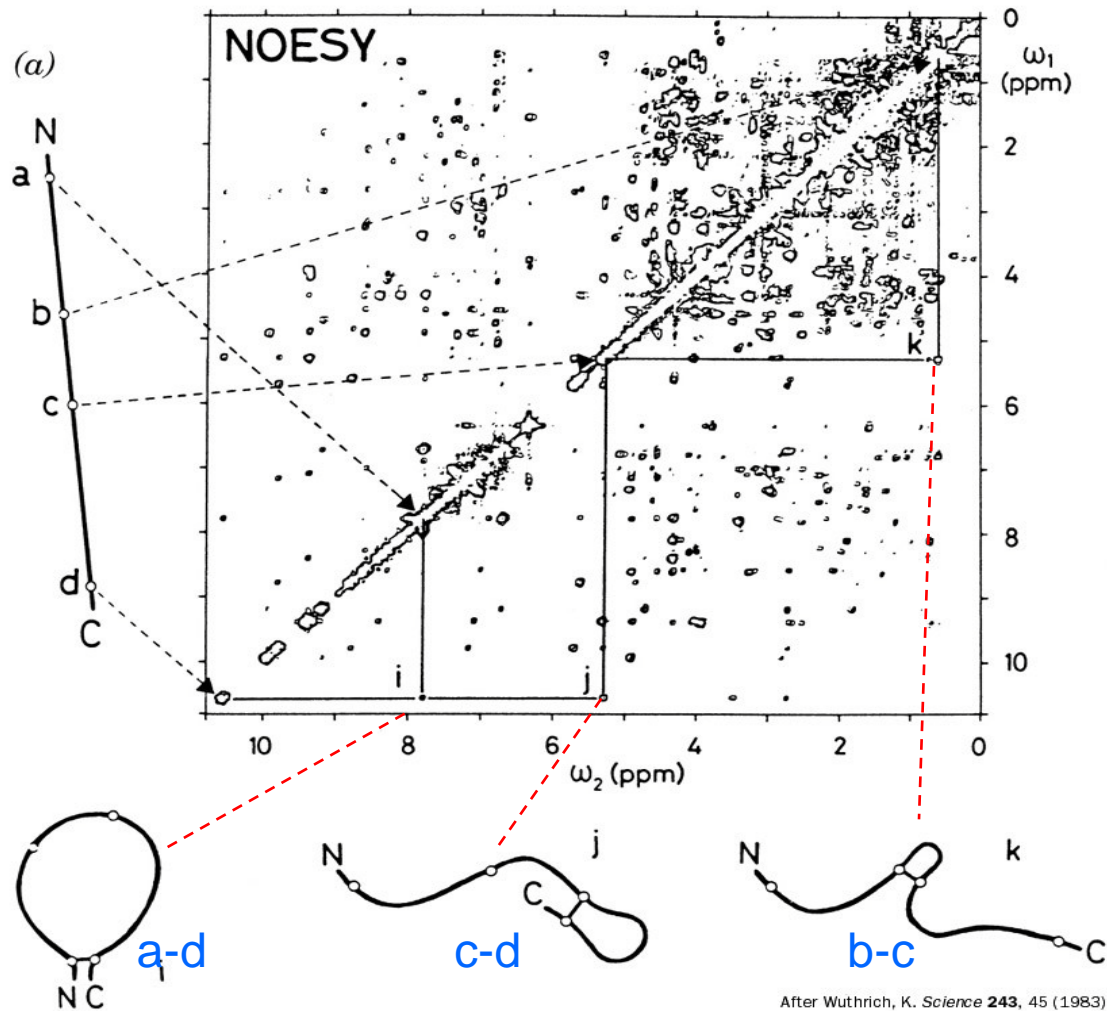
‘Zemljevid’ elektronske gostote hema.



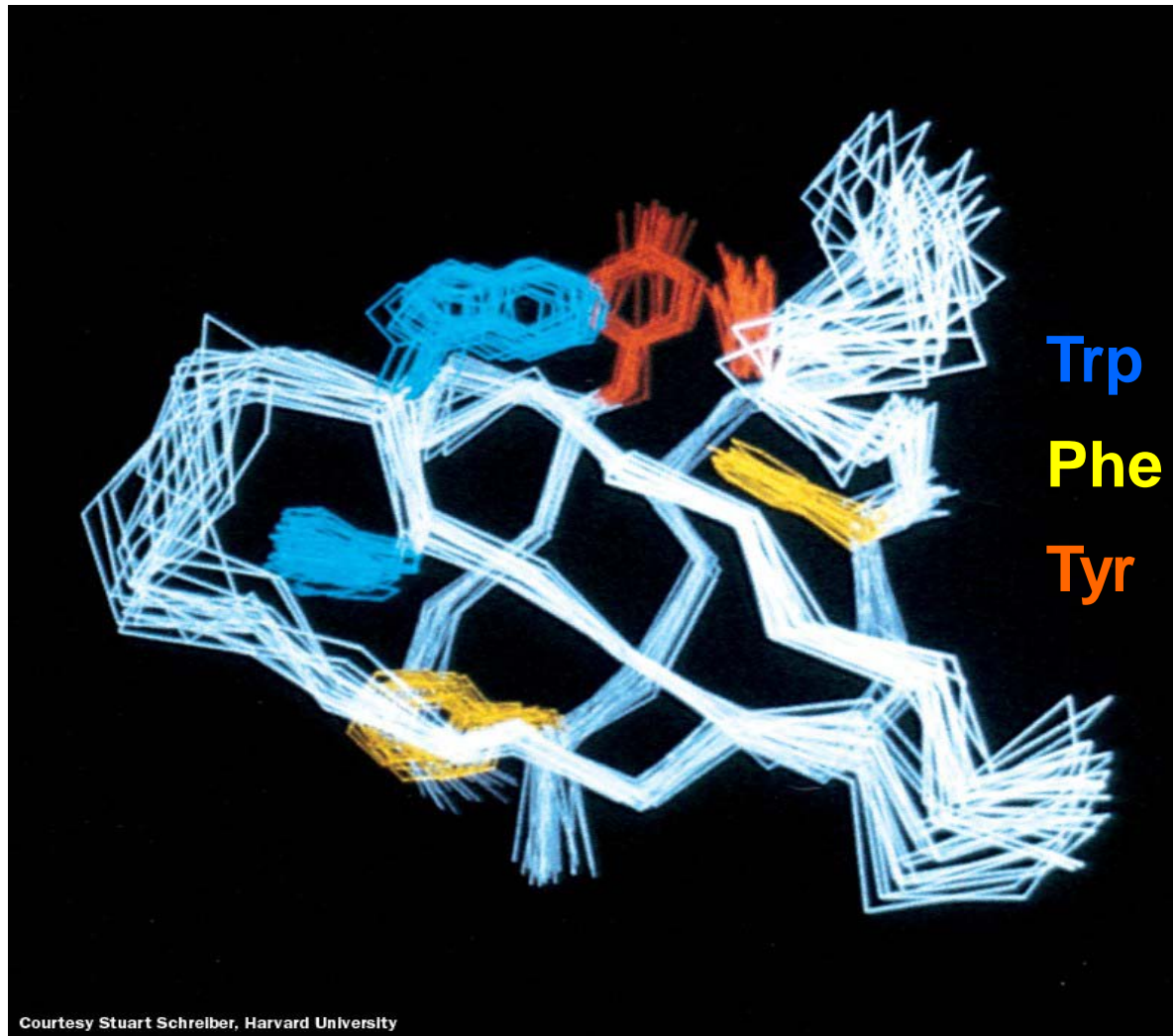
'Zemljevid' elektronske gostote proteina.



Spektri 2D protonske NMR tudi omogočajo razrešitev strukture proteinov (v raztopini!).

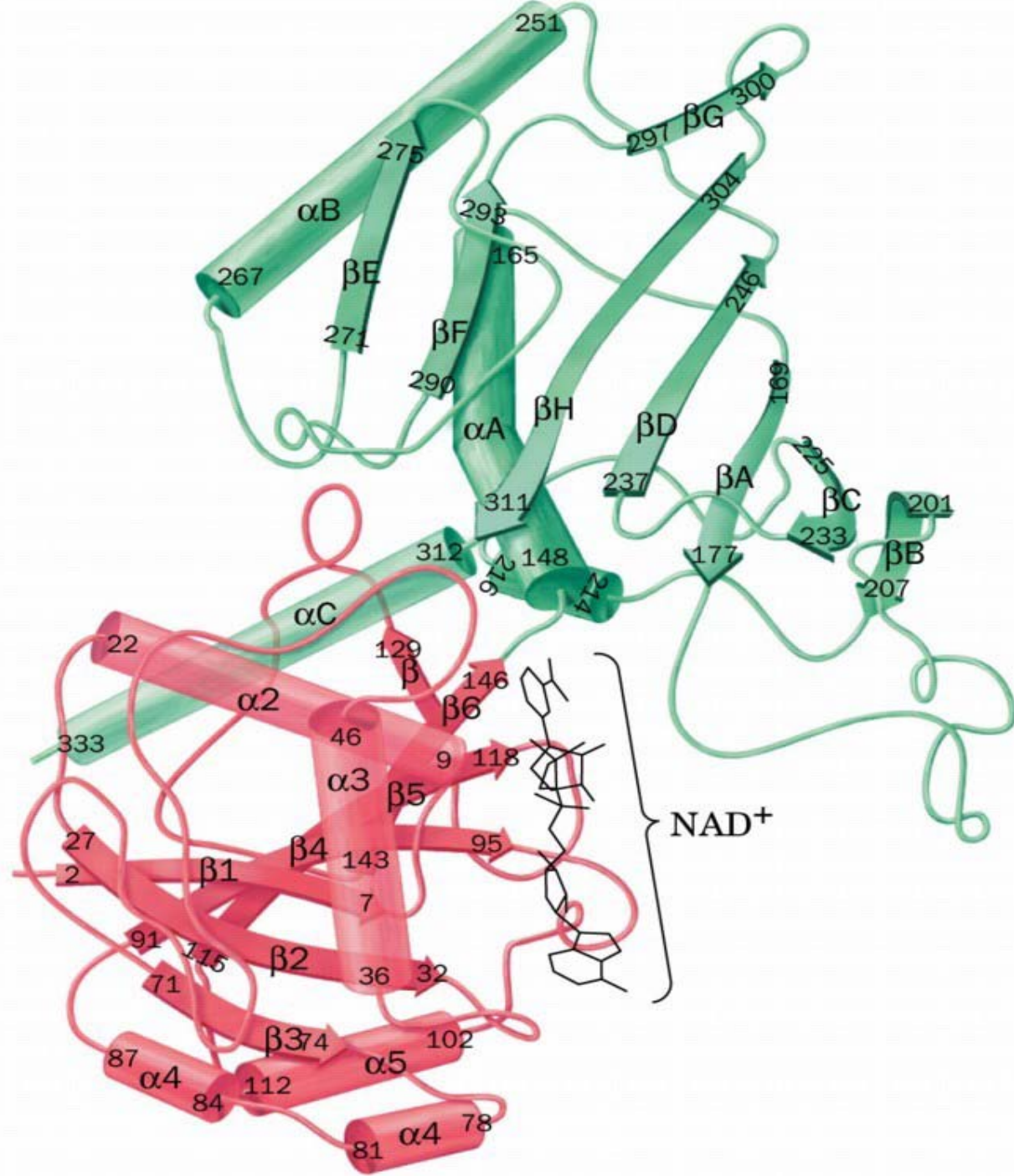


Struktura proteina (**Src protein SH3 domena**) dobljena z 2D
protonsko NMR; vidijo se področja večje fleksibilnosti.

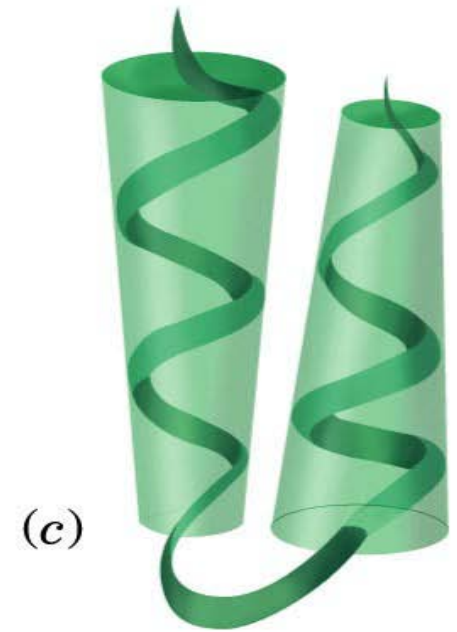
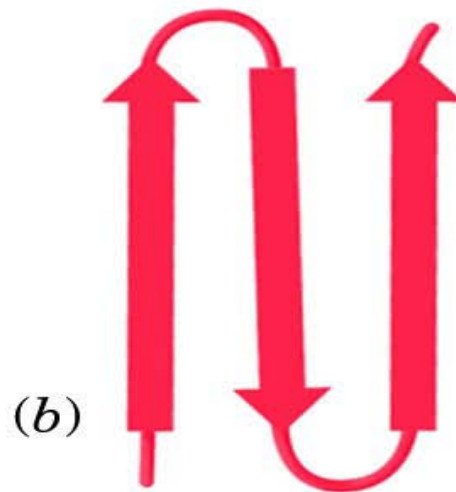
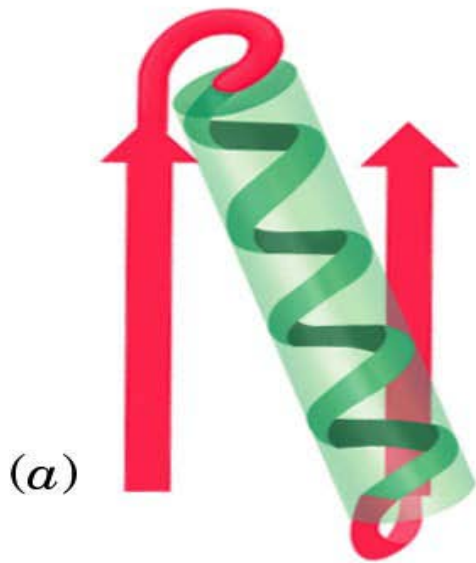


Ena podenota encima gliceraldehid-3-fosfate dehidrogenase iz *Bacillus stearothermophilusa*.

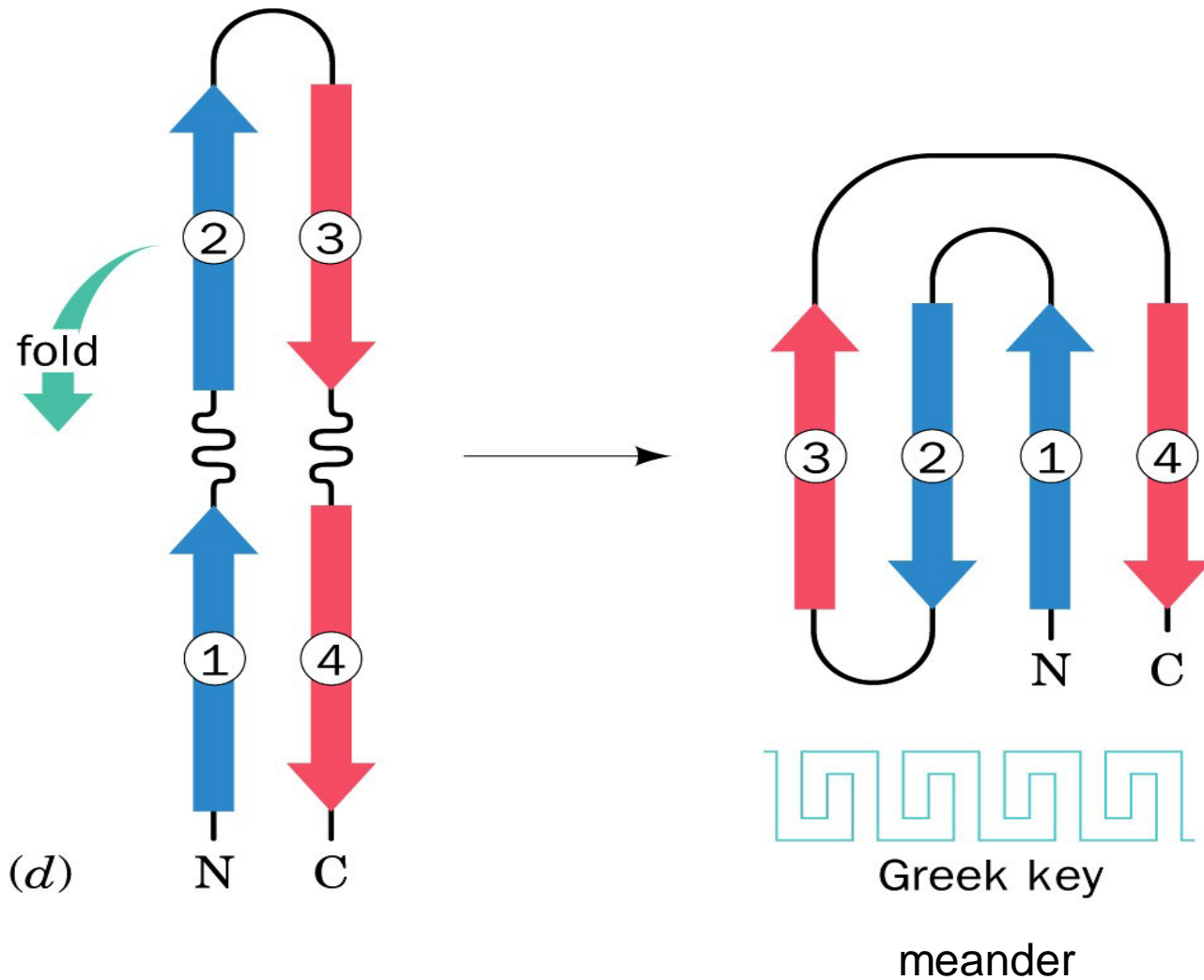
POZOR: Kombinacija različnih II. struktur. Jasno se vidita dve domeni (ena rdeče in druga zelene barve).



Različne supersekundarne strukture (motivi).

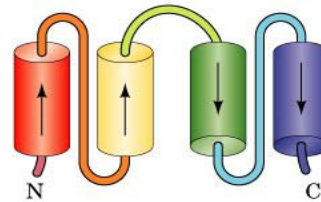
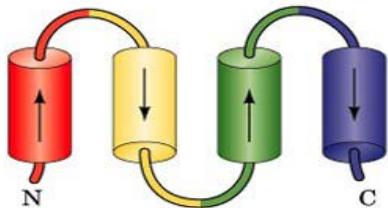


Različne supersekundarne strukture (motivi).

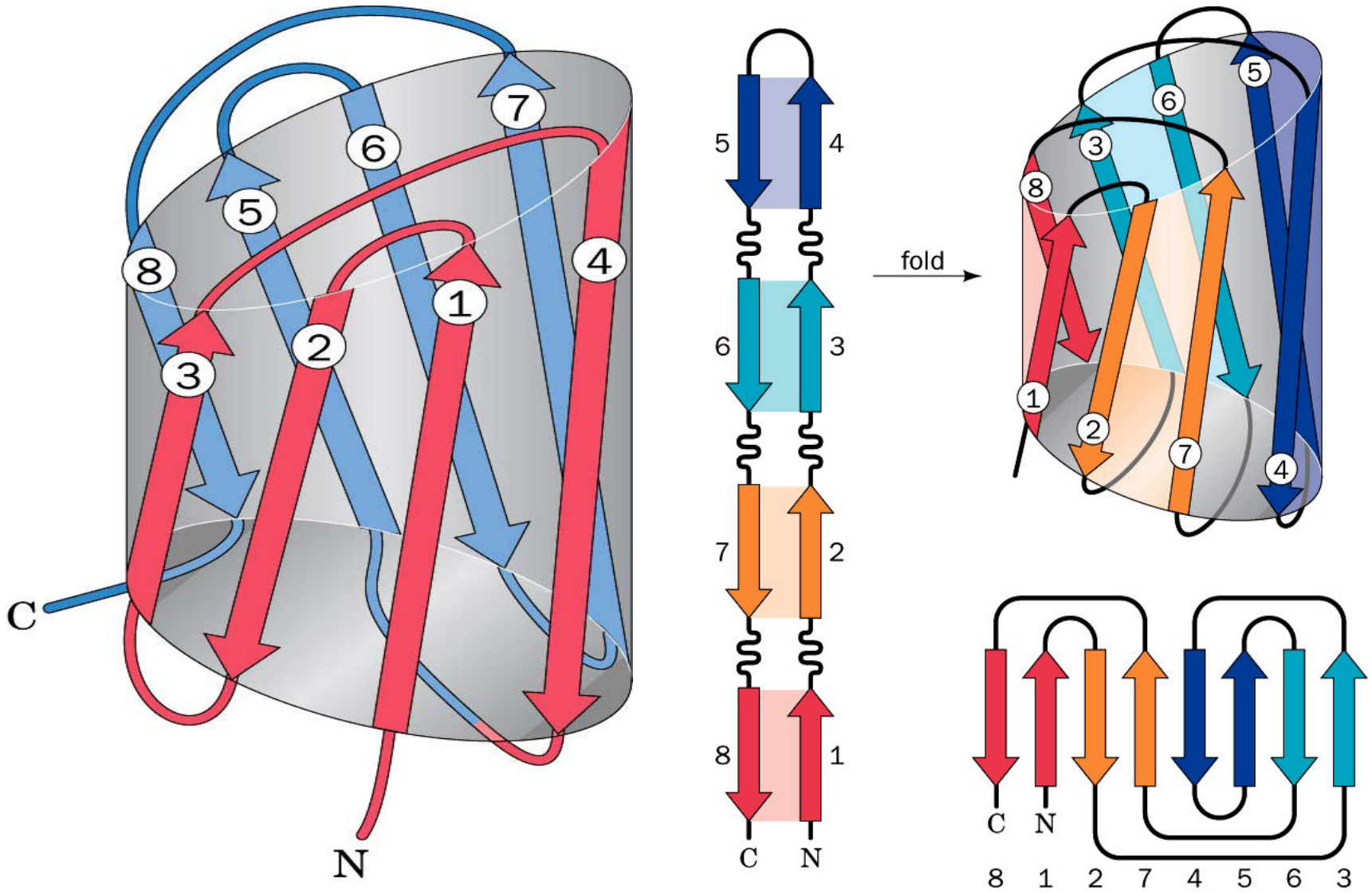


4-helix bundle proteins:

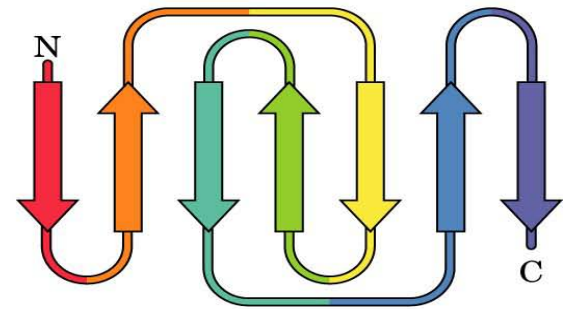
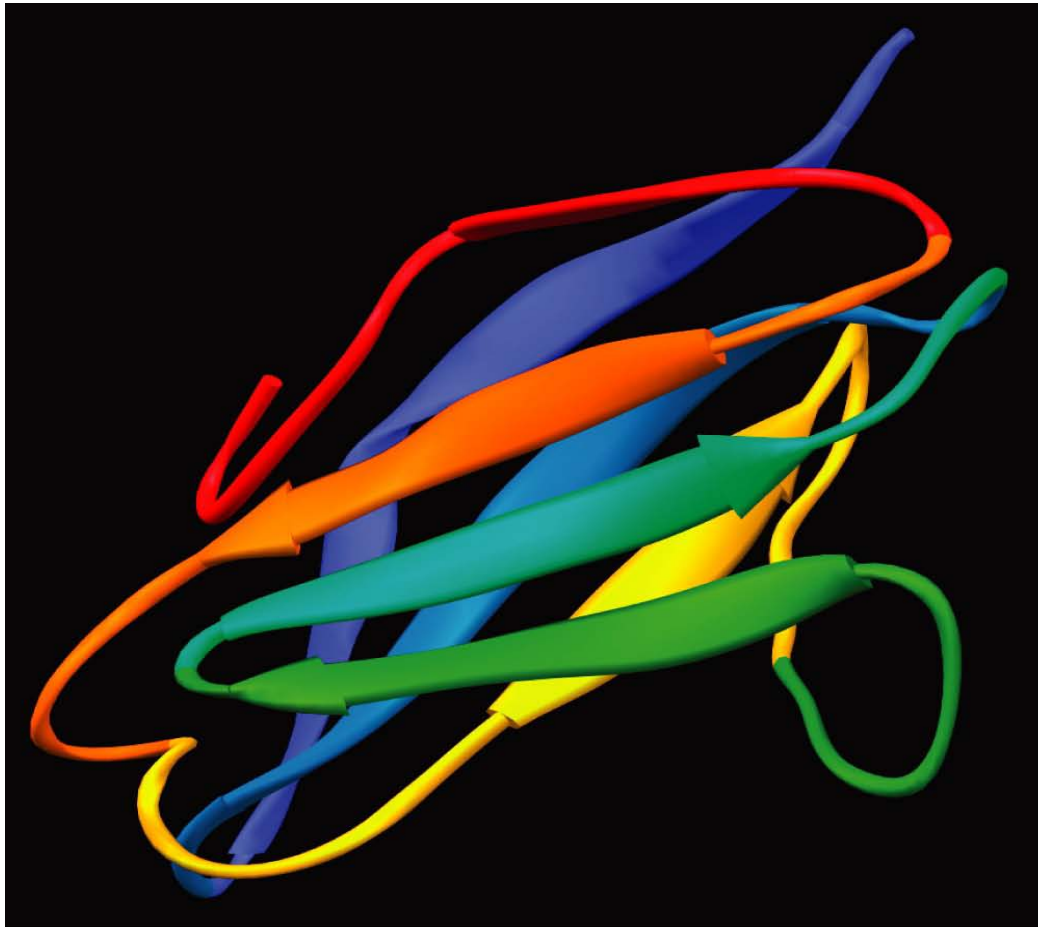
(a) *E. coli* cytochrome b_{562} and (b) Human growth hormone.



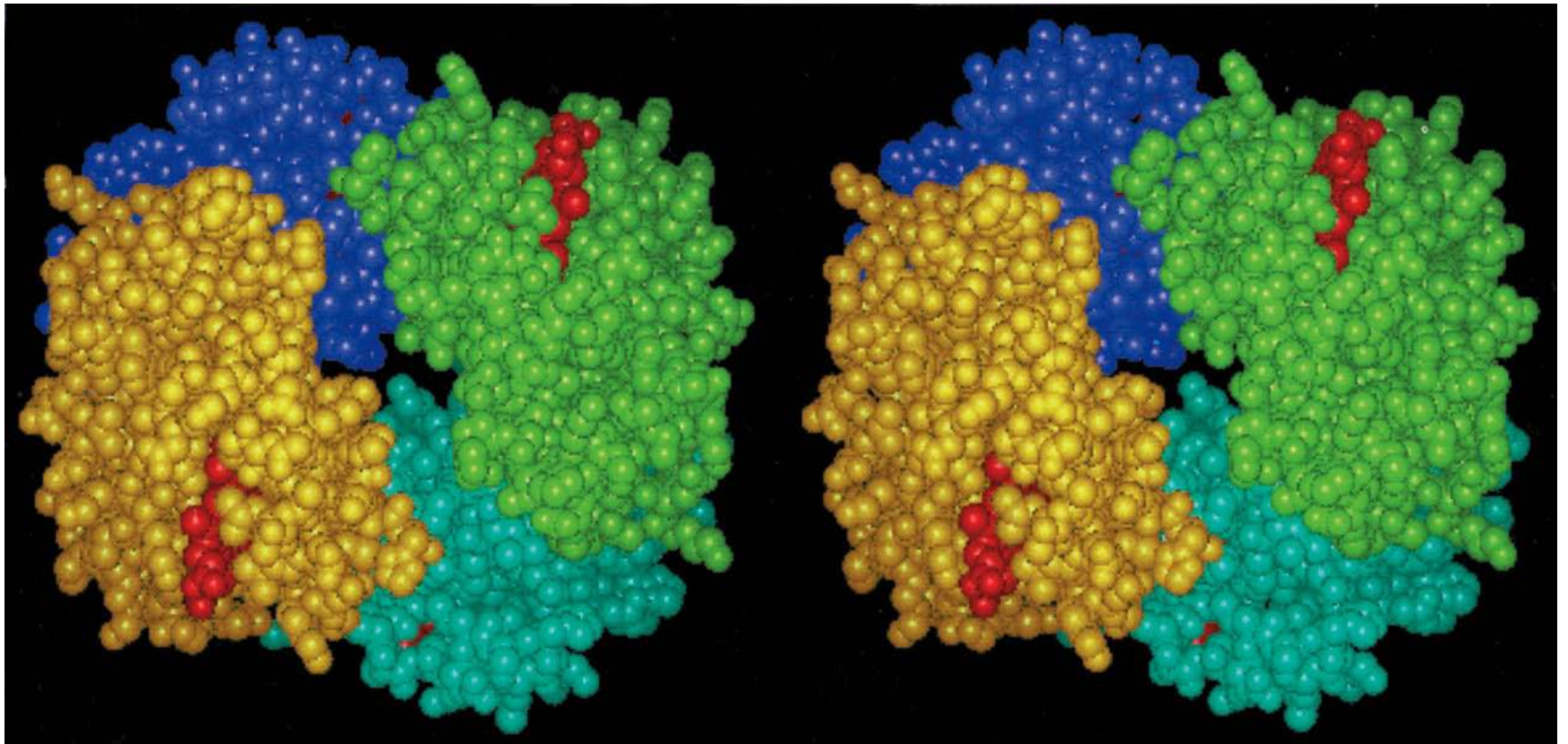
Struktura C-terminalne domene γ - β kristalina. (a) Organizacija meandrov (Greek key) v β sodček.



Imunoglobulinsko zvitje.

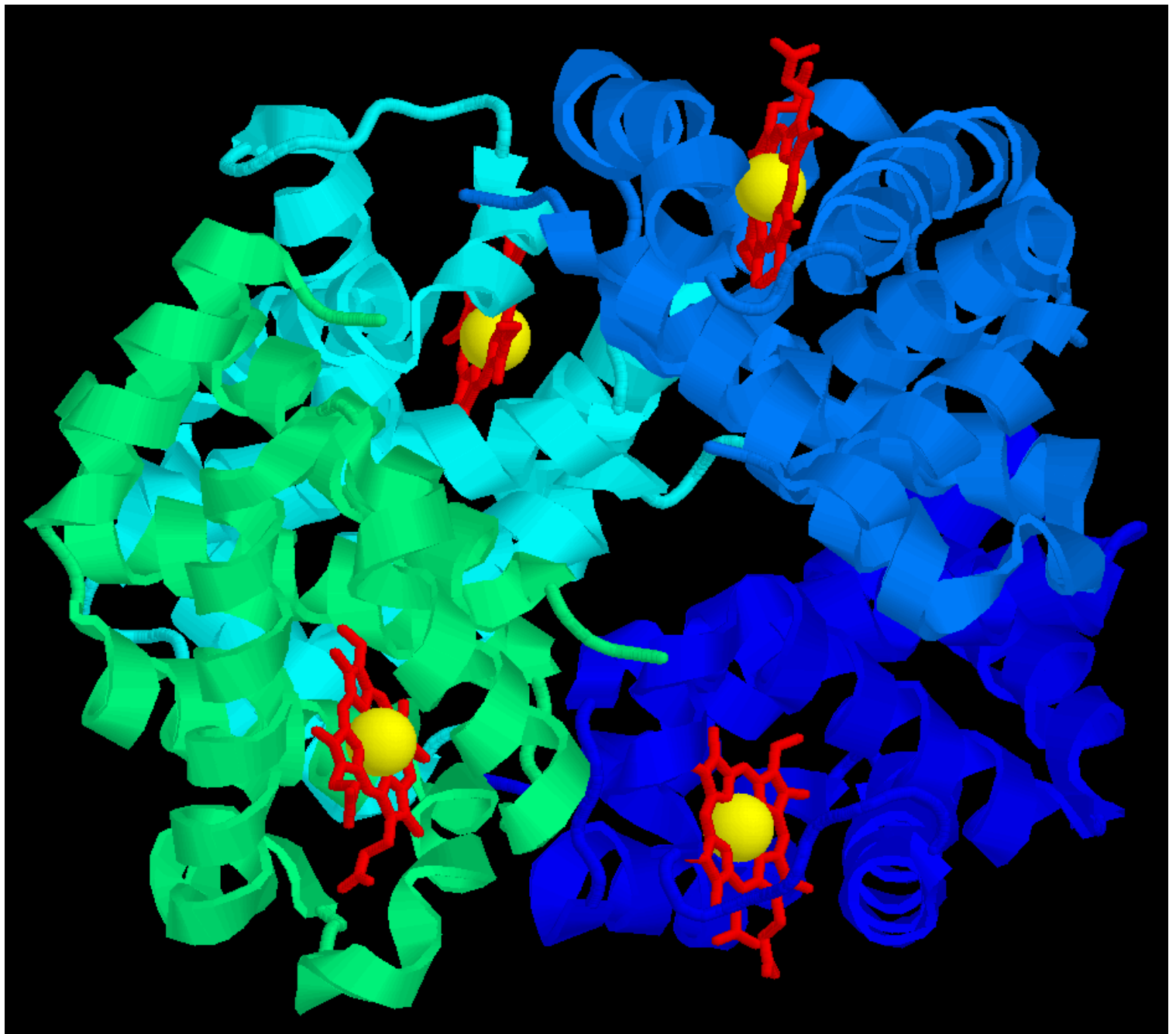


Kvartarna struktura hemoglobina.



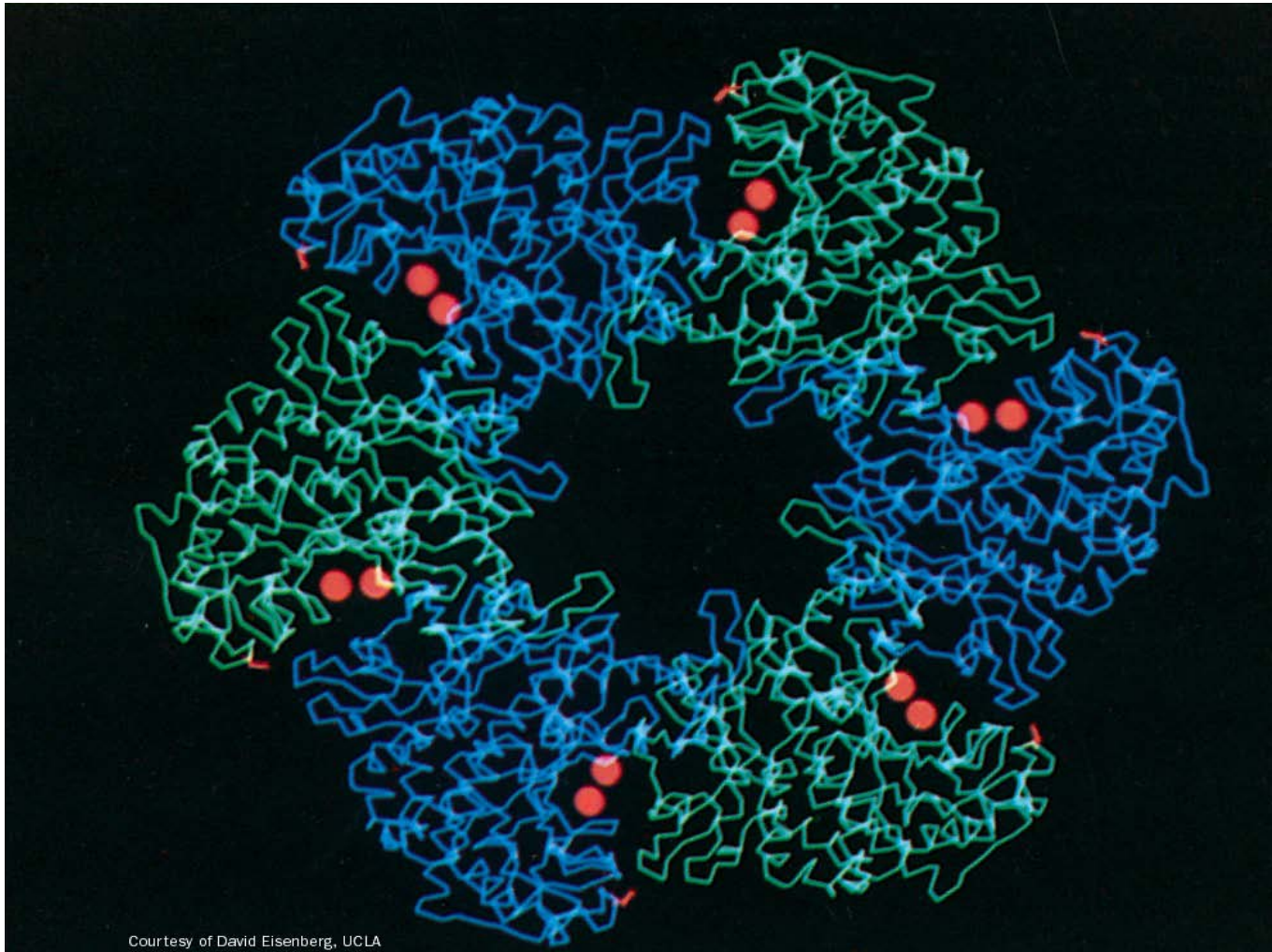
Hb

(2D5Z.pdb)



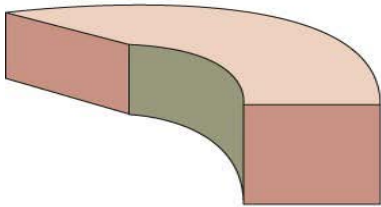
POKAŽI!

Glutamin-sintetaza (*Salmonella typhimurium*).

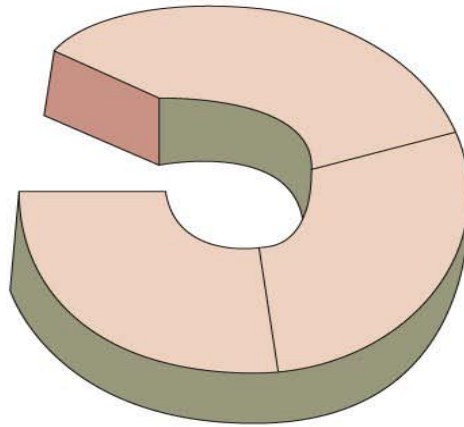


Kvartarna struktura iz ene same podenote.

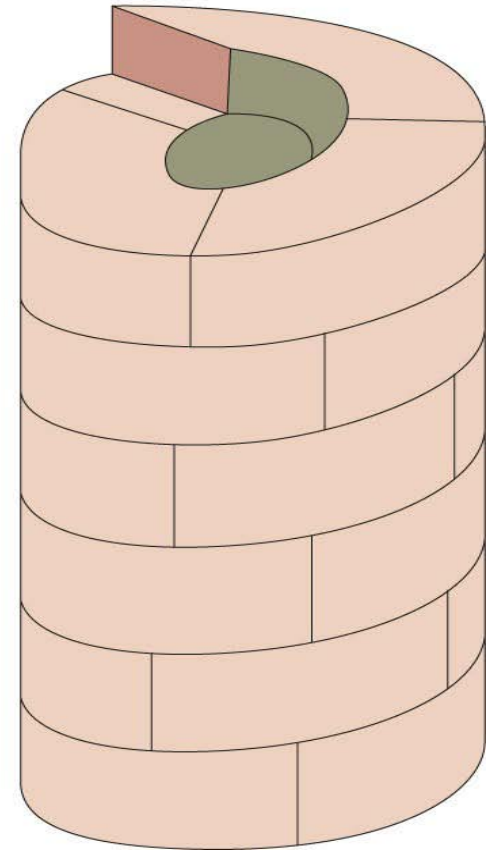
Subunit



Helix segment



Helix

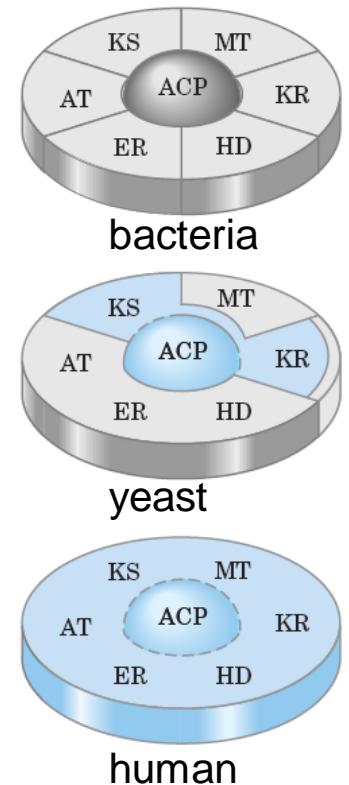
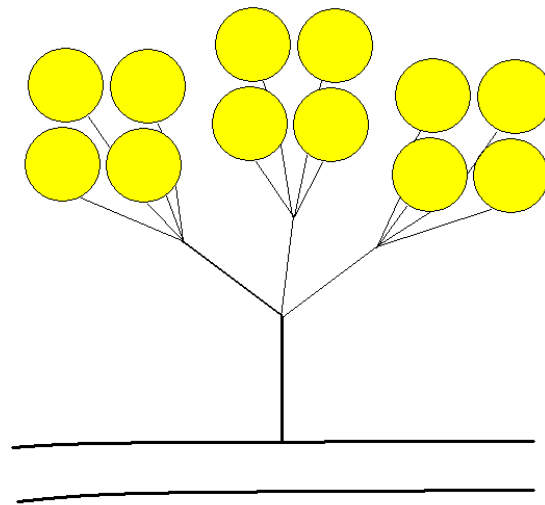
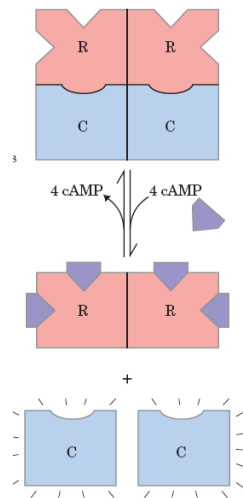
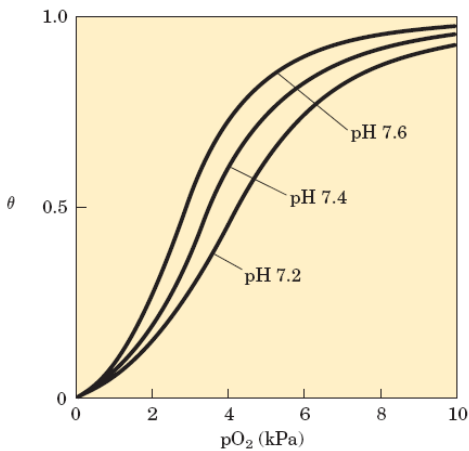


virus tobaka, aktinska vlakna, itd

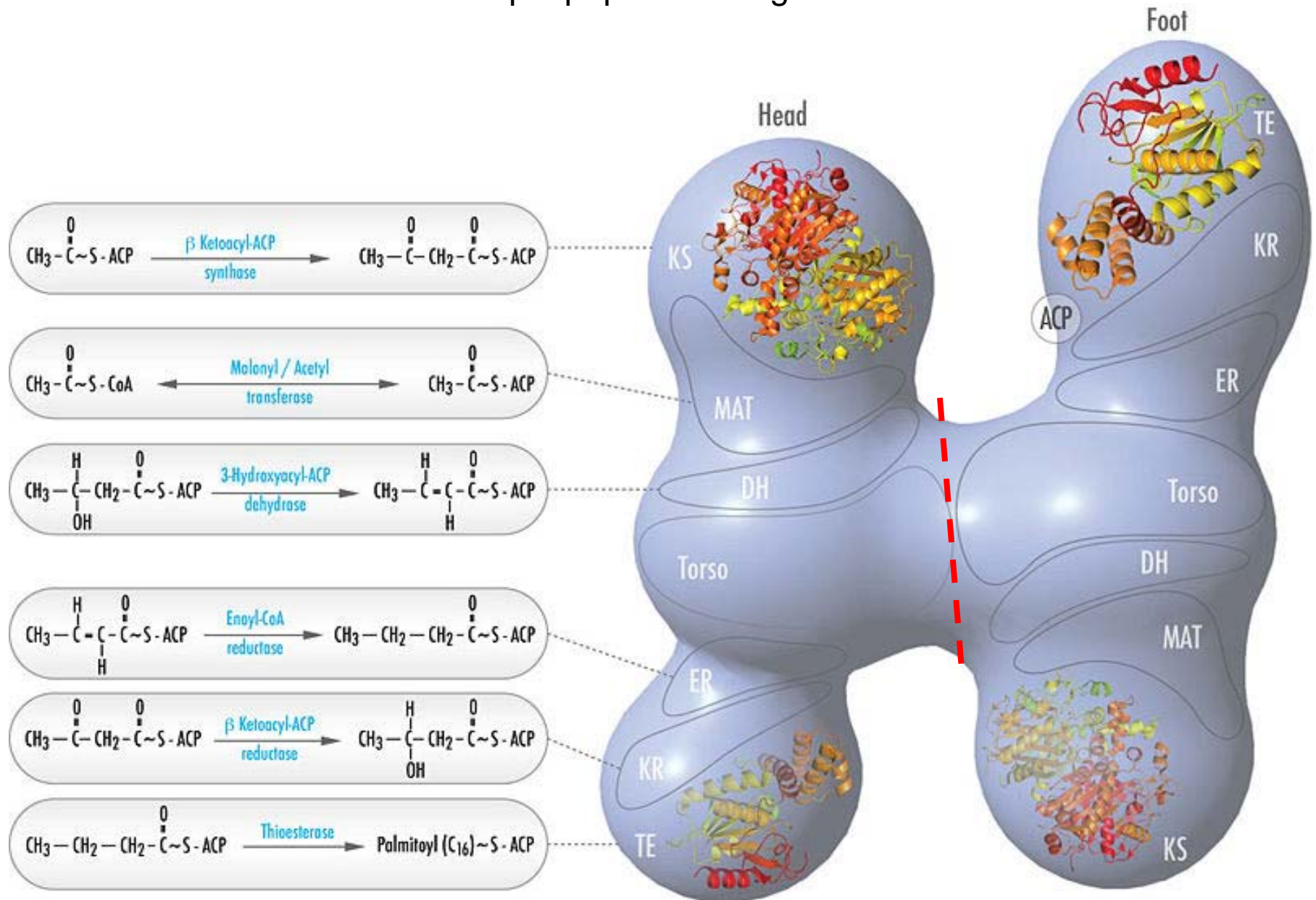
ZAKAJ KVARTARNA STRUKTURA?

- VARNOST (Hb: $2\alpha+2\beta$, 2 različna gena)
- ALOSTERIČNI EFEKTI (regulacija – Hb, alosterični encimi)
- POVEČANA & LOKALIZIRANA KONCENTRACIJA AKTIVNIH MEST (AChE)
- MULTIFUNKCIONALNI PROTEINI:

maščobnokislinska sintetaza: 7 aktivnih mest z različnimi funkcijami:

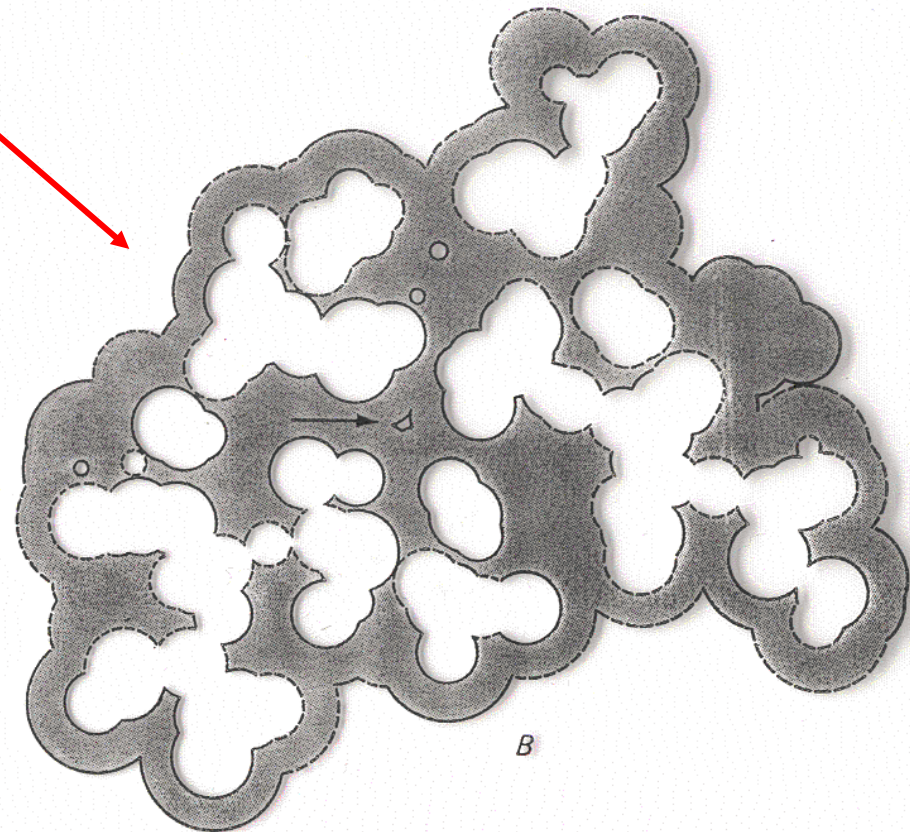
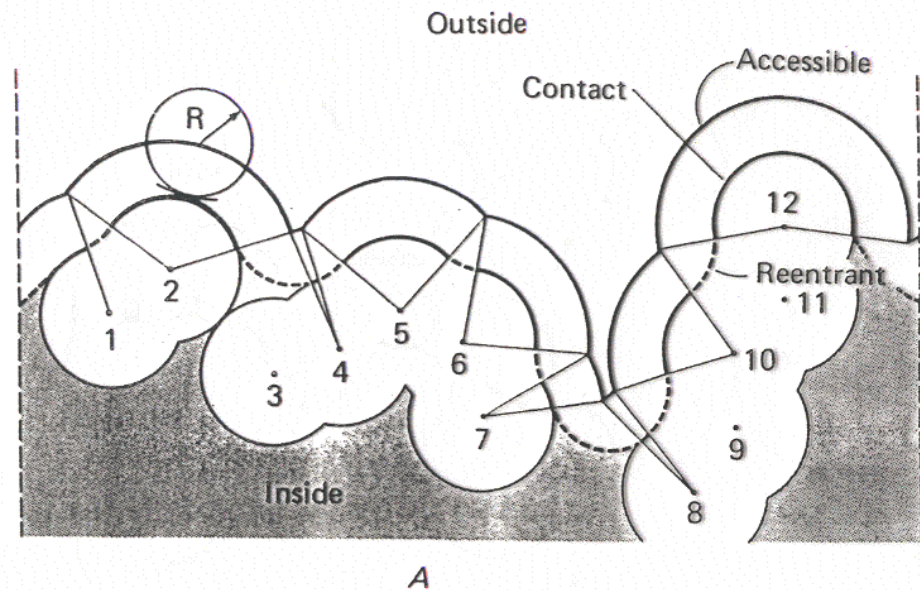


Multifunkcionalni proteini: maščobnokislinska sintetaza sesalcev: dimer dveh polipeptidnih verig

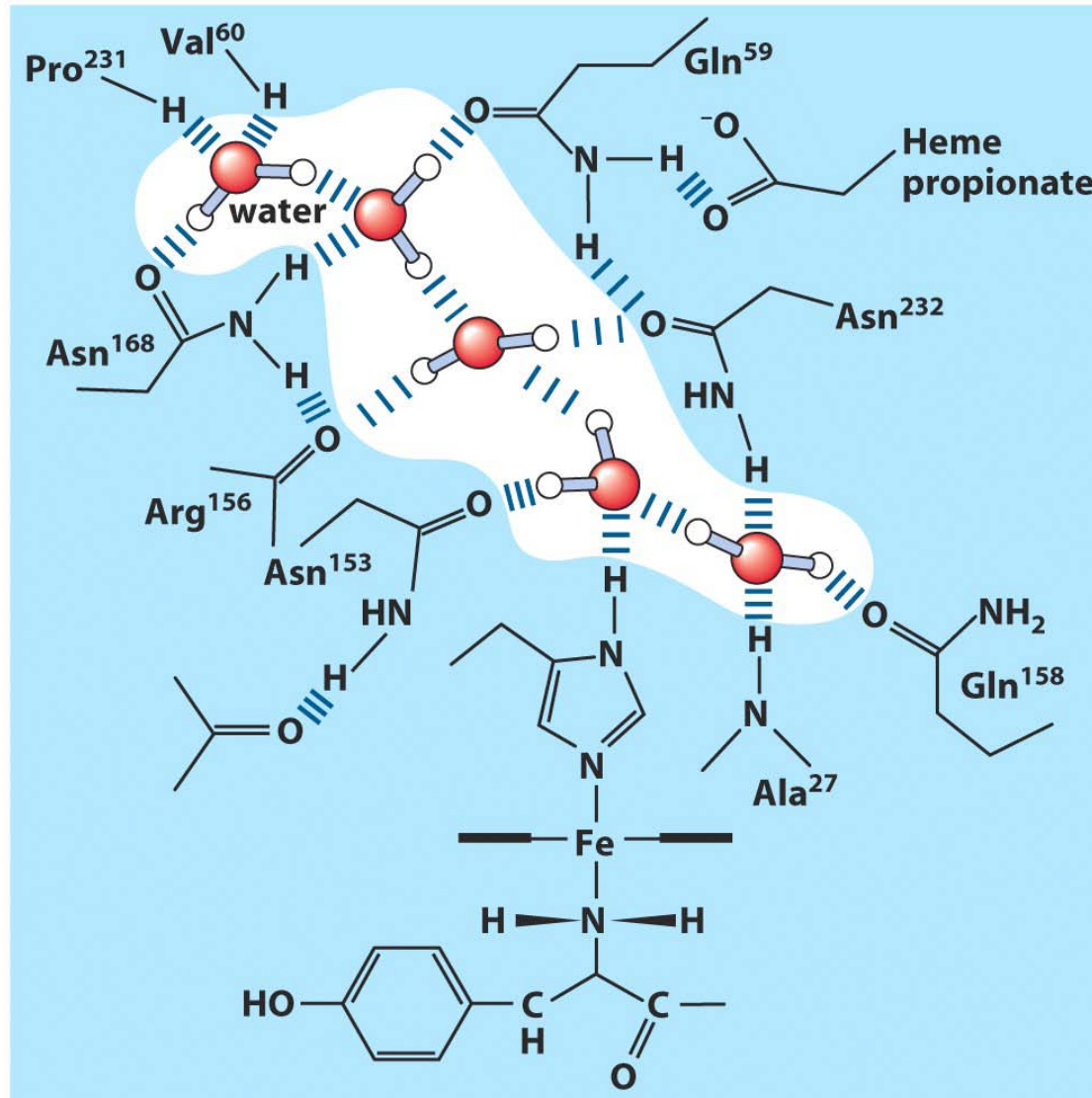


Nativna konformacija je:

- gosta
- kompaktna
- lahko ima vrzeli (luknje)
- voda je njen integralni del



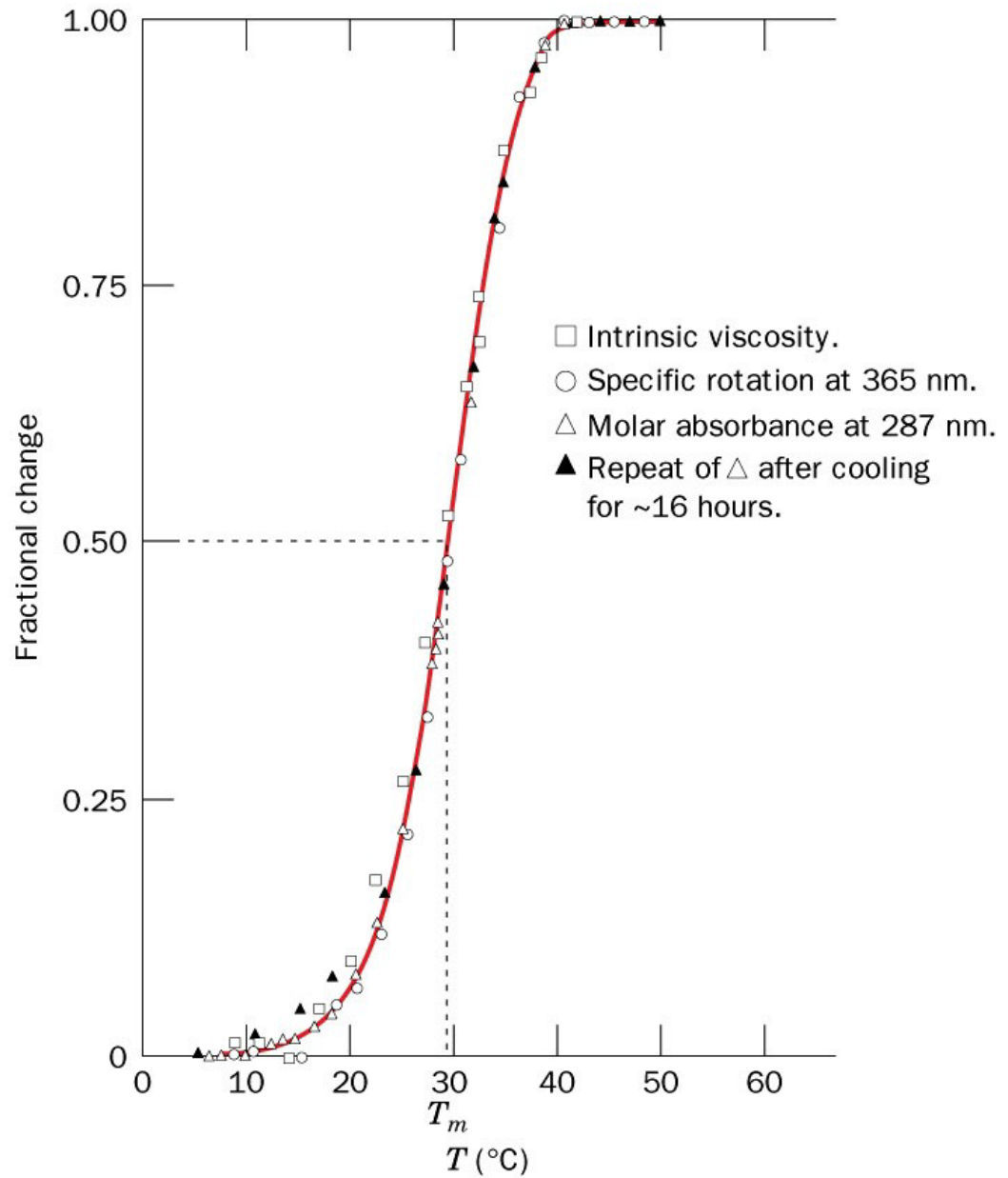
Voda je integralni del proteinske strukture in pomembno prispeva k njeni stabilnosti.



Proteine lahko denaturiramo – ključ do poznavanja zvijanja

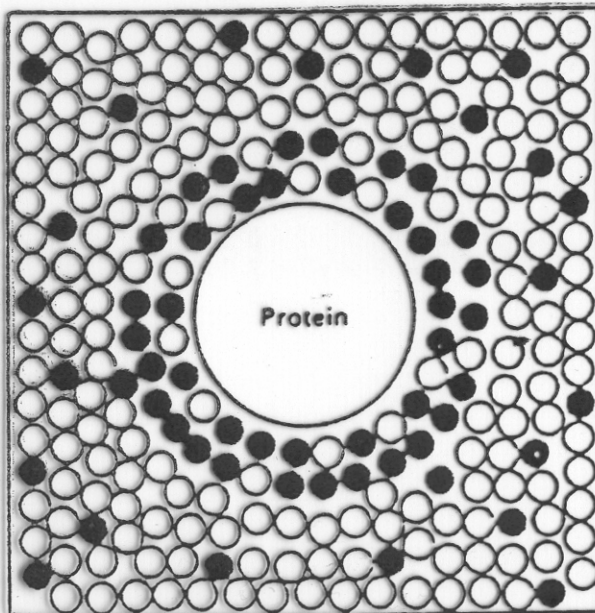
Denaturanti, pH, T in visok P porušijo strukturo. Proces je lahko reverzibilen.

POKAŽI DIAGRAM (NE)STABILNOSTI!

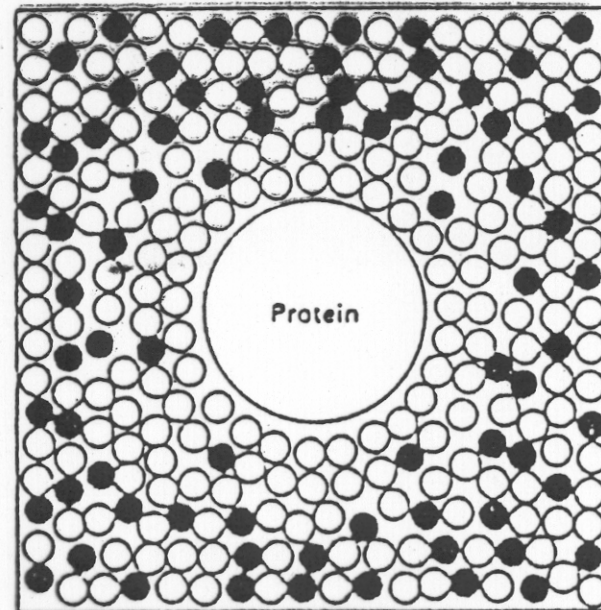


Zakaj denaturacija:

- toplota (visoka temperatura): neposreden vnos energije
- ekstremni pH: ionizacija le v razvitem stanju (premik ravnotežja); elektrostatski odboji na površini proteina; porušenje ionskih interakcij.
- denaturanti (urea, gvanidinijev klorid): preferenčna vezava (majhen vpliv na H-vezi!)

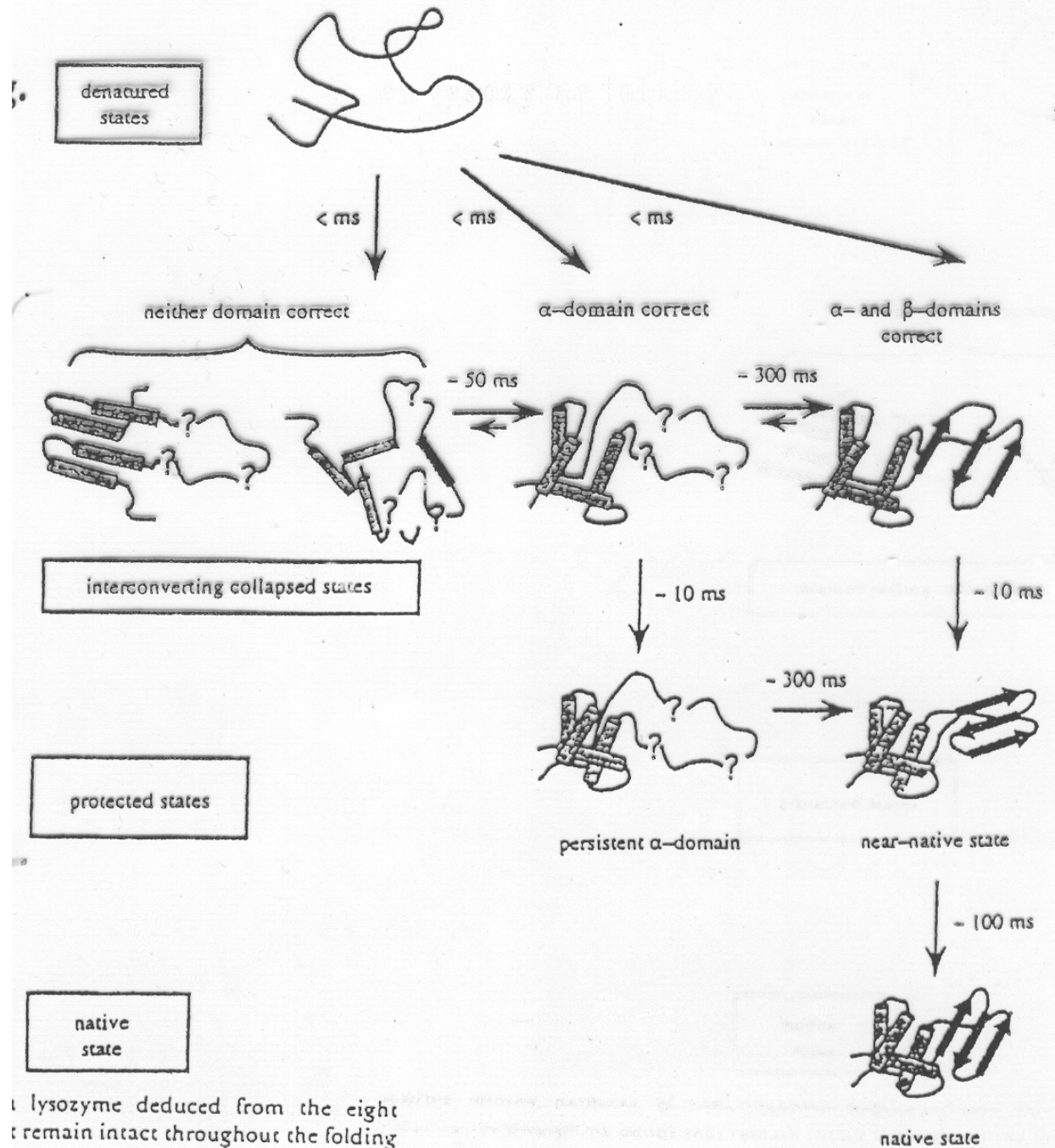
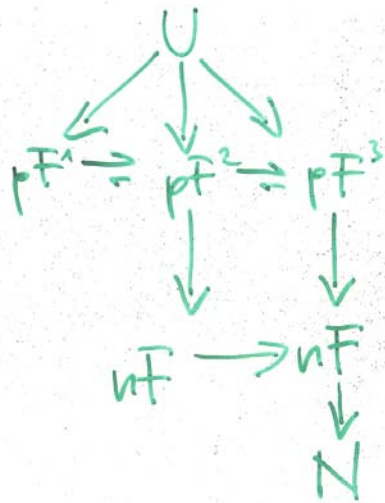


Preferential binding



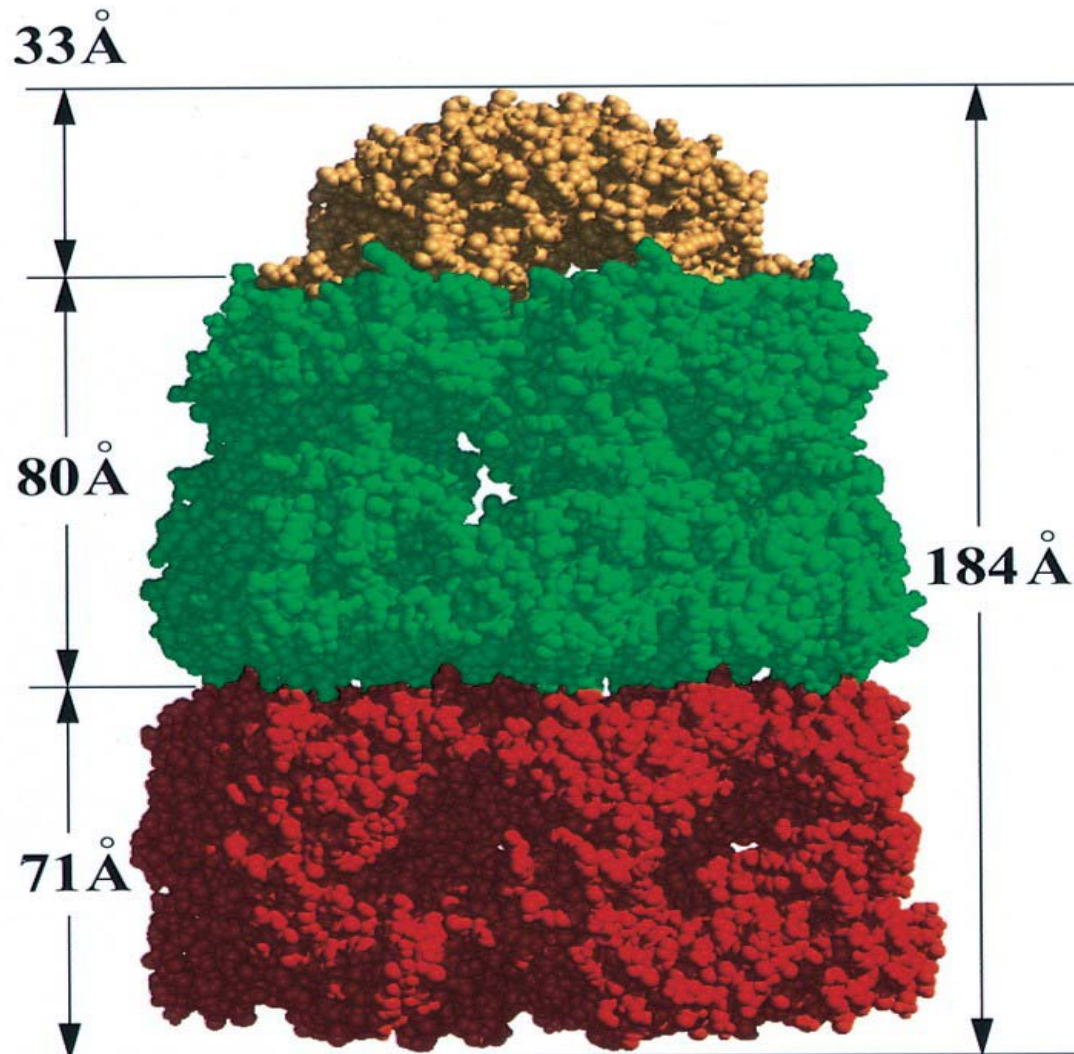
Preferential hydration

Poti zvijanja lizozima

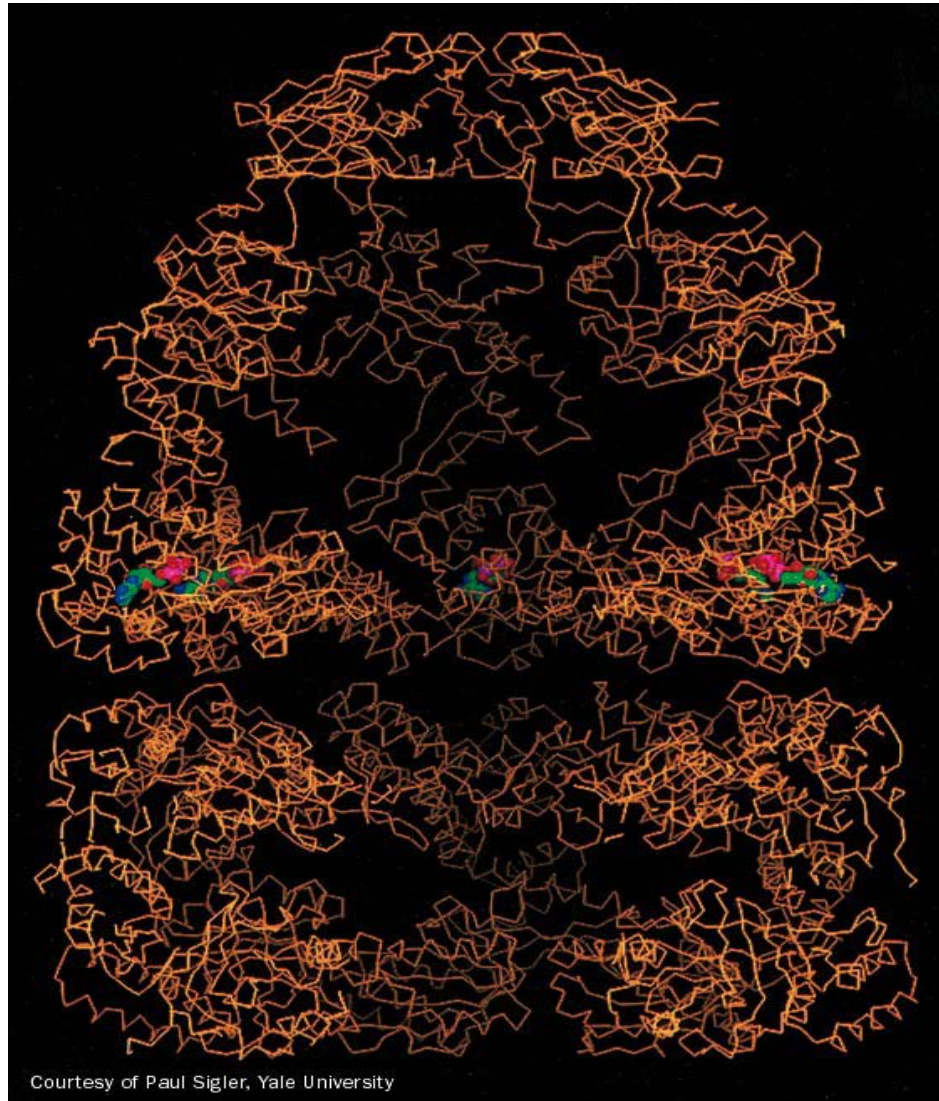


lysozyme deduced from the eight
remain intact throughout the folding

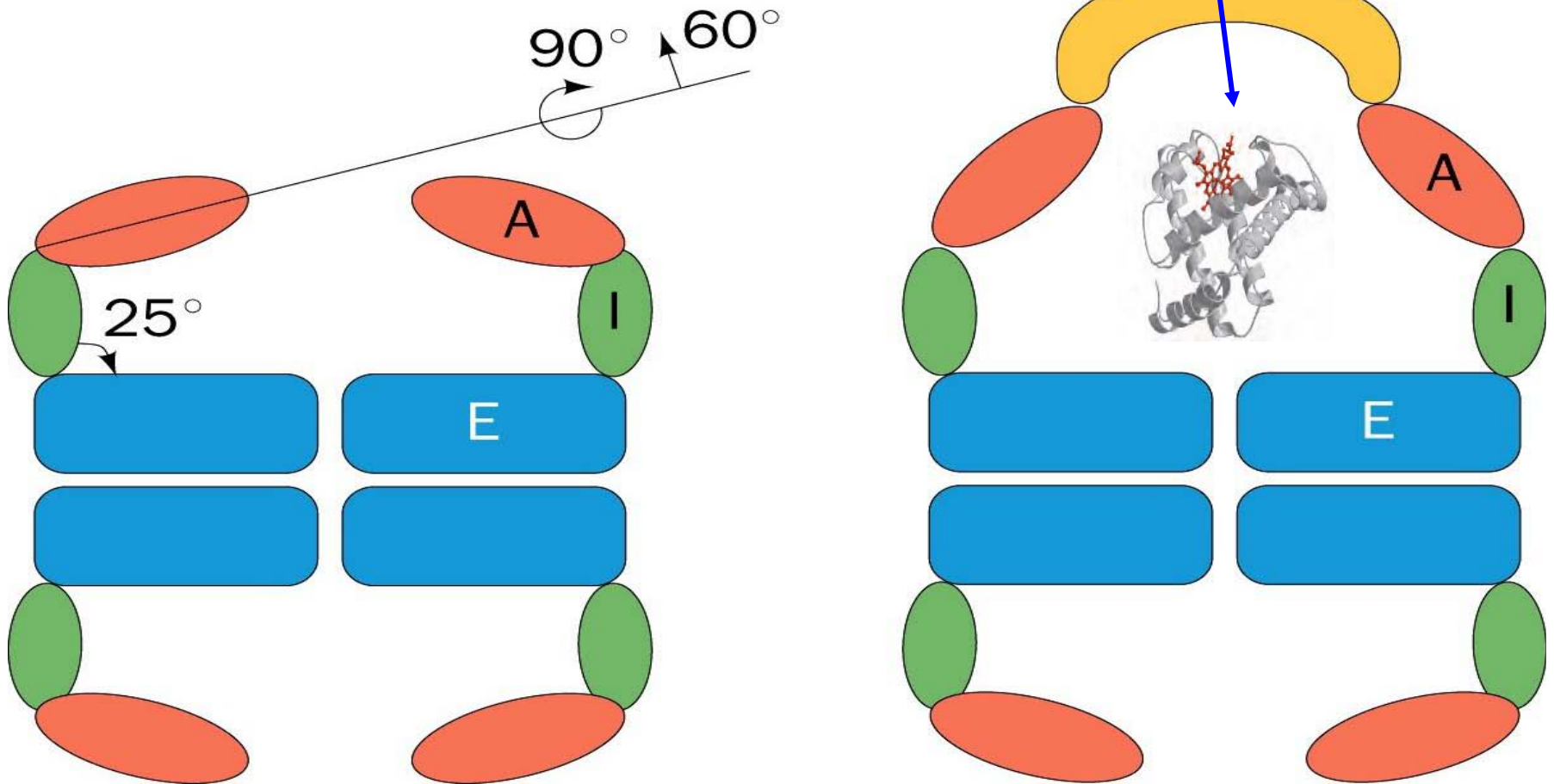
Šaperoni pomagajo proteinom pri zvijanju in skušajo ponovno zviti razvite proteine.



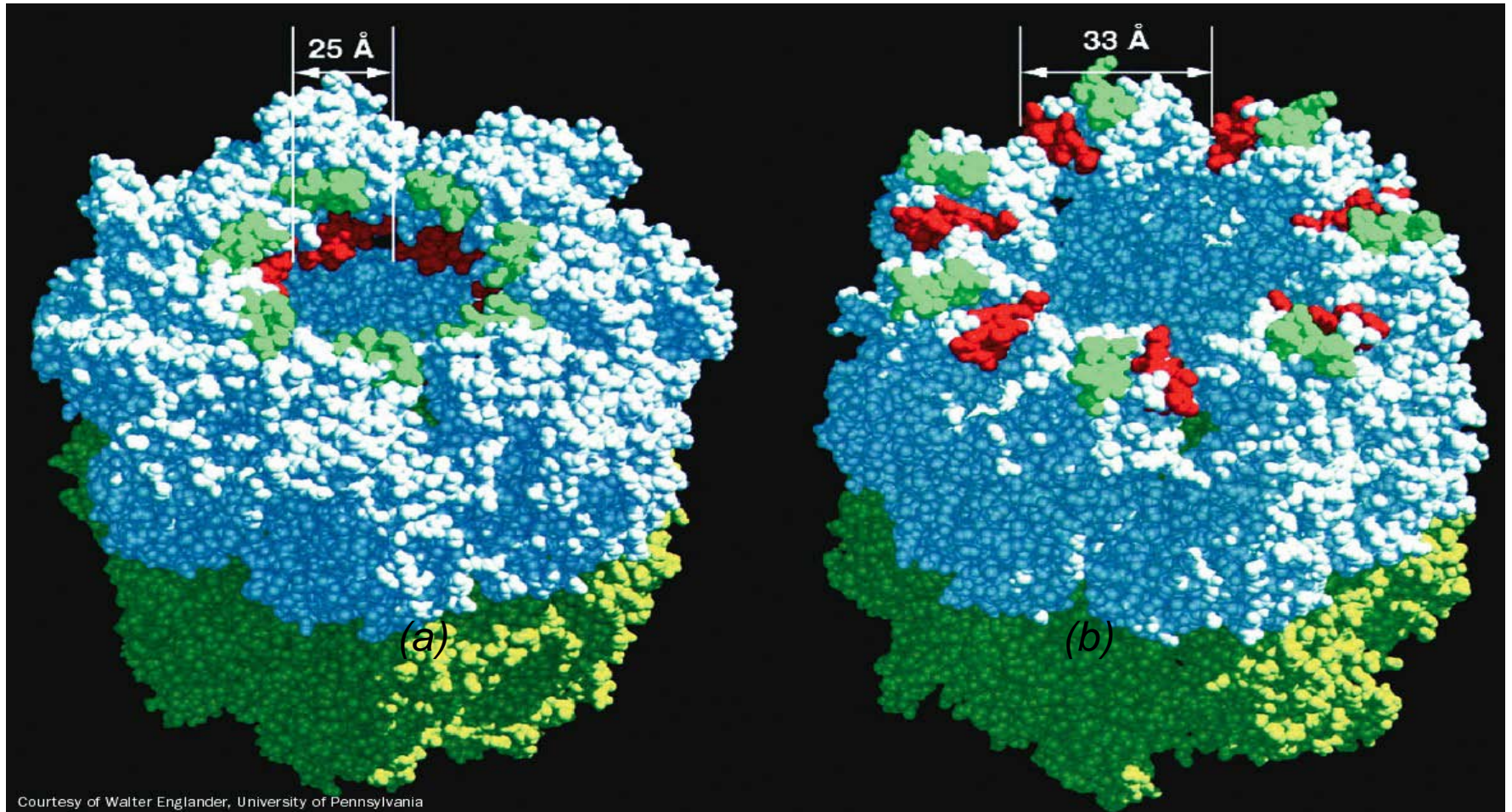
Notranjost šaperona je specifično okolje, ugodno za zvitje proteinske verige.



Premikanje domen v plašču šaperona pomaga pri zvijanju proteinske verige, ki je v notranjosti. Konformacijske spremembe omogoča hidroliza ATP.



Premikanje heliksov (rdeče), ki vežejo protein v notranjosti šaperona.

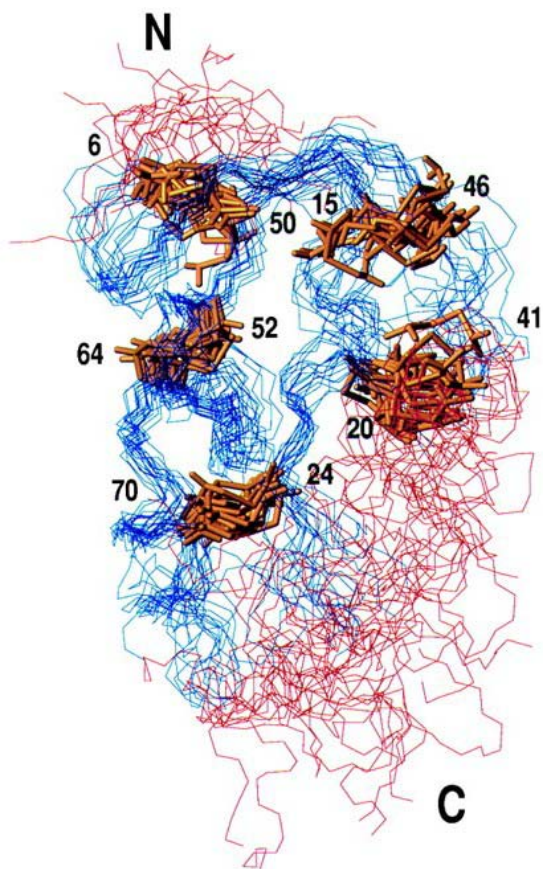


Fleksibilnost molekule Mb – hrbtenica polipeptida (modro), hem (rumeno) in proksimalni His (rdeče).

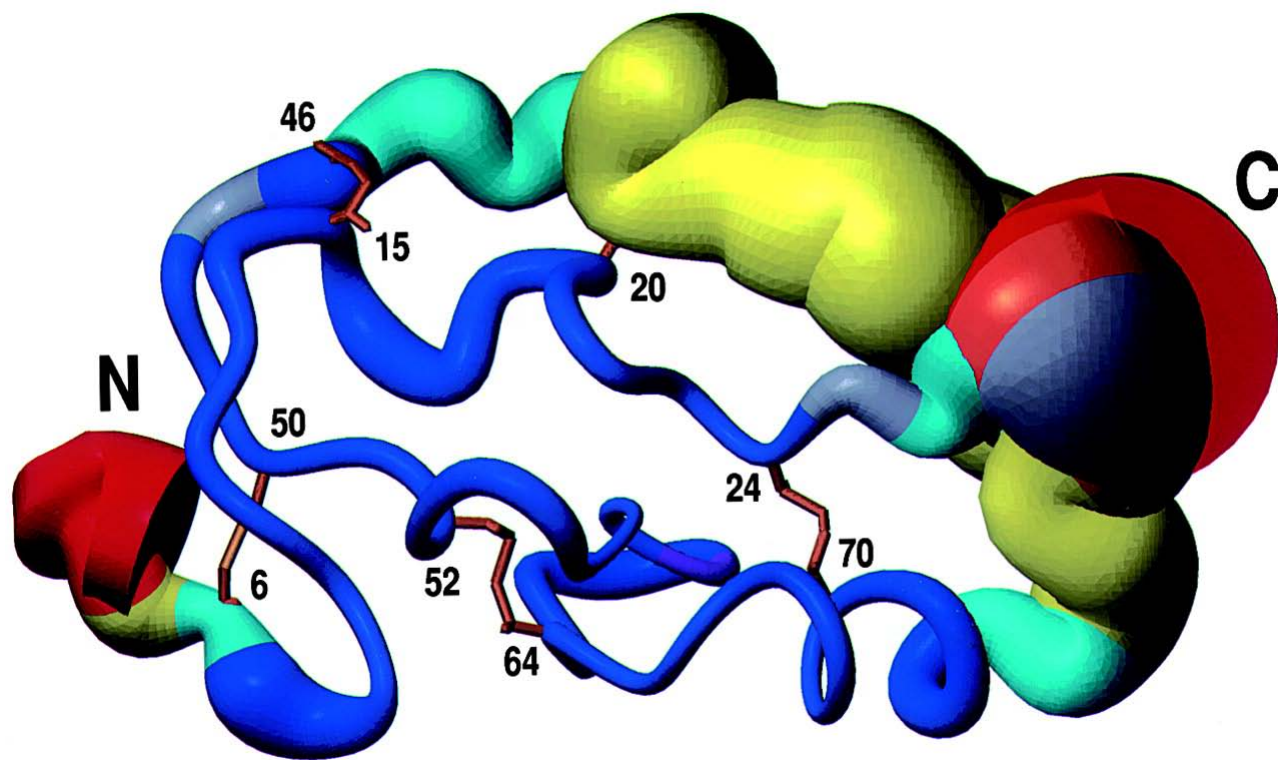


Nekateri proteini so 'namerno' nestabilni (vsaj lokalno); to omogoča:

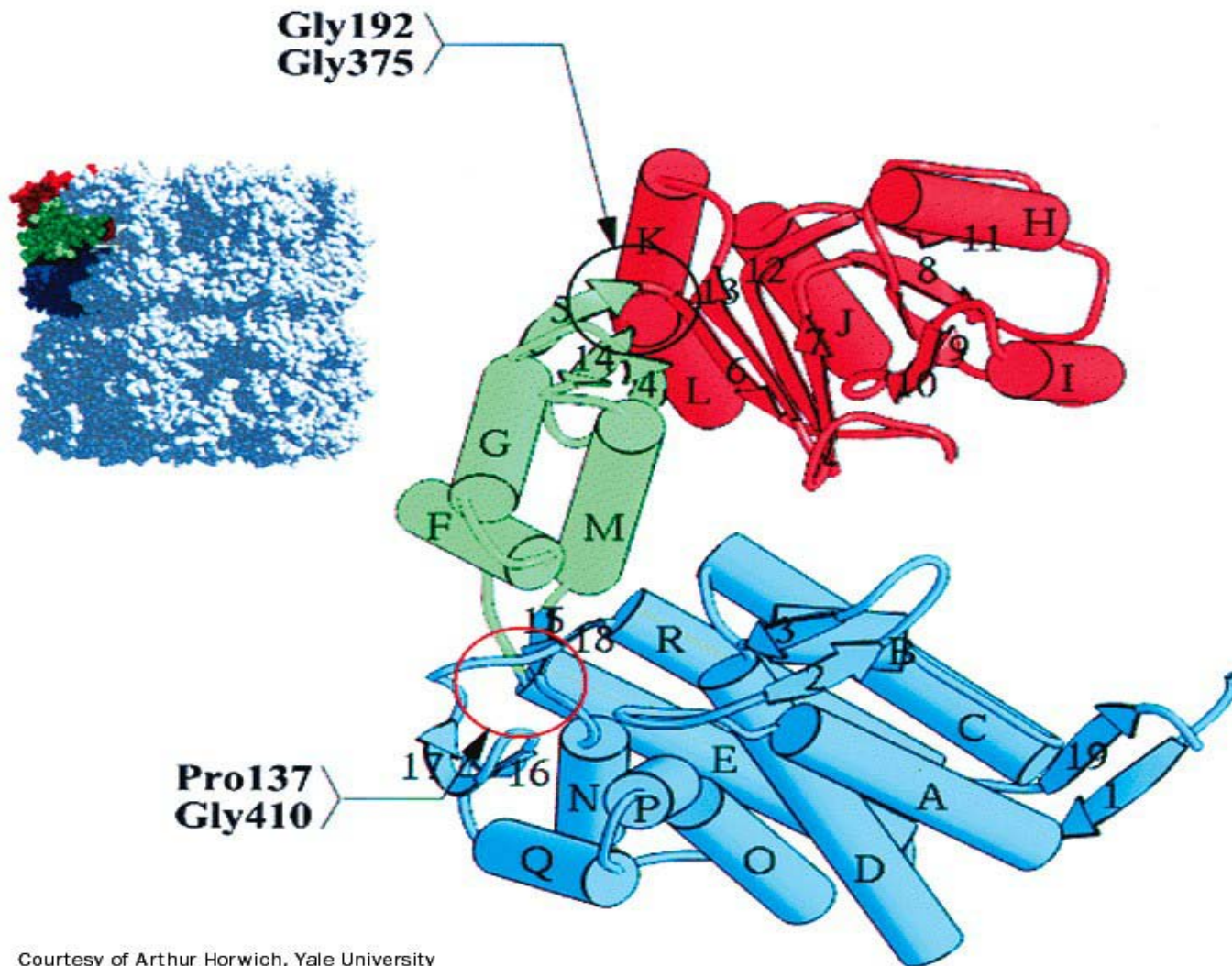
- konformacijske spremembe
- prilagoditve pri vezavi na druge molekule (interakcije protein - protein in protein - nukleinska kislina)



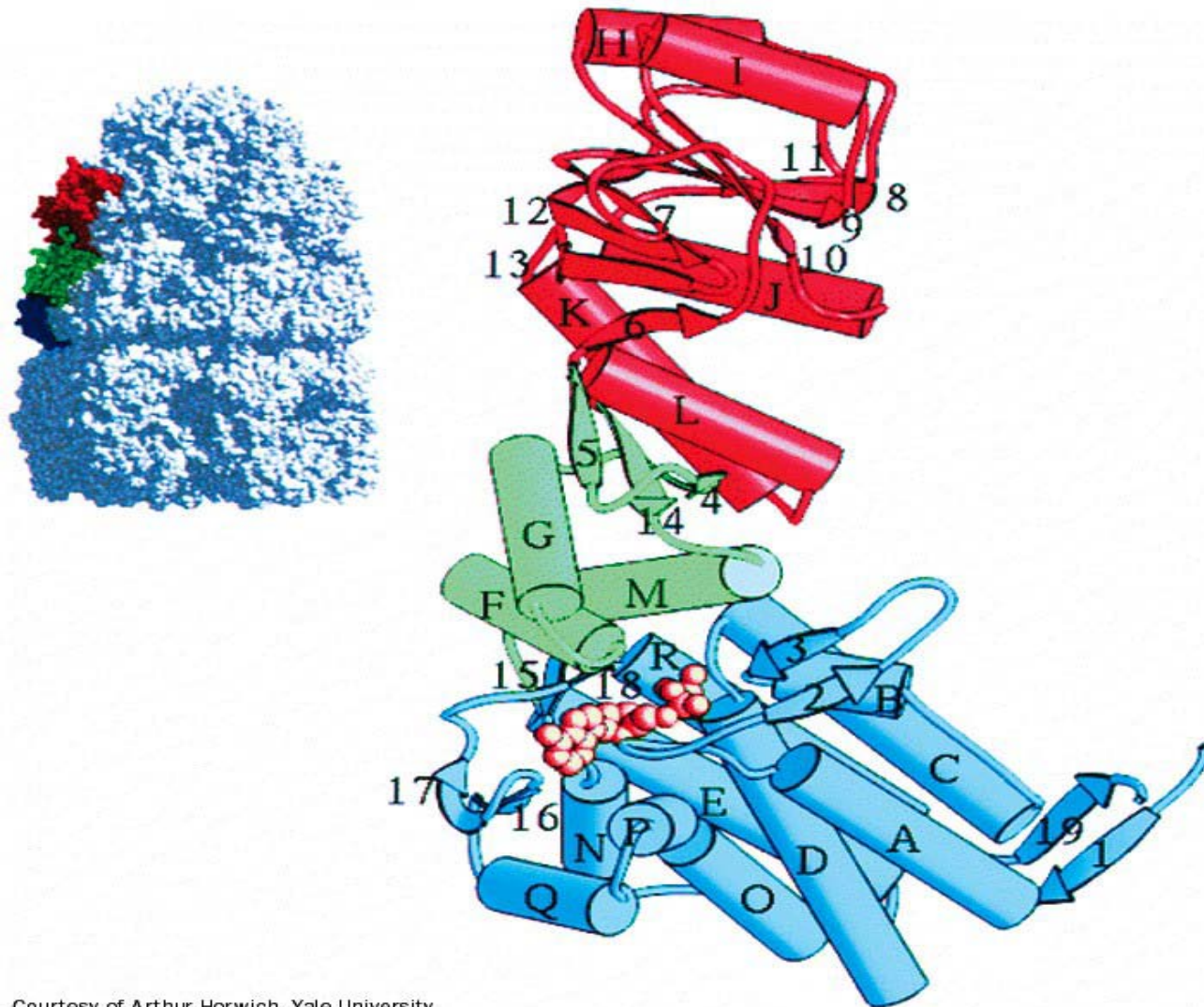
transkripcijski faktor (P. Wright, Scripps)



Premikanje domen v šaperonu GroEL (1).



Premikanje domen v šaperonu GroEL (2).



Metabolično 'obračanje' proteinov:

Proteini 'živijo' v celici različno dolgo.

N.pr.: razpolovni čas encimov v jetrih je od 0.2 do 150 ur.

Pravilo N-konca: Razpolovni čas proteinov je povezan s strukturo N-terminalne AK:

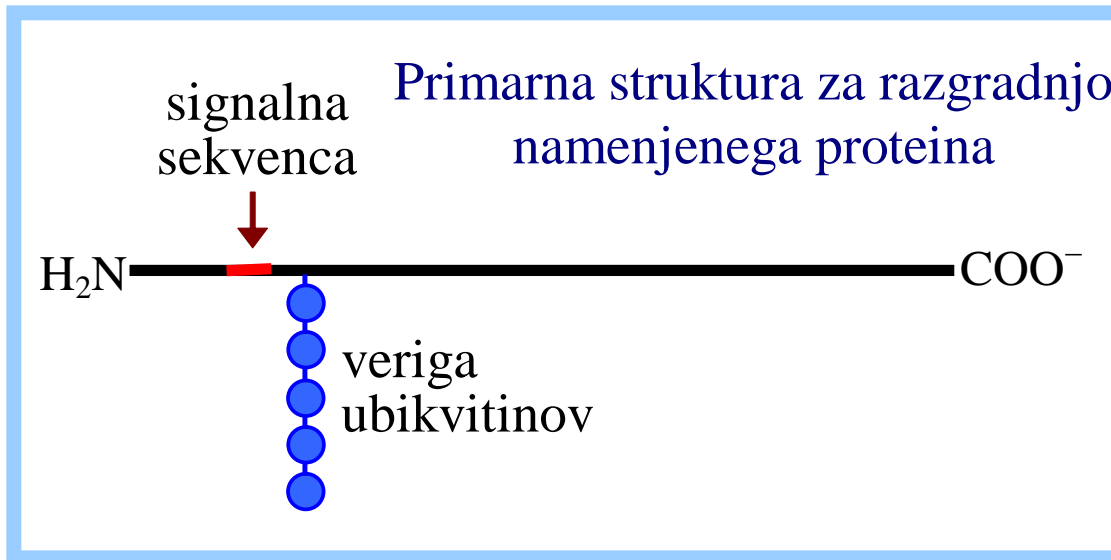
- ◆ Za proteine z N-terminalnimi Met, Ser, Ala, Thr, Val ali Gly je razpolovni čas večji od 20 ur.
- ◆ Za proteine z N-terminalnimi Phe, Leu, Asp, Lys ali Arg je razpolovni čas 3 min ali manj.

Pravilo PEST: proteini, ki imajo veliko Pro (P), Glu (E), Ser (S) ali Thr (T) se hitreje razgradijo kot drugi proteini.

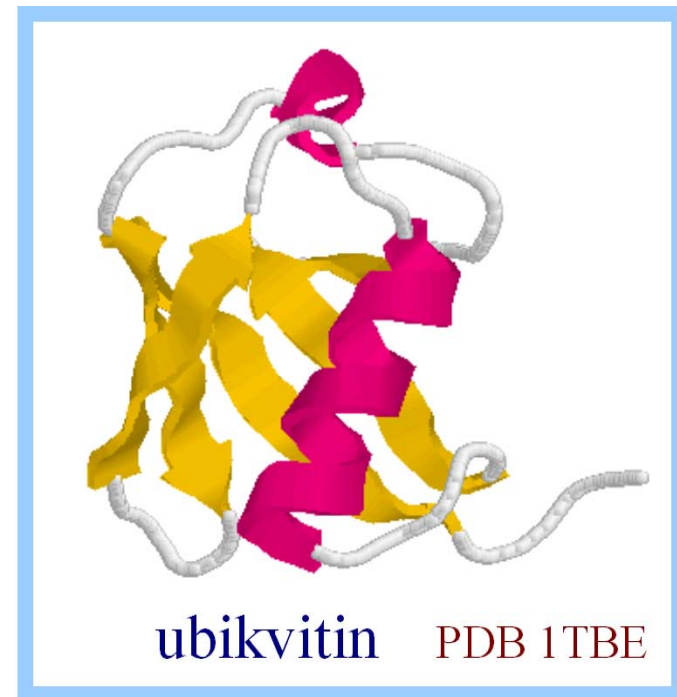
Selektivno razgradnjo proteinov uravnavajo znotrajcelični in zunajcelični signali.

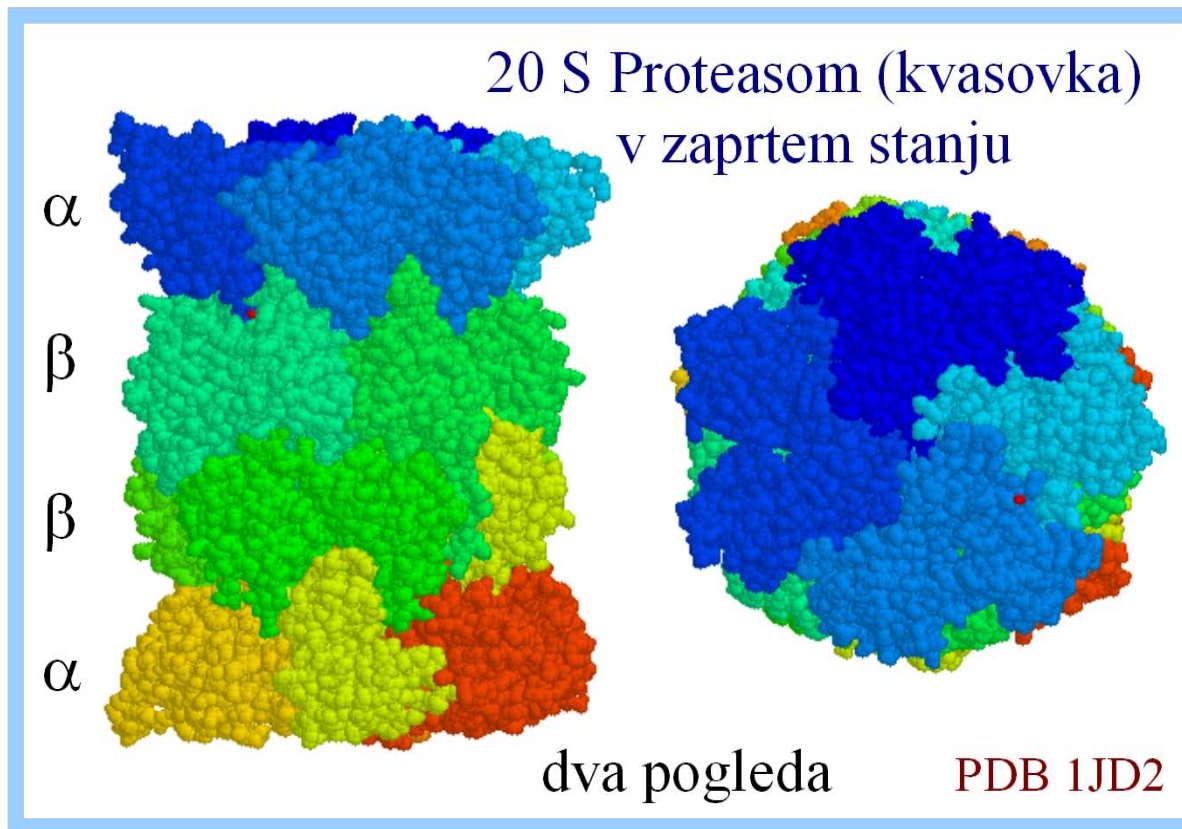
Eden od njih je vezava **ubikvitina** (majhen protein) na za razgranje namenjen (predvsem delno razvit) protein.

Razgradnja ubikvitiniranih proteinov poteka v **proteasomih**.



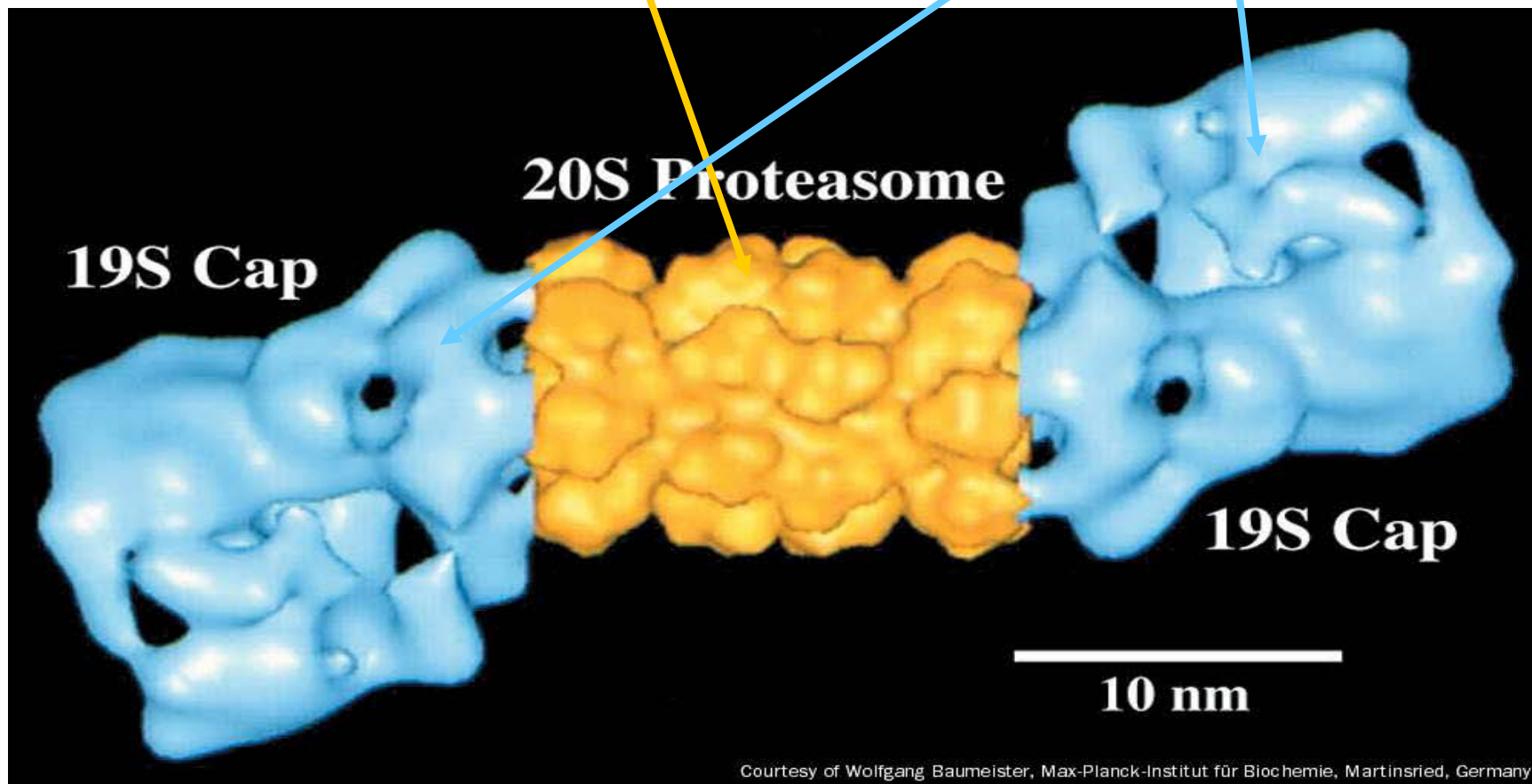
Veriga **4 ali več** ubikvitinov usmeri protein v razgradnjo.





Proteasome je velik kompleks. Jedro ima 4 obroče (2 α in 2 β), vsak je iz 7 proteinov, znotraj pa je votlina kjer se razgrajujejo ubikvitirani proteini. Po 3 podenote v β obročih so proteolitični encimi.

Elektronskomikroskopska slika 26S proteasoma *Drosophile melanogaster*. Poleg jedra (20S) sta prikazani dve 'kapi' (19S), ki usmerjata ubikvitinirane proteine v jedro, kjer se razgradijo.

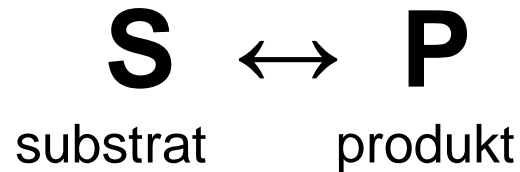


Kontrola kvalitete proteinske sinteze

- Proteasomi prepoznajo in razgrajujejo poli-ubikvitinirane proteine, ki so celici lastni.
- Nekateri celici lastni proteini se le mono-ubikvitinirajo; ti verjetno doživijo razgradnjo v lizosomih.
- Lizosomi so organeli, ki vsebujejo proteolitične encime (katepsine); namenjeni so razgradnji predvsem izvenceličnih proteinov.
- Oba sistema sta potrebna za odstranjevanje 'izrabljenih' in napačno zvitih ali razvitih proteinov.
- Okrog 50 % vseh sintetiziranih proteinov se takoj razgradi!

Kako encimi delujejo?

nekatalizirana reakcija



encimsko katalizirana reakcija



E – encim

S – substrat, **P** – produkt

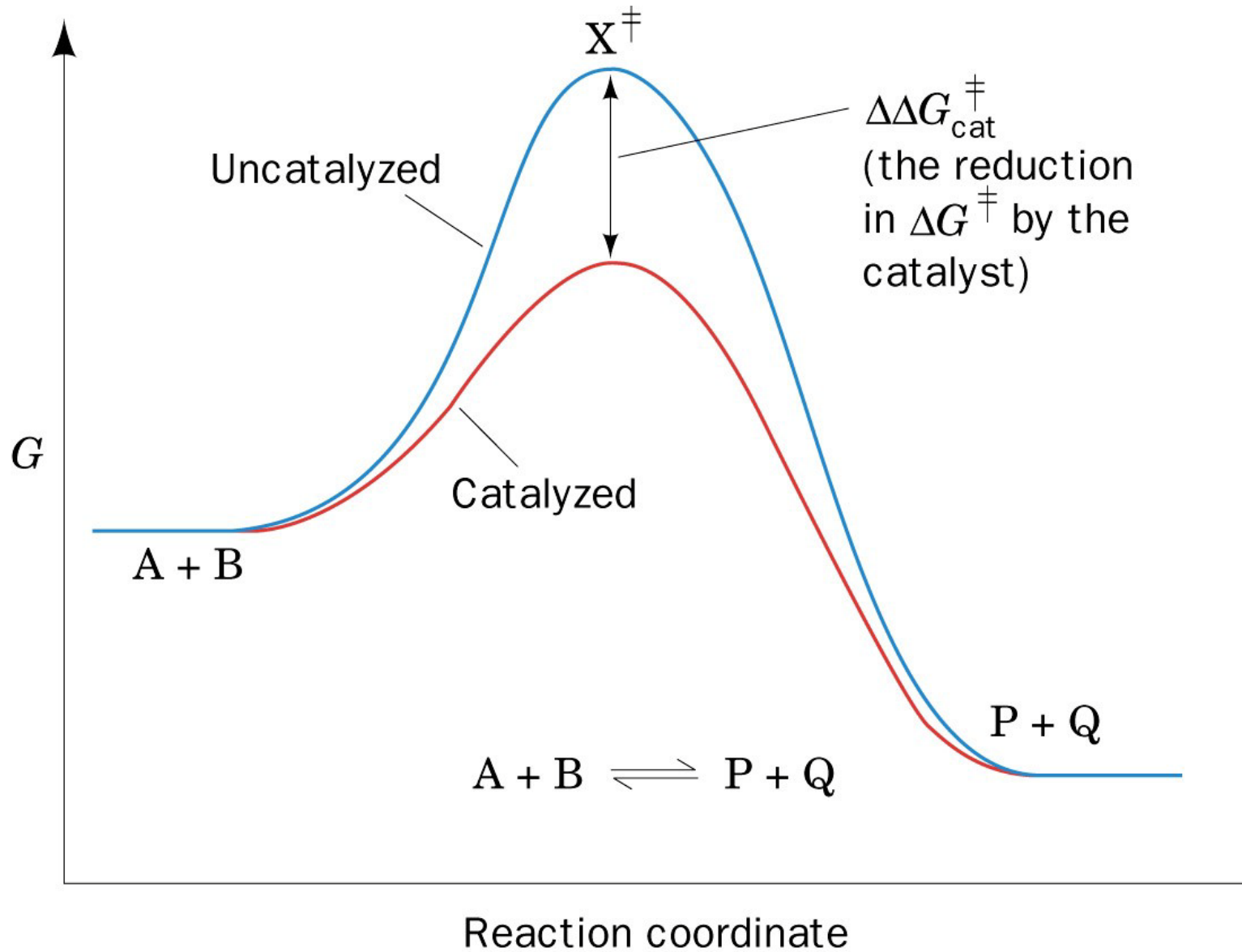
ES – kompleks encim-substrat

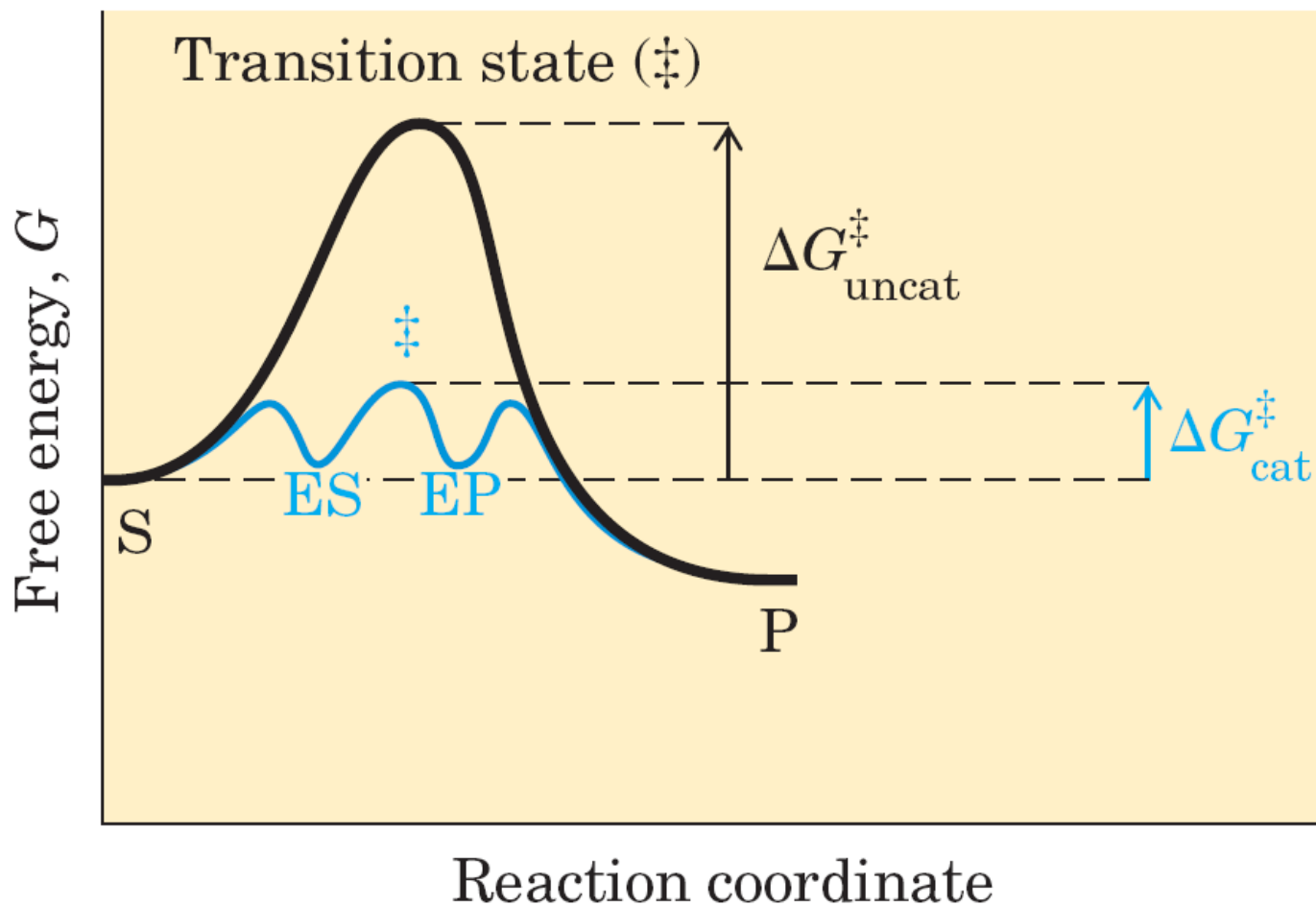
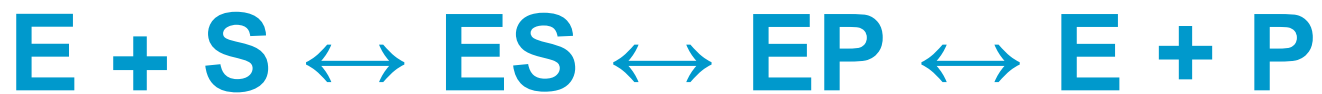
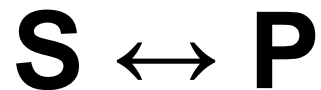
EP – kompleks encim-produkt

}

POZOR, nista aktivirana kompleksa!

Encimi so bio-katalizatorji, ki zmanjšajo aktivacijsko energijo reakcije



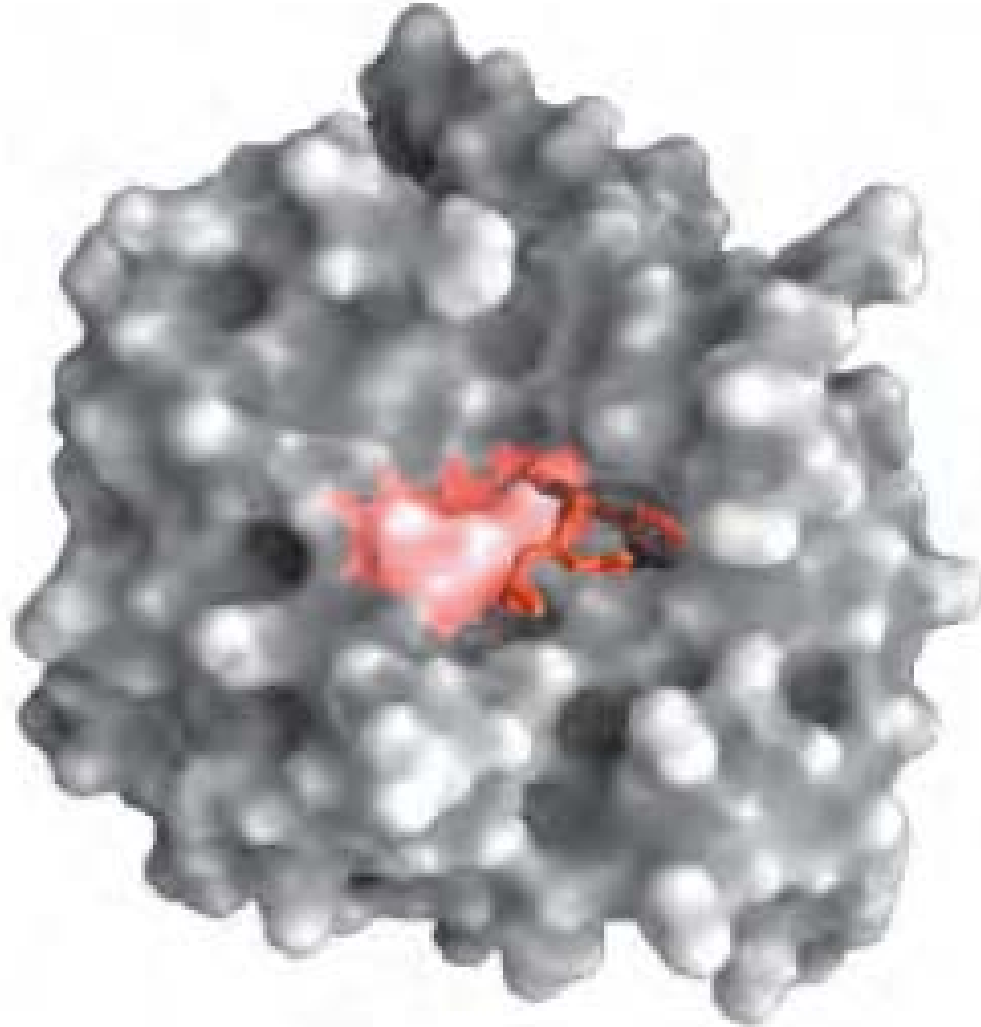


$$\Delta G^{\circ} = \Delta G^{\circ} = -RT \ln K$$

$$\Delta G^{\#} < \Delta G^{\#}$$

$$v > v$$

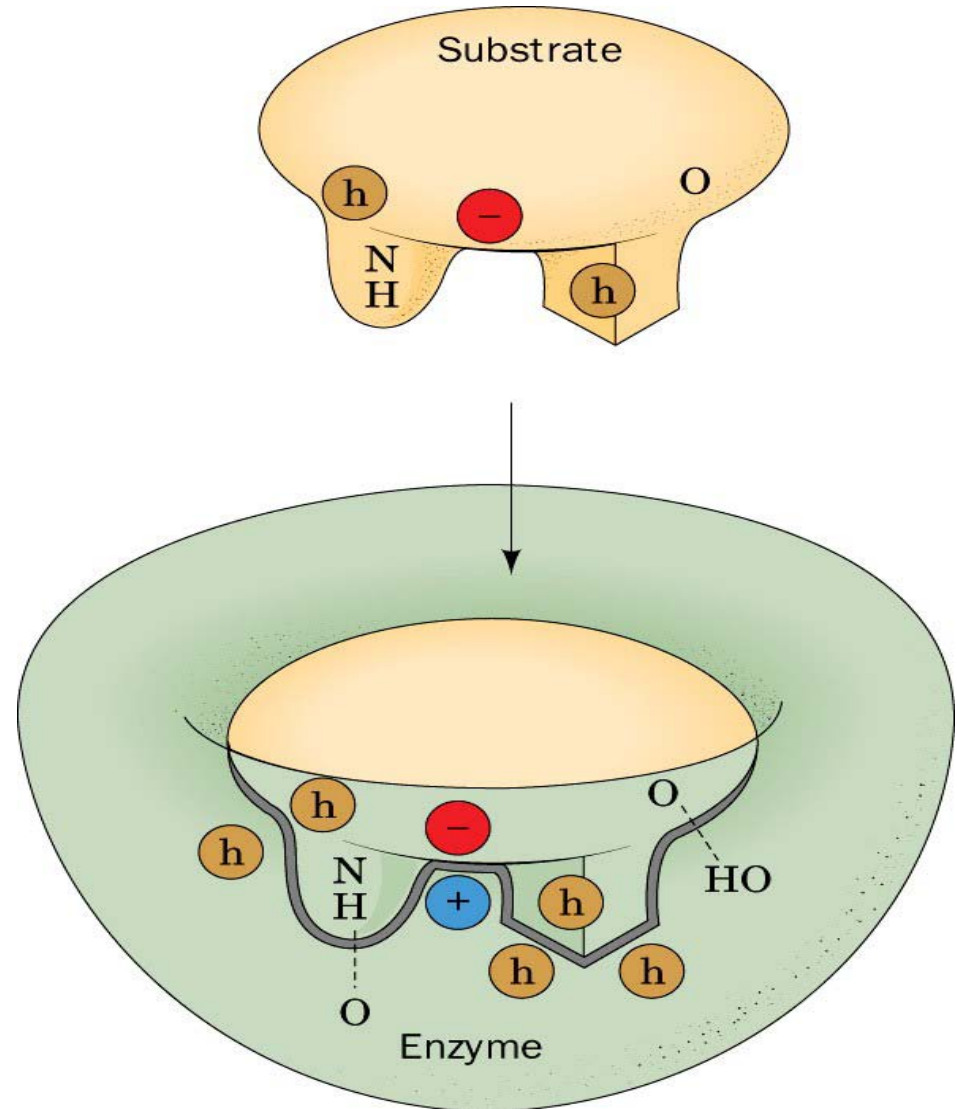
Substrat se veže na aktivno mesto, ki je navadno le manjši del encimske molekule



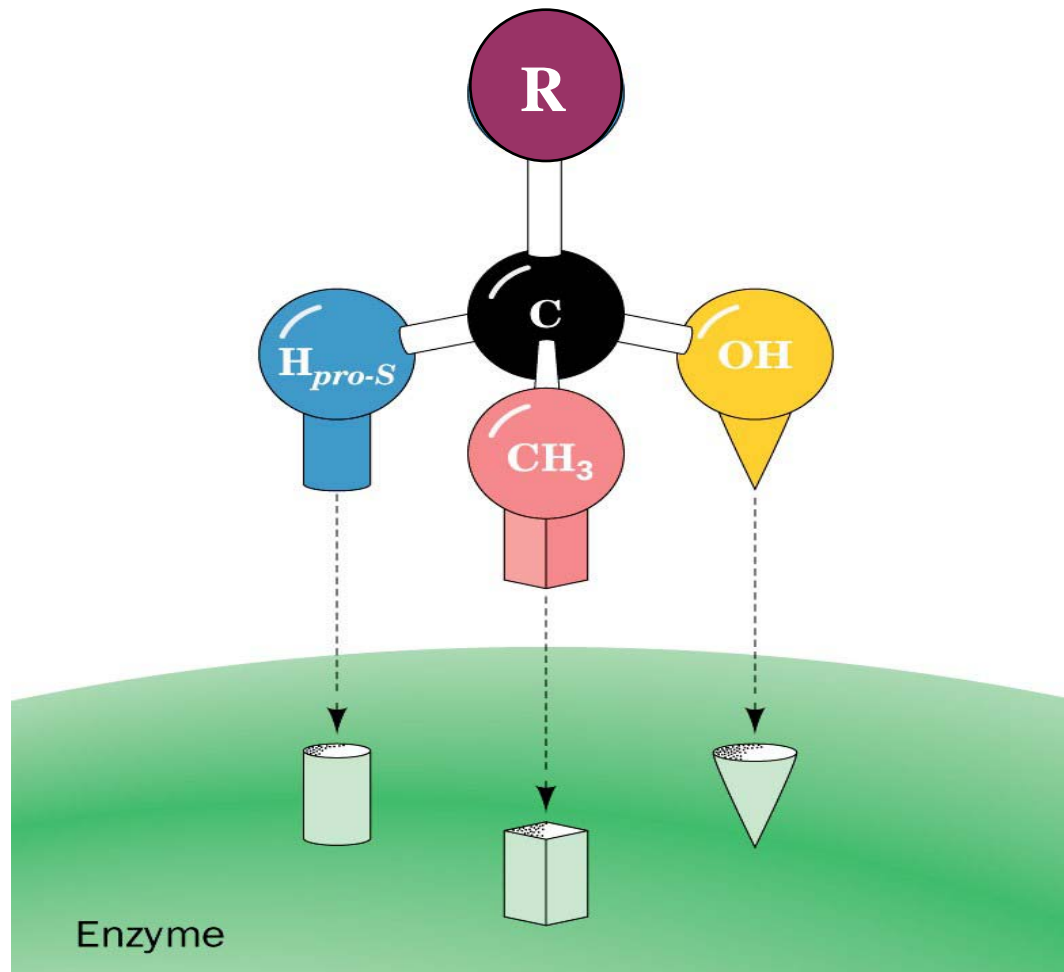
Encim predstavlja specifično okolje za molekulo substrata.

Koncept aktivnega mesta

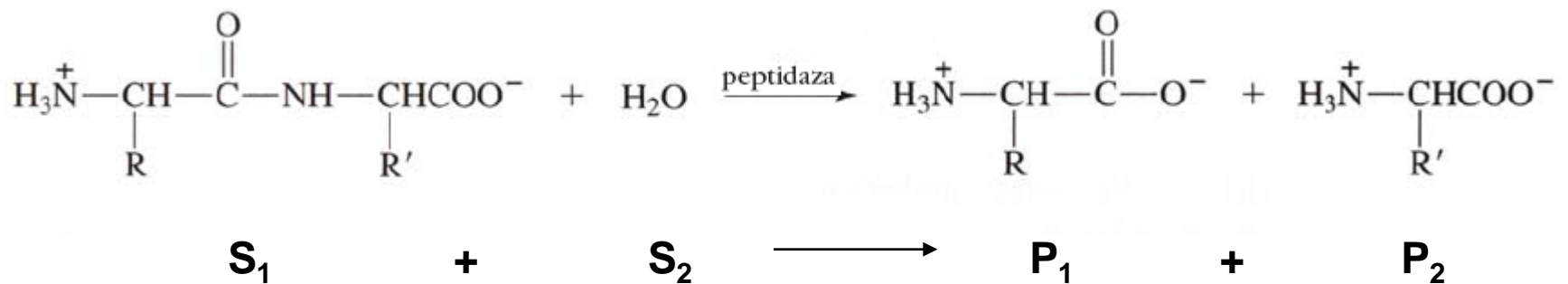
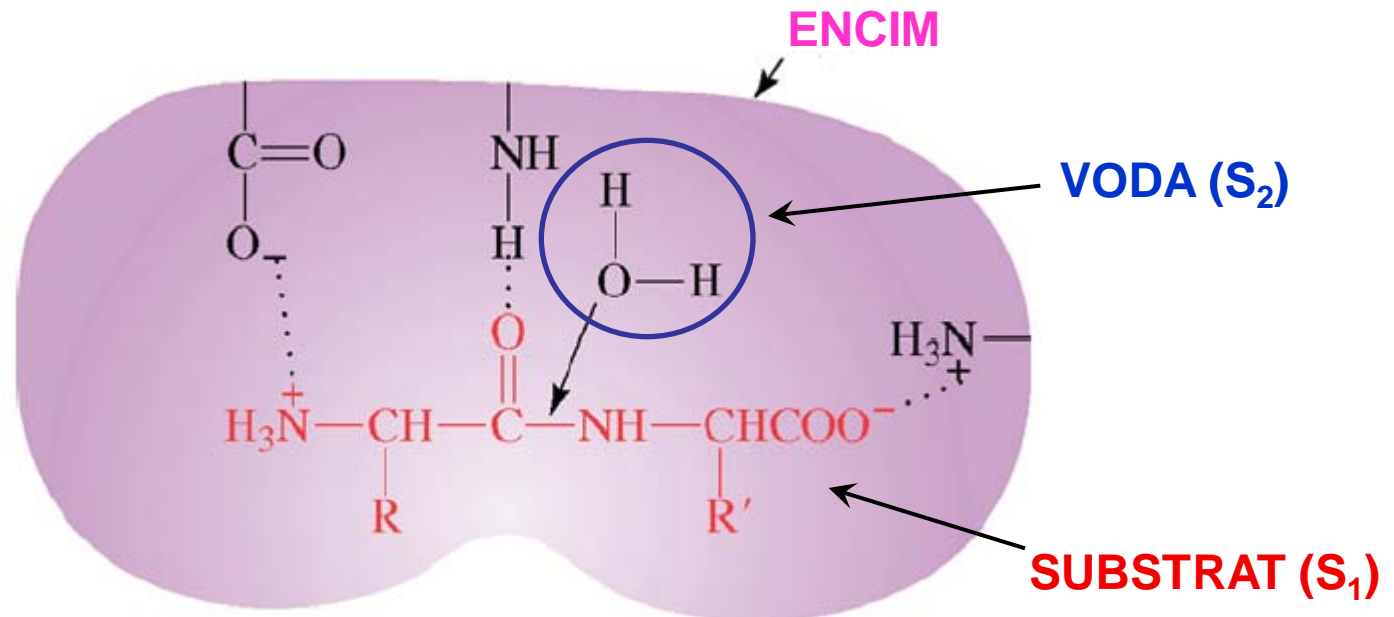
Nastanek kompleksa encim–substrat omogoča geometrijska in fizikalno-kemijska komplementarnost encima in substrata. Skupine na substratu in v aktivnem mestu encima si ustrezajo tako, da se lahko medsebojno povežejo.



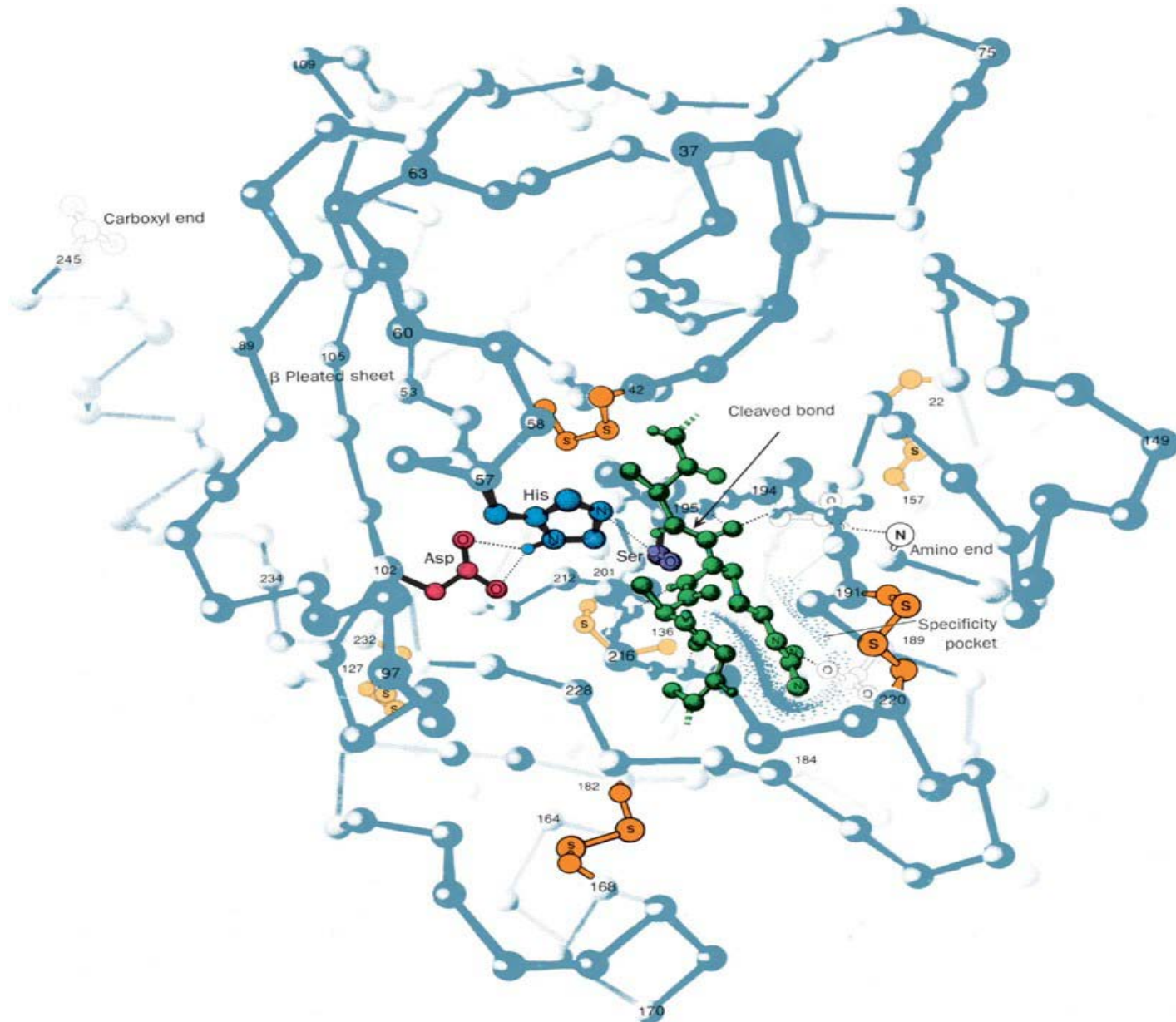
Za razlikovanje enantiomerov mora imeti encim vsaj tri specifična mesta za vezavo substrata.



Peptidaza veže substrat (dipeptid) s tremi šibkimi vezmi, vsili substratu novo konformacijo, kar destabilizira peptidno vez (prehodno stanje!)

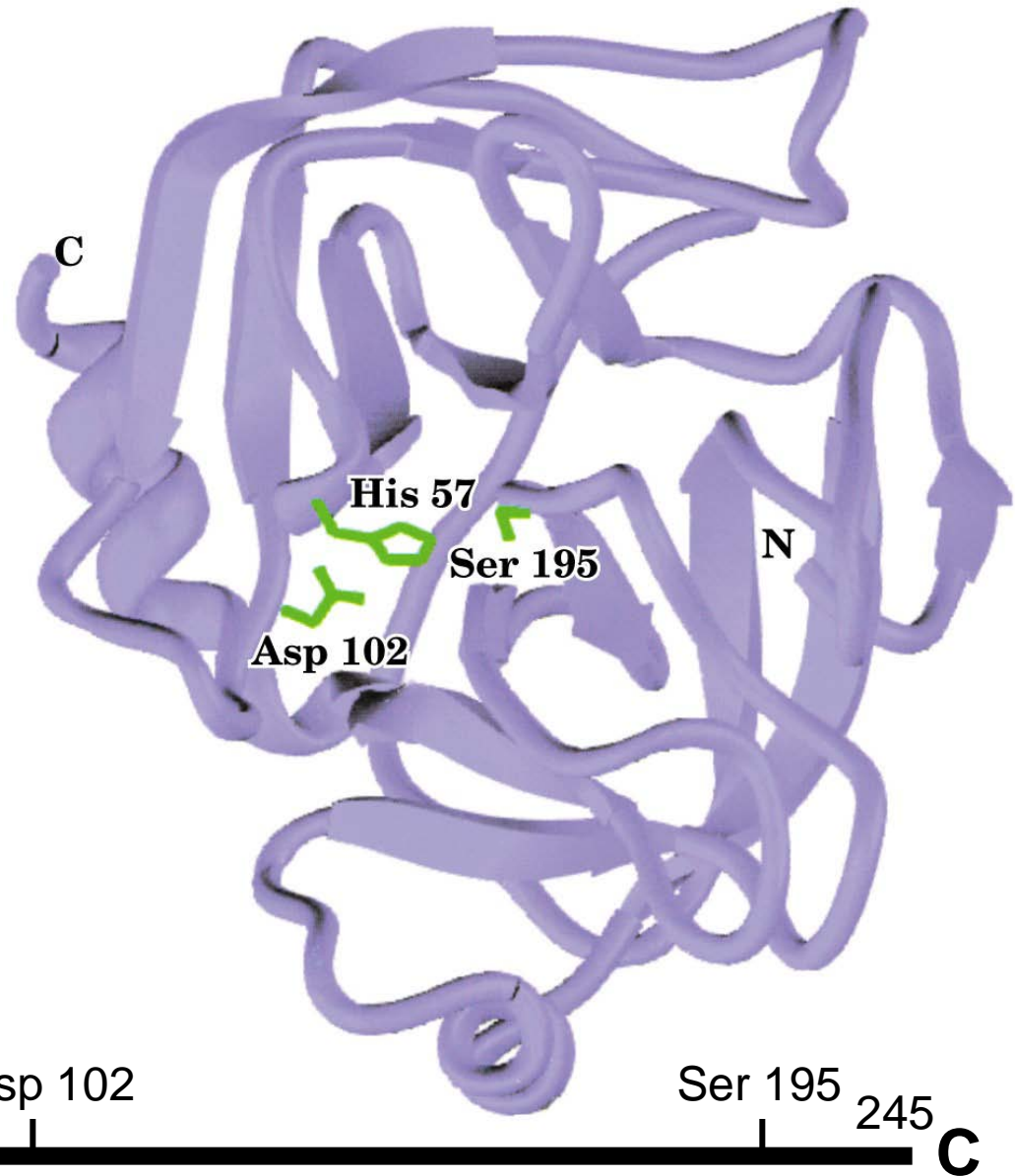


Struktura tripsina s substratom



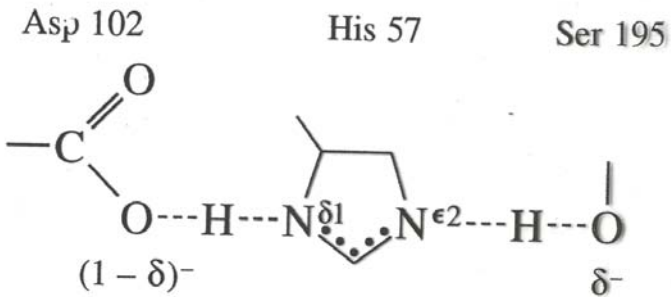
Struktura tripsina

V aktivnem mestu pridejo zaradi III. strukture skupaj AK, ki so navadno v I.strukturi med seboj precej oddaljene.



Katalitična triada

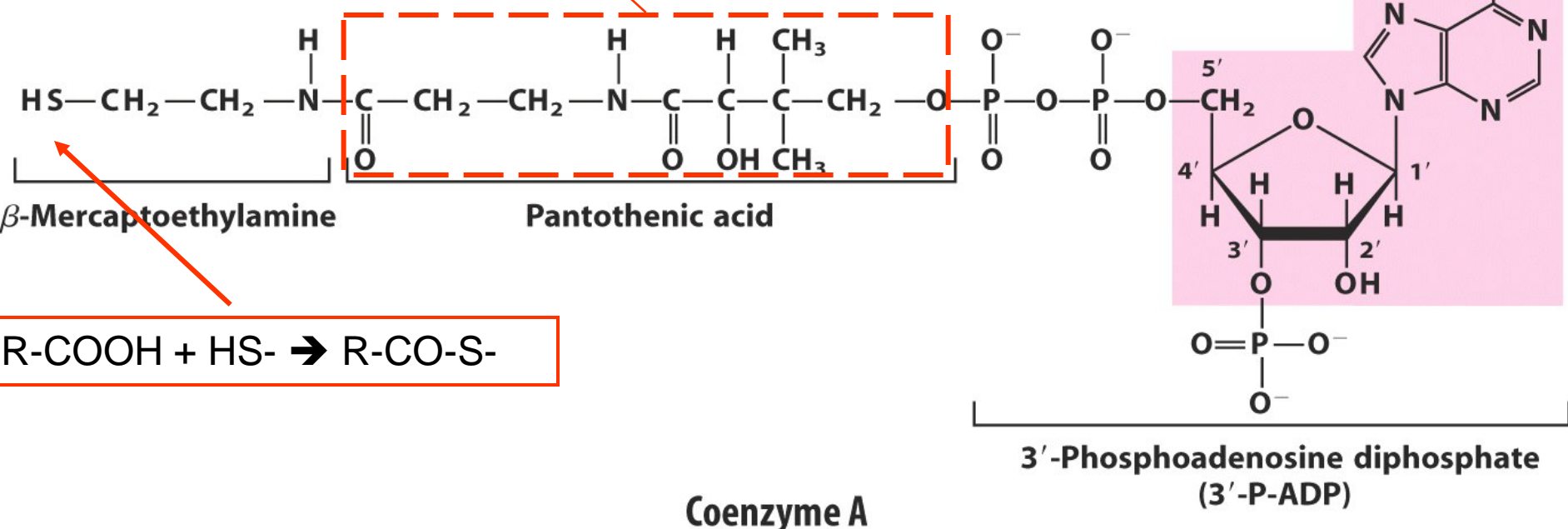
Namen: 'aktivacija' Ser (OH)



KOENCIMI IN PROSTETIČNE SKUPINE (nastanejo iz vitaminov), ZAGOTOVIJO DODATNE FUNKCIONALNE SKUPINE

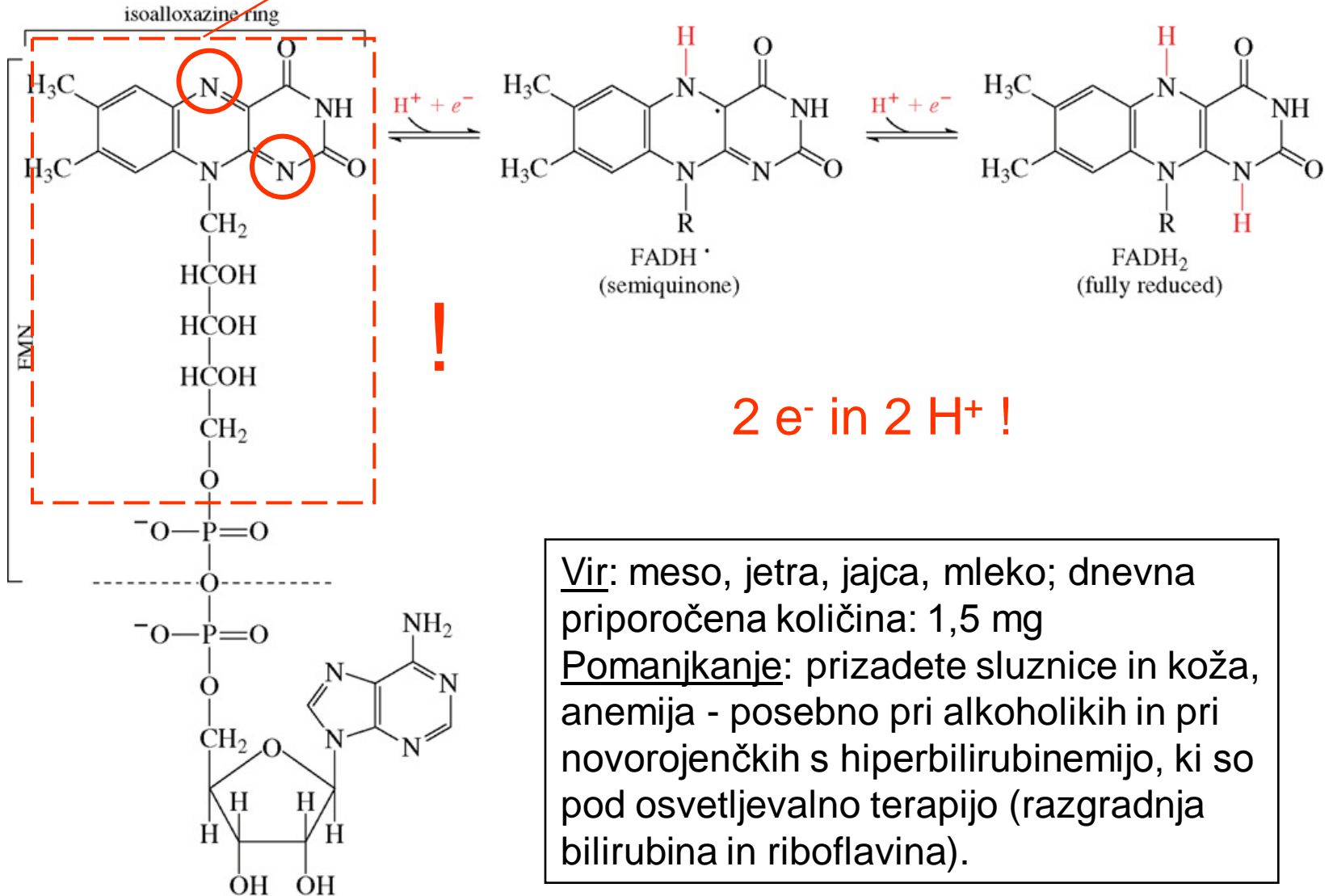
PREKURZOR (VITAMIN)	SKUPINA PRENOSA	KOENCIM
tiamin (vitamin B1)	aldehidna skupina	tiamin pirofosfat
riboflavin (vitamin B2)	elektroni in protoni	FMN, FAD
nikotinska kislina (niacin)	hidridni ion (H⁻)	NAD
pantotenska kislina in druge molekule	acilne skupine	koencim A
piridoksin (vitamin B6)	amino skupine	piridoksal fosfat
vitamin B12	H-atomi in alkilne skupine	(ciano)kobalamin
biotin	CO₂ (-COOH)	biocitin
folna kislina	skupine z enim C-atomom	tetrahidrofolna kislina
lipojska kislina	elektroni in acilne skupine	lipojska kislina

PREKURZOR (VITAMIN)	SKUPINA PRENOSA	KOENCIM
<u>pantotenska kislina in druge molekule</u>	acilne skupine	koencim A



Vir: vsaka hrana; dnevna priporočena količina: 4 - 7 mg.
Pomanjkanja niso nikoli opazili, s težavo se ga da umetno izzvati v strogo nadzorovanih eksperimentalnih razmerah.

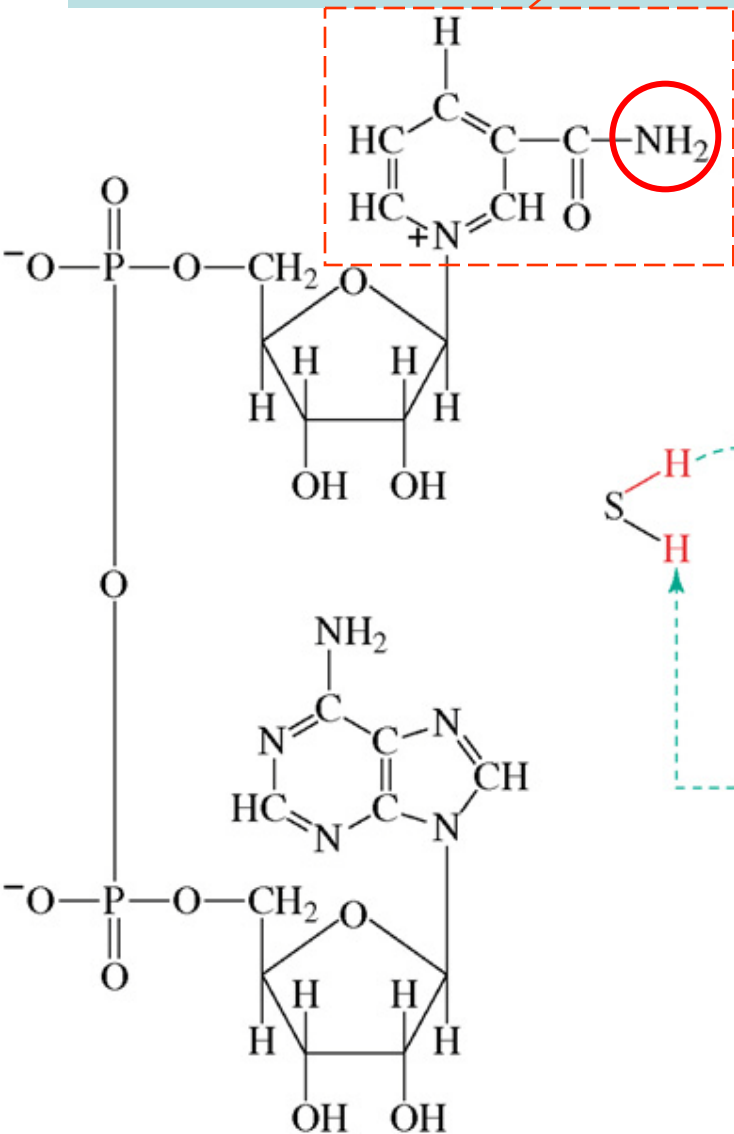
PREKURZOR (VITAMIN)	SKUPINA PRENOSA	KOENCIM
<u>riboflavin (vitamin B2)</u>	elektroni in protoni	FMN, FAD



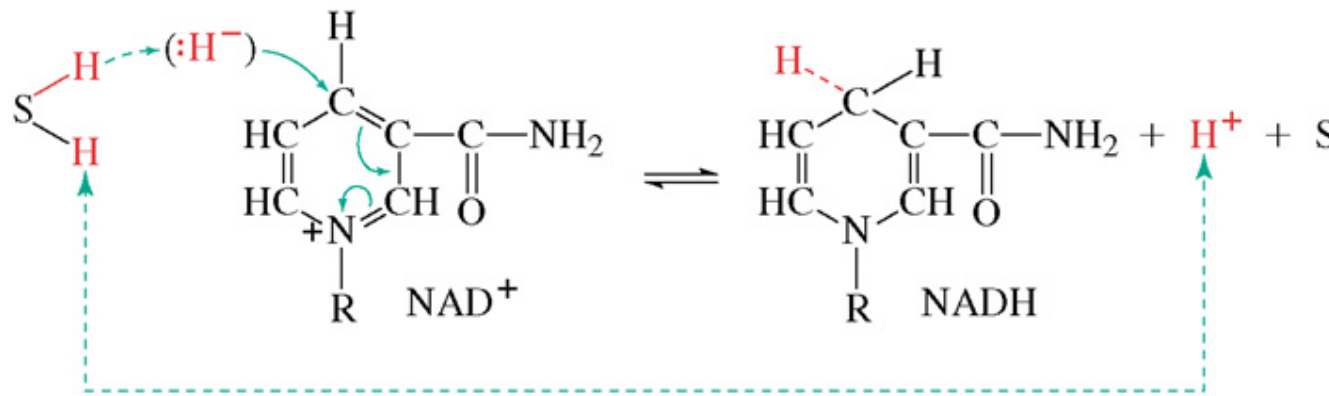
Vir: meso, jetra, jajca, mleko; dnevna priporočena količina: 1,5 mg
Pomanjkanje: prizadete sluznice in koža, anemija - posebno pri alkoholikih in pri novorojenčkih s hiperbilirubinemijo, ki so pod osvetljevalno terapijo (razgradnja bilirubina in riboflavina).

Flavin adenine dinucleotide (FAD)

PREKURZOR (VITAMIN)	SKUPINA PRENOSA	KOENCIM
<u>nikotinska kislina (niacin)</u>	hidridni ion (H^-)	NAD



2 e⁻ in 1 H⁺ = H⁻ = hidridni ion!



Vir: meso, jetra, kvas, semena; dnevna priporočena količina: 15 - 20 mg (tudi neuč. sinteza iz Trp).
Pomanjkanje (koruza!): **pelagra** - prizadeta je koža, v resnejši obliki pa dermatitis, diareja in demenca (lahko tudi ireverzibilna). V razvitem svetu redka.

Nicotinamide adenine dinucleotide (NAD⁺)